

10. CARS

Photo: Henrik Bladh

- Introduction
- Theory
- Rotational CARS/vibrational CARS
- Temp . & conc. measurements
- Applications

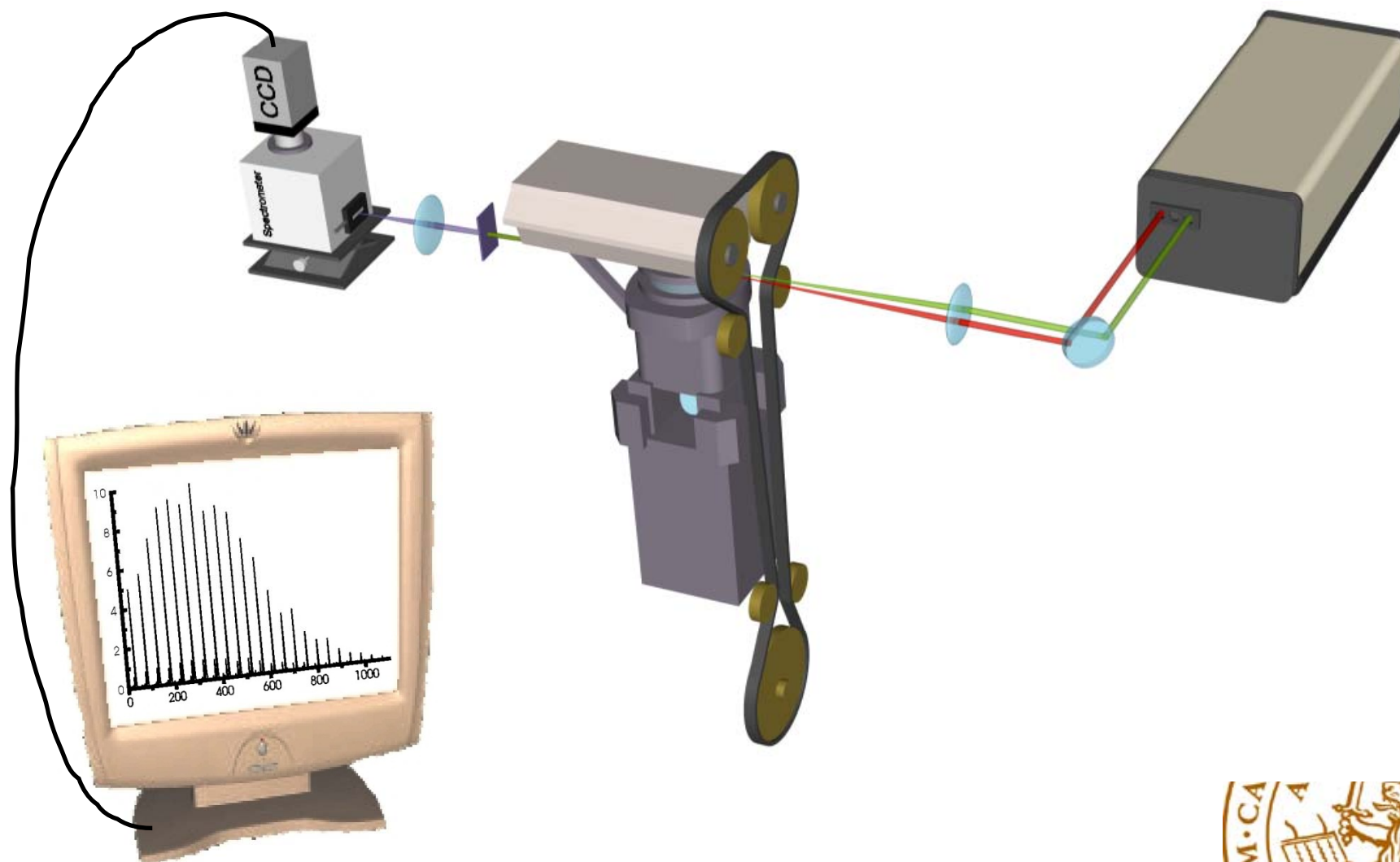


Classification of techniques

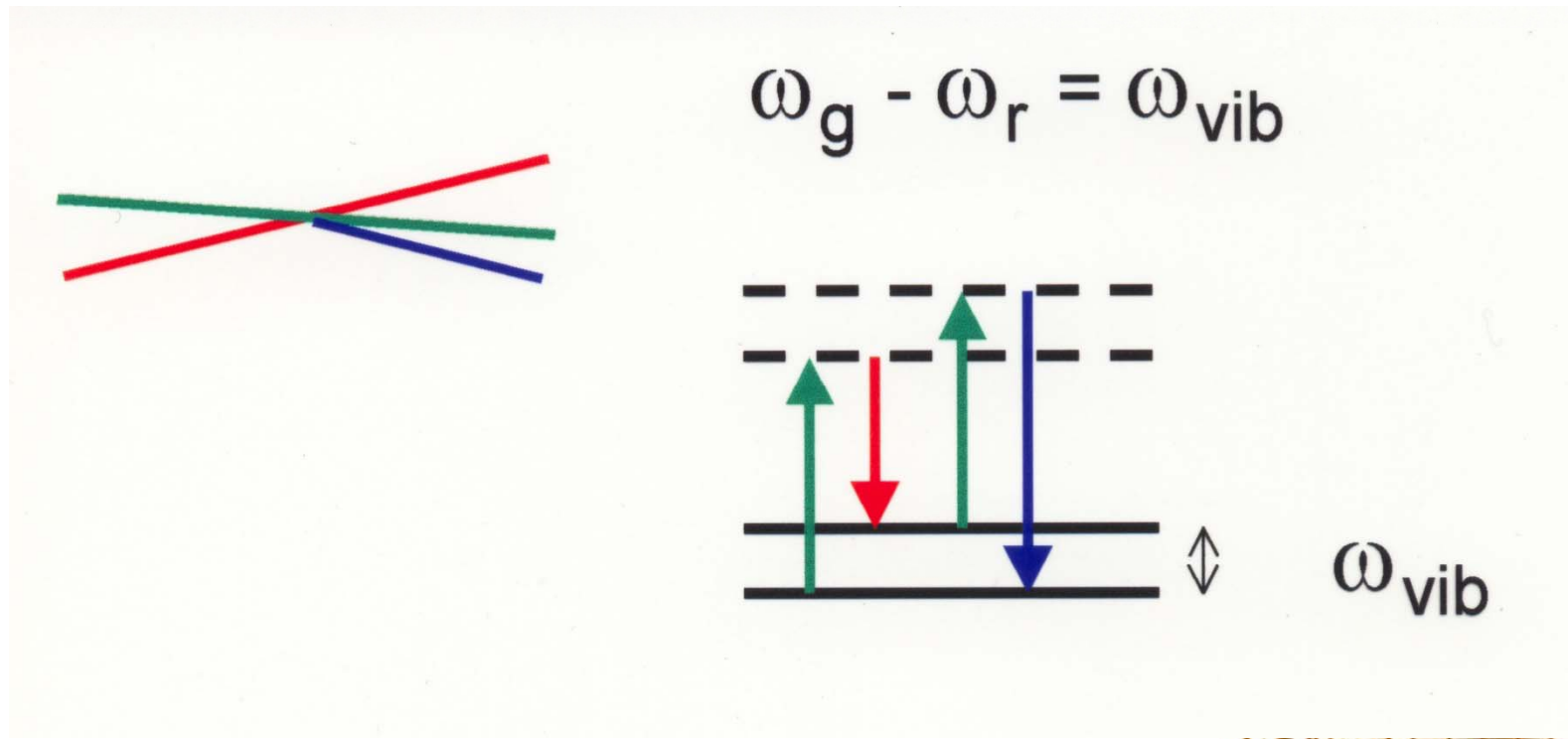
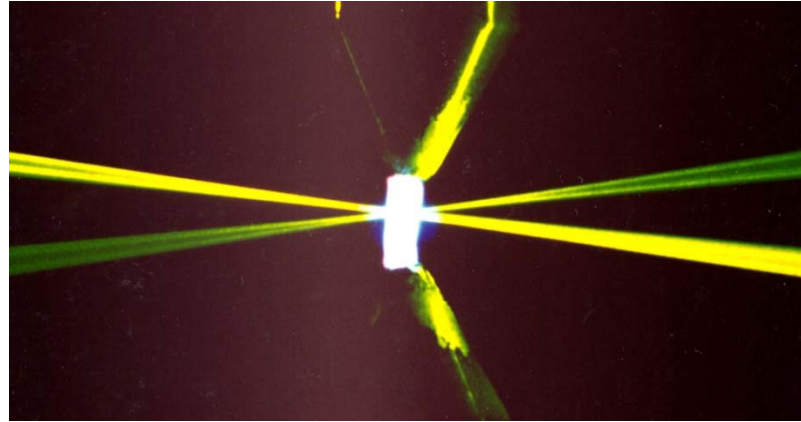
Approach Level	Incoherent	Coherent
Major species	Raman Scattering	CARS
Minor constituents	LIF	PS/DFWM



CARS



CARS conceptual behaviour



Background - CARS

Coherent anti-Stokes Raman scattering/spectroscopy (CARS) is a technique that is used for

- temperature measurements
- concentration measurements

Species with high concentrations ($> \sim 1\%$) are probed N_2 , O_2 , CO_2 , H_2O , CO , H_2

The first demonstration of CARS in flames was made in 1973 by Taran et al. and in an IC engine by Stenhouse et al. (1979)



Characteristics of CARS

- Signal generated as a new laser beam
- Pointwise thermometry with high accuracy
- Signal blue shifted relative to the primary lasers
- Complex theory
- Relatively complicated experiments
- Operator skill is needed
- Two approaches: - Vibrational CARS
- Rotational CARS



Nonlinear optics

Thus far in the course, the induced polarization of molecules has almost always (except for frequency doubling/mixing) been assumed to depend linearly on the applied electromagnetic field. This is, however, only valid for incident radiation of low intensity.

Generally, the induced polarization is a nonlinear function of the applied electromagnetic field:

$$\vec{P} = \vec{P}^{(1)} + \vec{P}^{(2)} + \vec{P}^{(3)} + \dots$$

$$\vec{P} = \epsilon_0 \left(\chi^{(1)} \vec{E} + \chi^{(2)} \vec{E}^2 + \chi^{(3)} \vec{E}^3 + \dots \right)$$

For gases, which are isotropic (inversion symmetry), the even order polarizations vanish

CARS is a four-wave mixing process based on the nonlinear response via the third-order susceptibility ($\chi^{(3)}$)



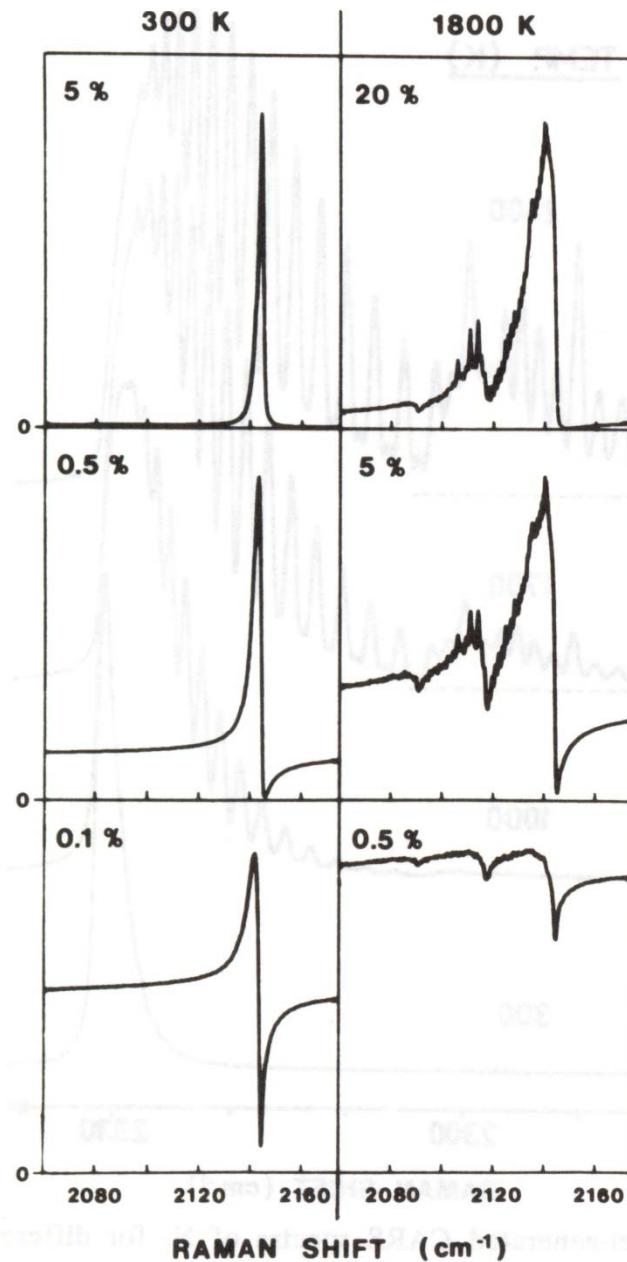
CARS theory

$$S \propto \omega_{CARS}^2 I_1 I_2 I_3 |\chi_{CARS}|^2 l^2 \left(\frac{\sin(\Delta kl/2)}{\Delta kl/2} \right)^2, \quad \chi_{CARS} = \chi_R + \chi_{NR}$$

- ω_{CARS} is the CARS signal frequency ,
- I_i is the irradiance in laser beam i ,
- χ_{CARS} is the CARS susceptibility,
- l is the interaction length,
- the last factor is the phase-matching condition, that takes the value 1 if perfect phase-matching is achieved.

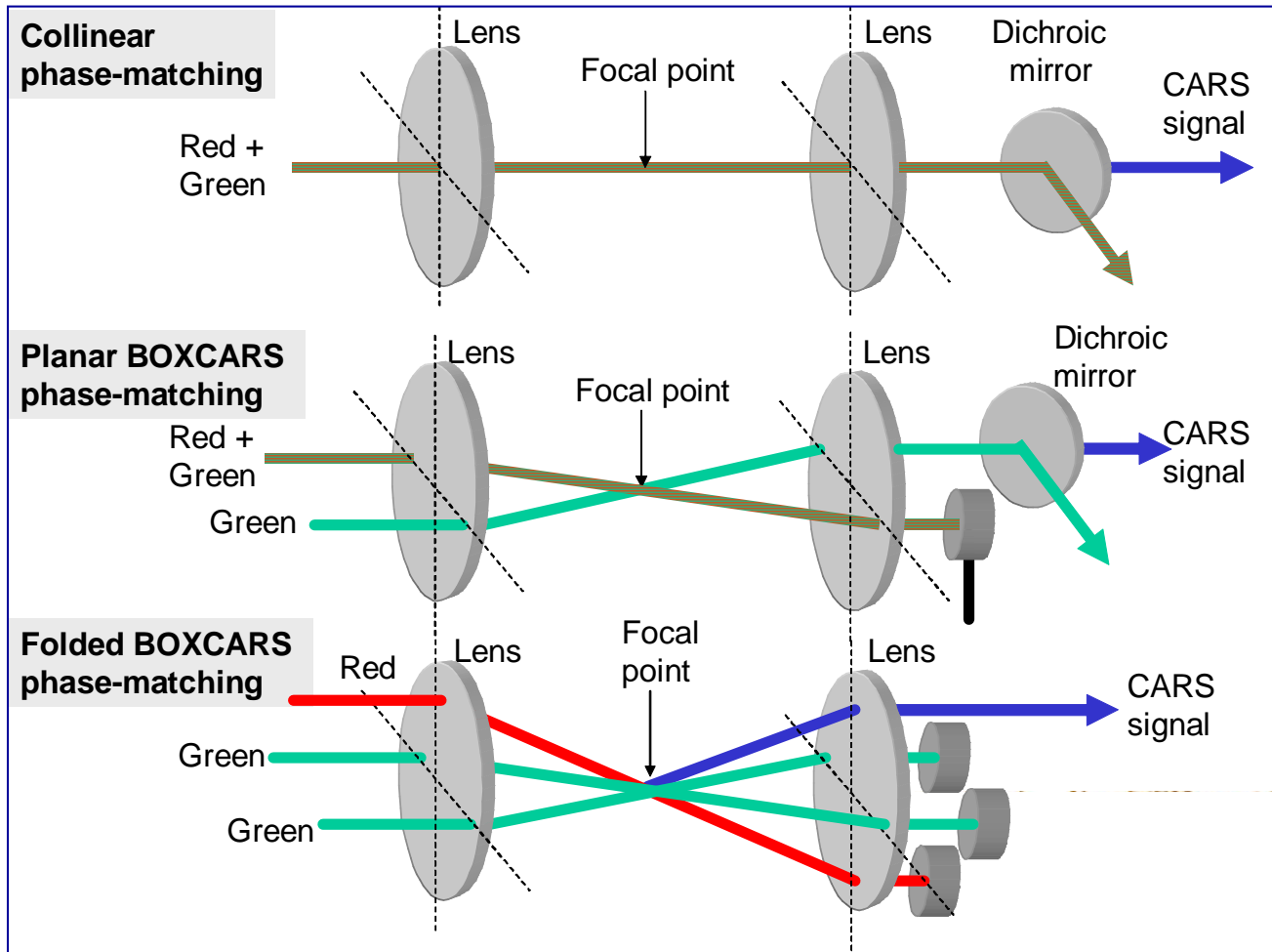
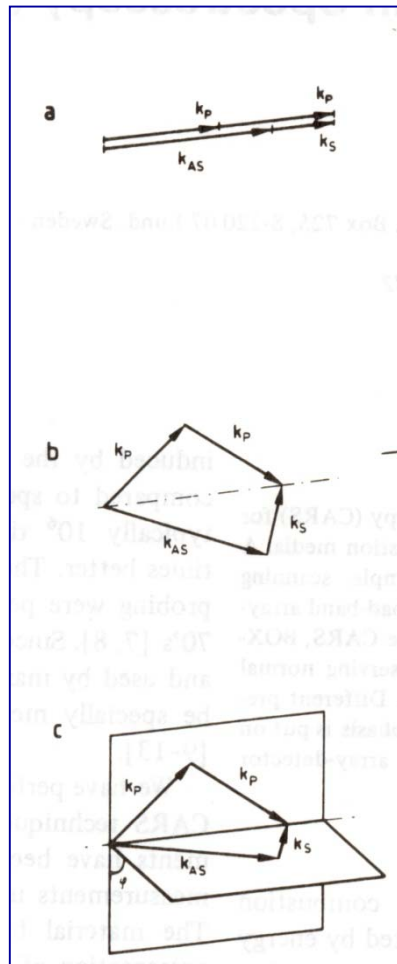


Calculated CARS spectra of CO



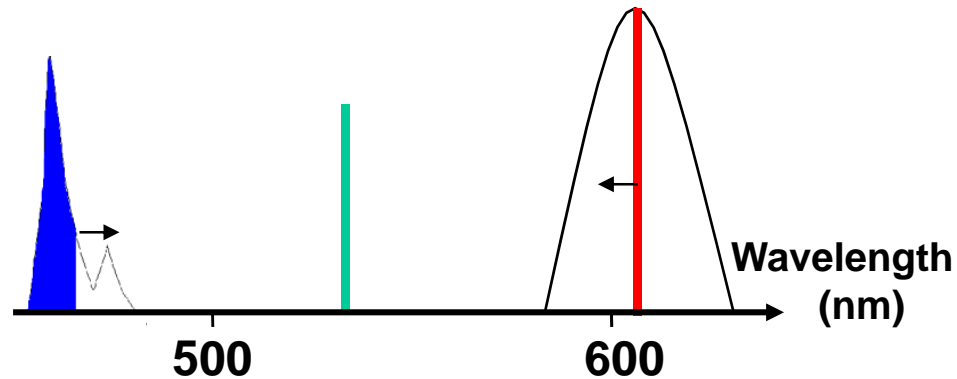
Phase matching implication on the CARS set-up

$$\vec{k}_{CARS} = 2\vec{k}_1 - \vec{k}_2 \quad \left| \vec{k}_i \right| = \frac{2\pi n_i}{\lambda}$$



Generation of CARS spectra

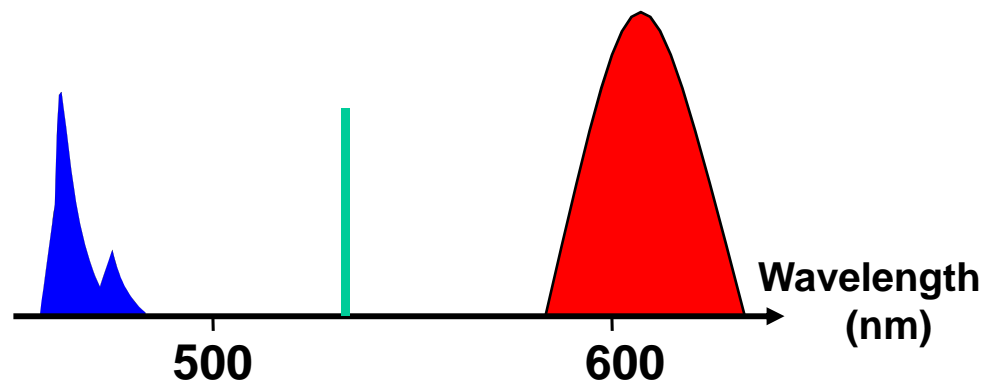
Scanning
CARS



Drawback:

It takes time to scan a spectrum \Rightarrow
can be used in
stationary flames only

Broadband
CARS



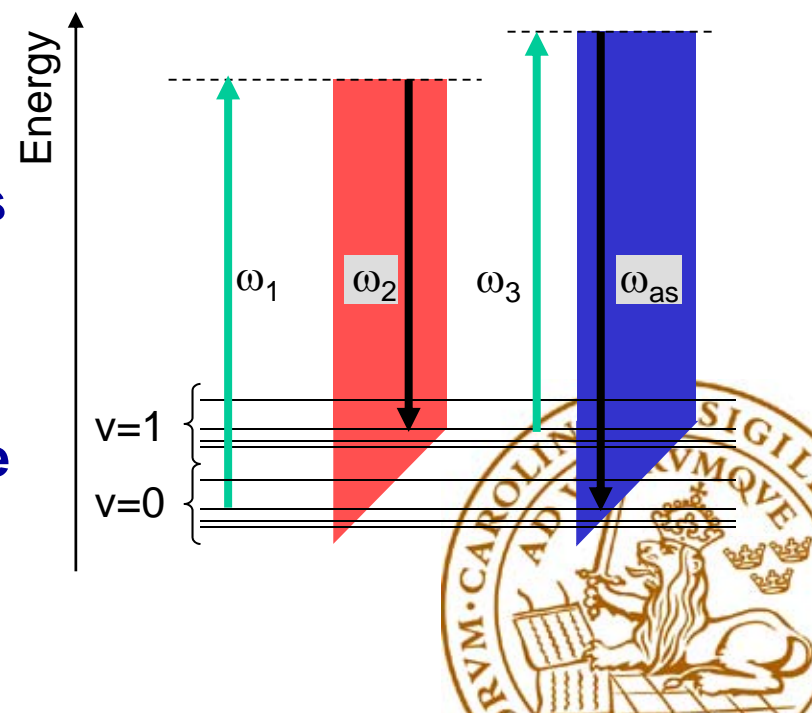
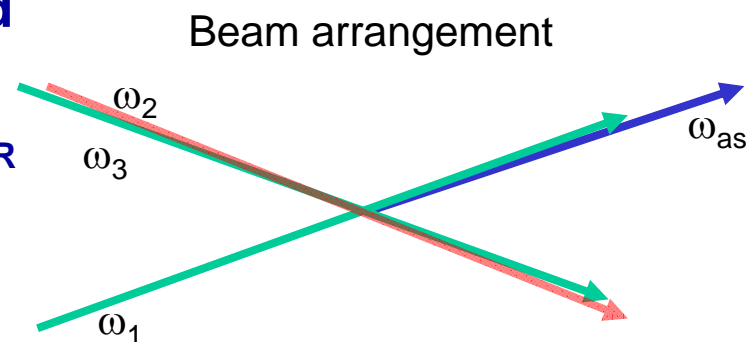
Advantage:

Spectra is obtained
within 10 ns \Rightarrow
can be used in
turbulent flames

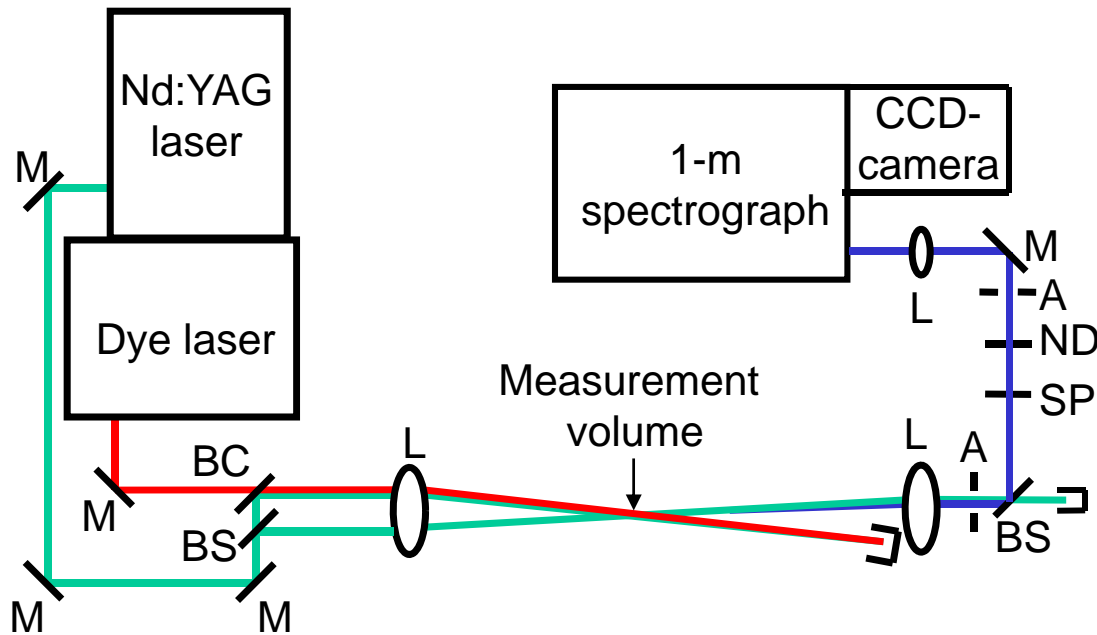


The selection of wavelengths

- The energy difference between ω_1 and ω_2 must match a vibrational Raman resonance in the molecule. $\omega_1 - \omega_2 = \omega_R$
- For nitrogen, with $\omega_R = 2331 \text{ cm}^{-1}$, it means a Nd:YAG laser wavelength of 532 nm and a dye laser spectral profile centred around 607 nm.
- The CARS signal is at frequency: $\omega_{\text{CARS}} = \omega_1 - \omega_2 + \omega_3$, which for nitrogen is at a wavelength of 473 nm. ($\omega_1 = \omega_3$)
- By operating the dye laser in broadband mode all transitions can be monitored simultaneously.



Experimental setup (vibrational CARS)



M = Mirror
BS = Beam splitter
BC = Beam combiner
A = Aperture
ND = Neutral density filter
CCD = Charge-coupled device
L = Lens
SP = Short-pass filter

Laser source

The most common laser system for CARS is a Nd:YAG + dye laser system with repetition rate 10 Hz and pulse duration of ~10 ns.

Optical components

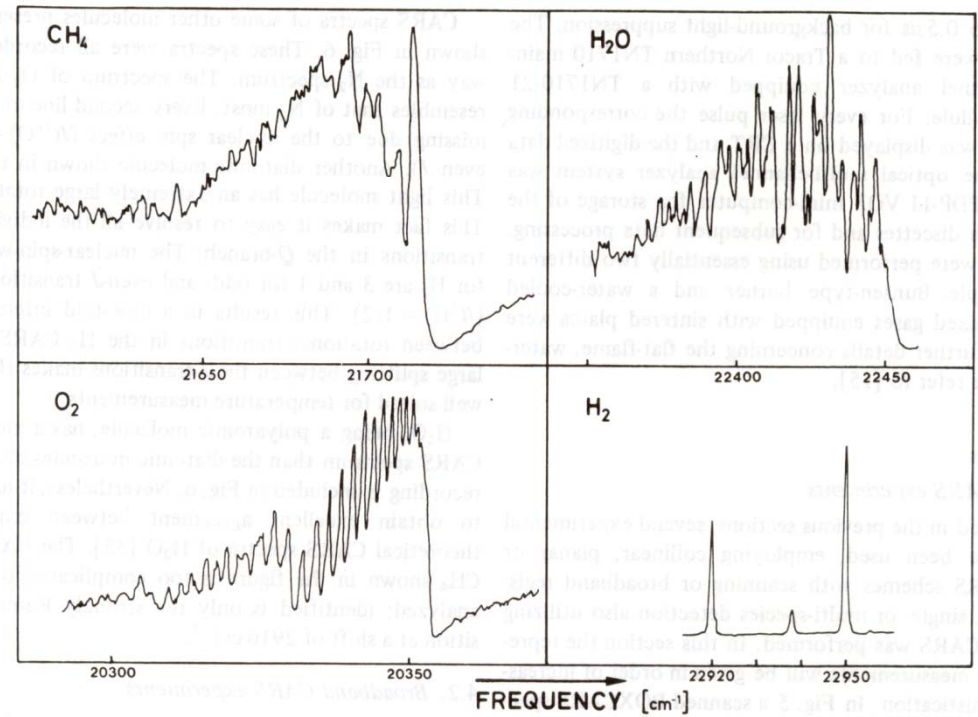
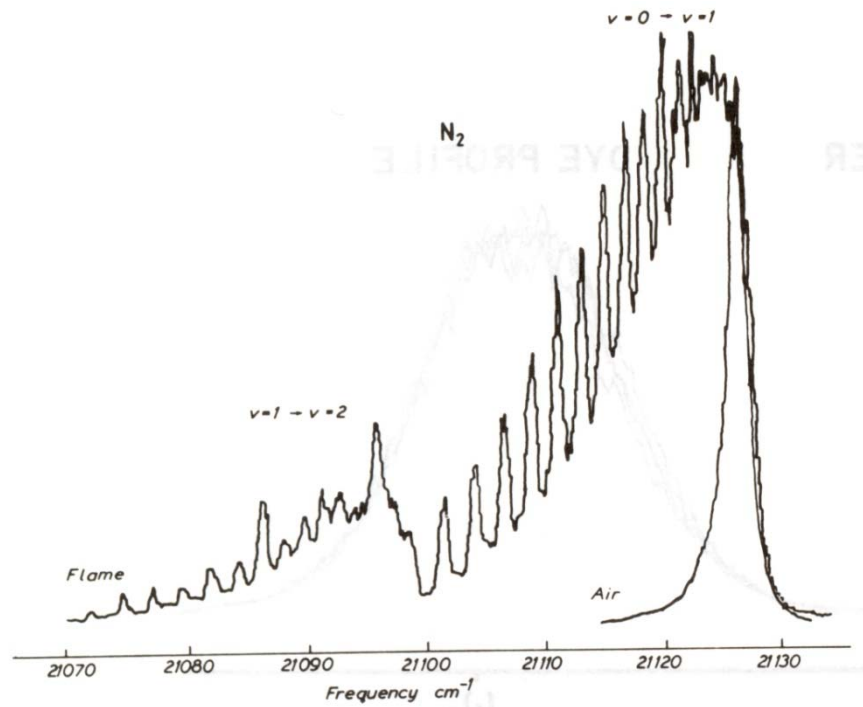
Special optics that can stand high pulse energies are needed.

Detection system

High-resolution spectrometer + CCD-camera

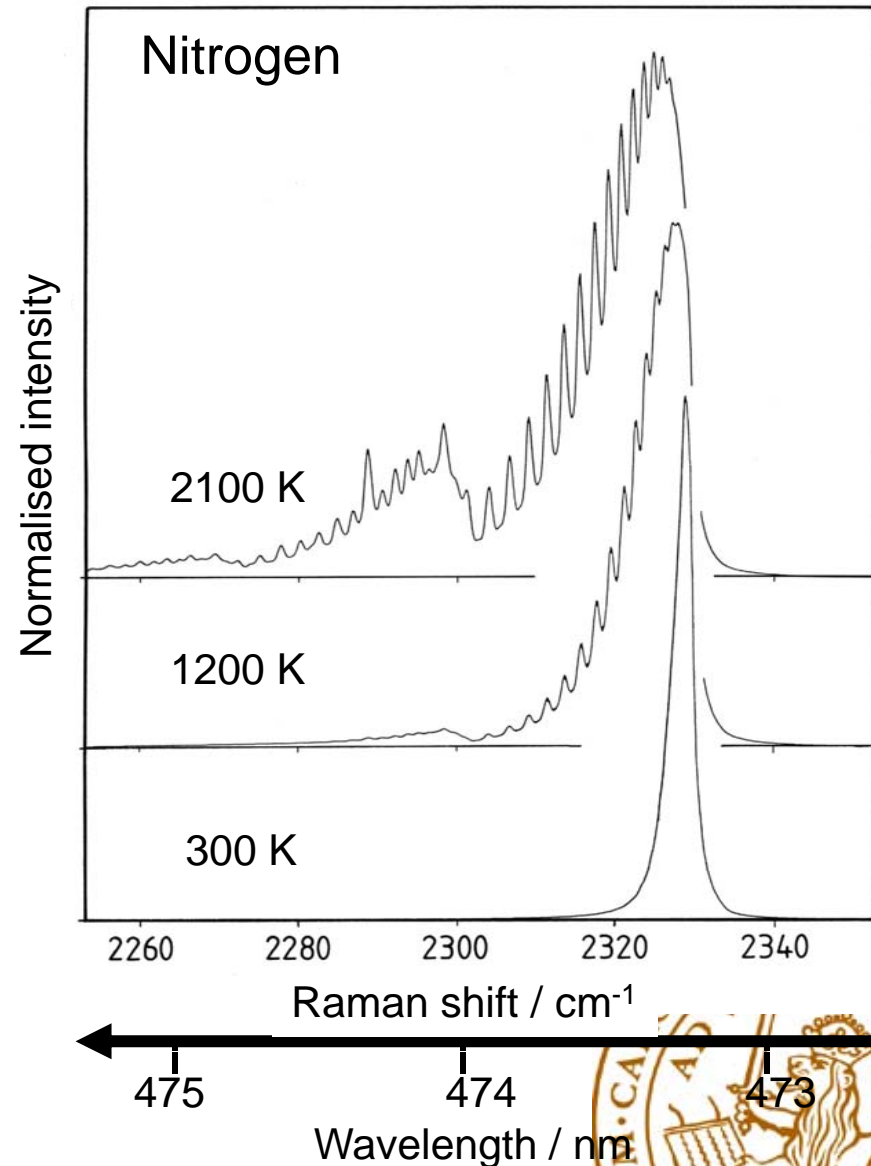


Experimental CARS spectra (scanned)

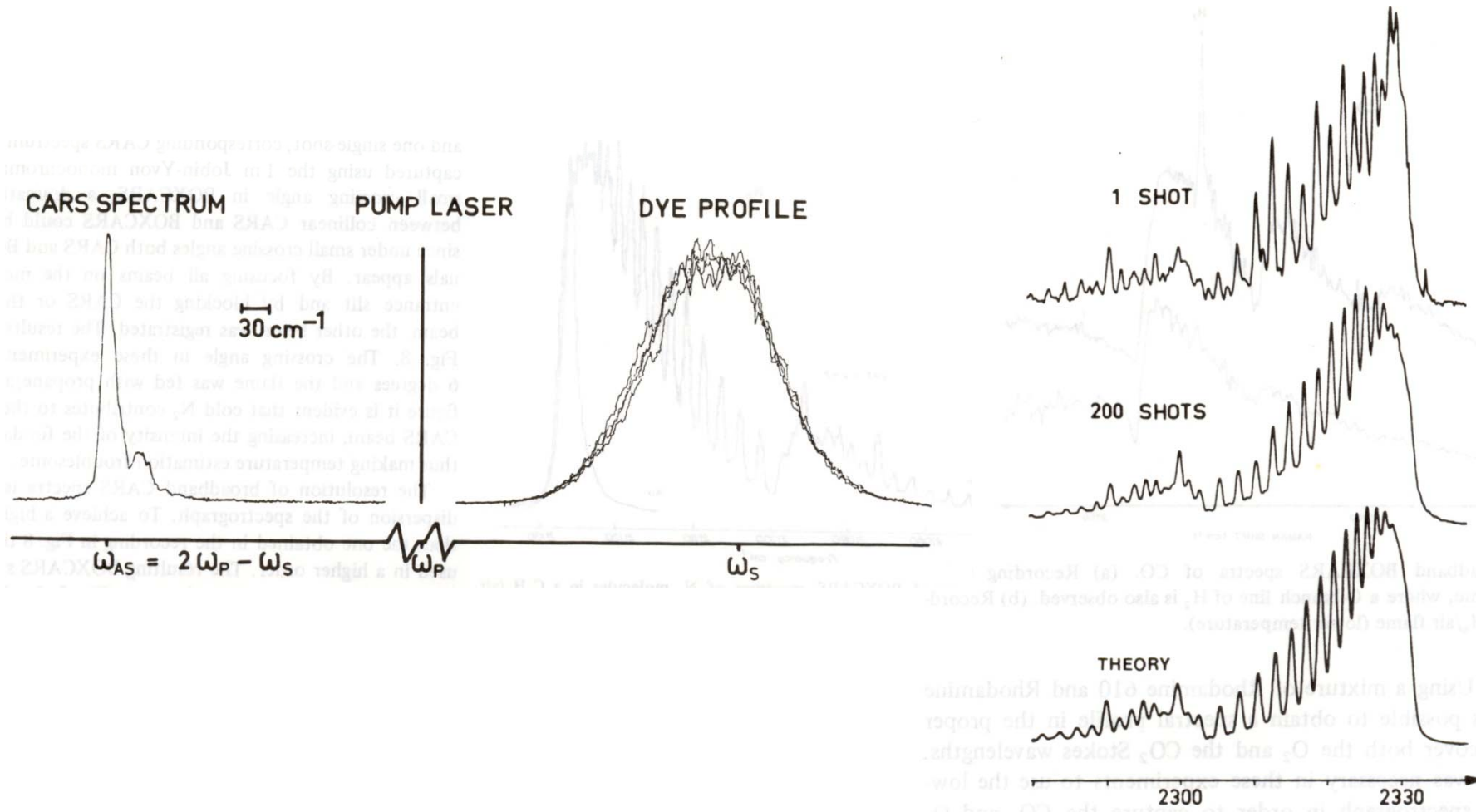


Temperature measurements

- The temperature is normally measured from the CARS spectrum of nitrogen, since nitrogen often is present at high concentration before as well as after combustion.
- The temperature is evaluated from the spectral shape and not the total intensity.
- The temperature is evaluated by fitting the experimental spectra using a library of theoretical spectra.



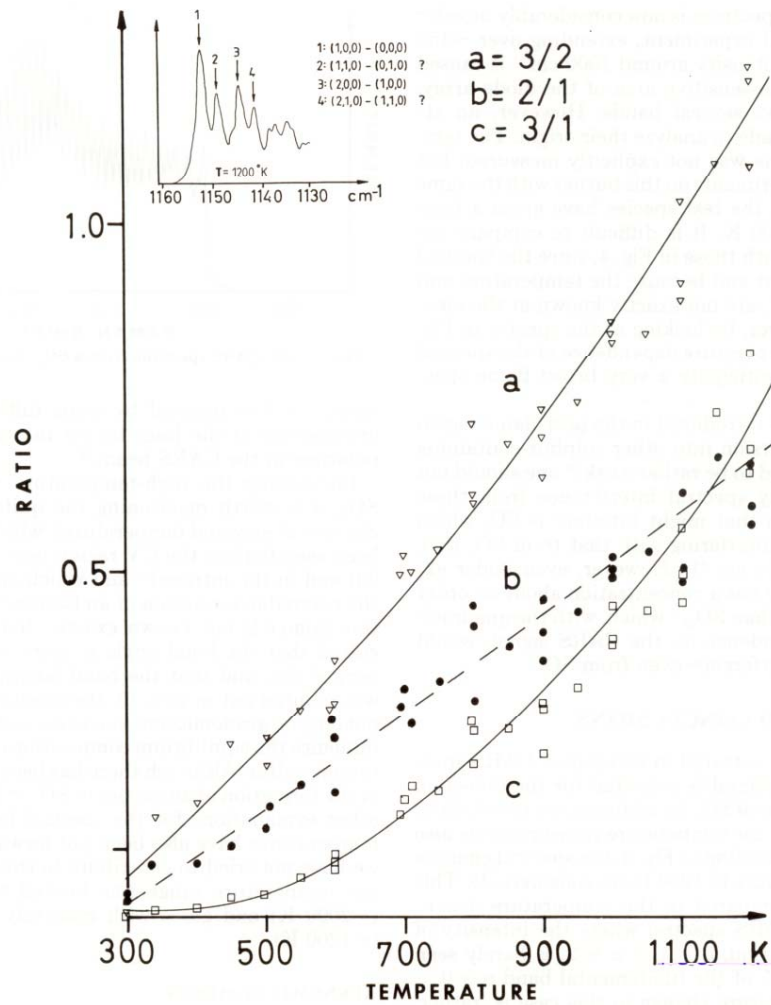
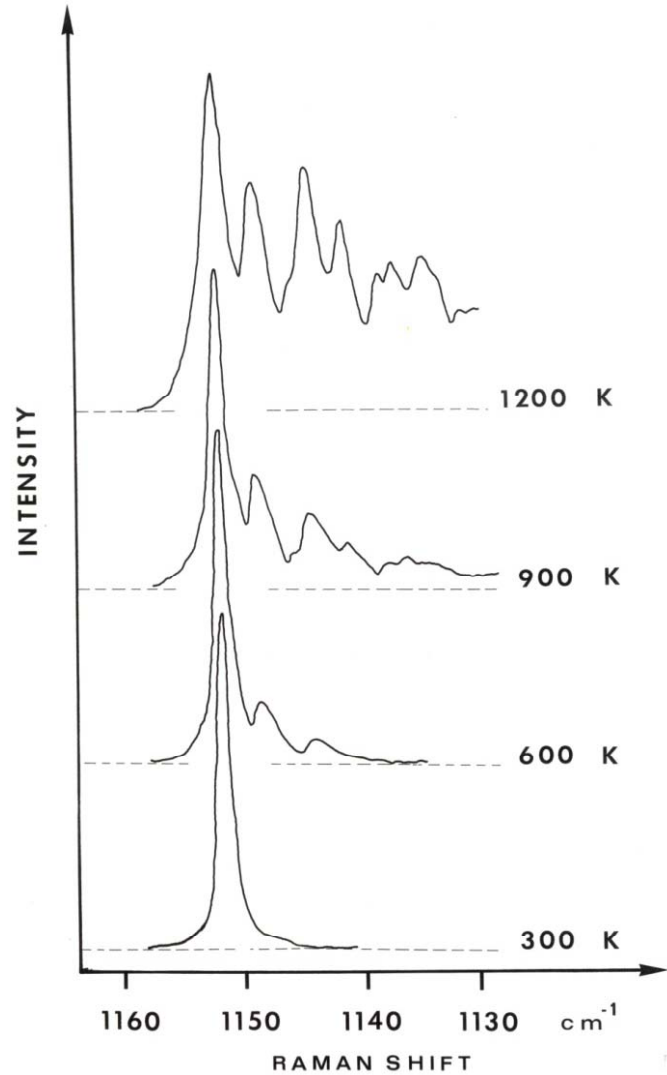
Broadband CARS (N₂)



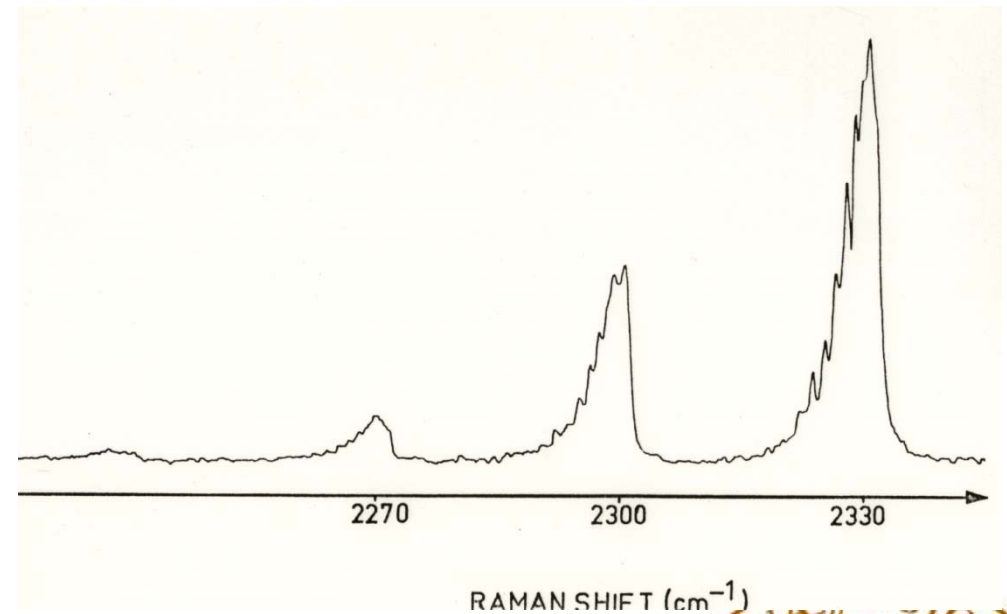
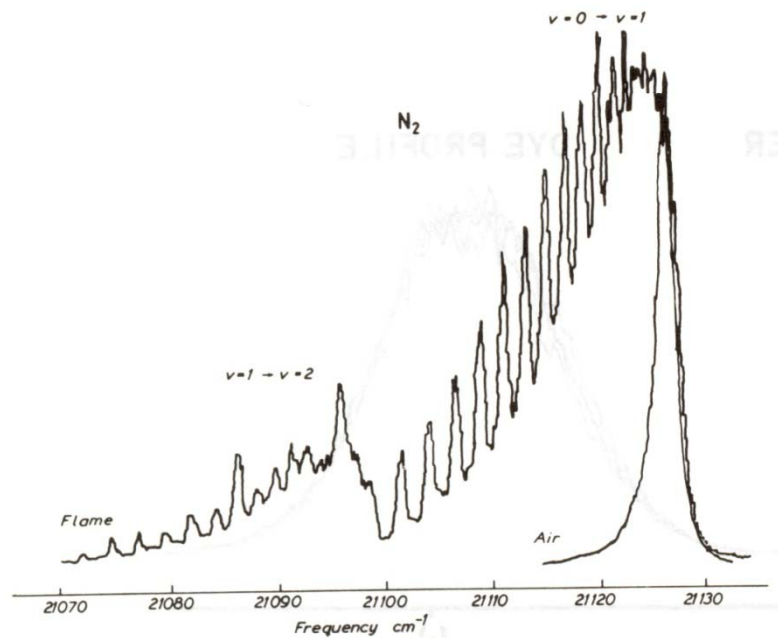
T = 1900 K



CARS on SO₂

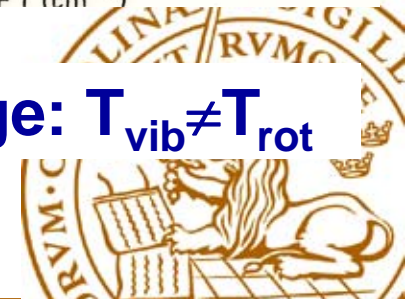


What does thermal equilibrium mean?



Flame: $T_{vib} = T_{rot}$

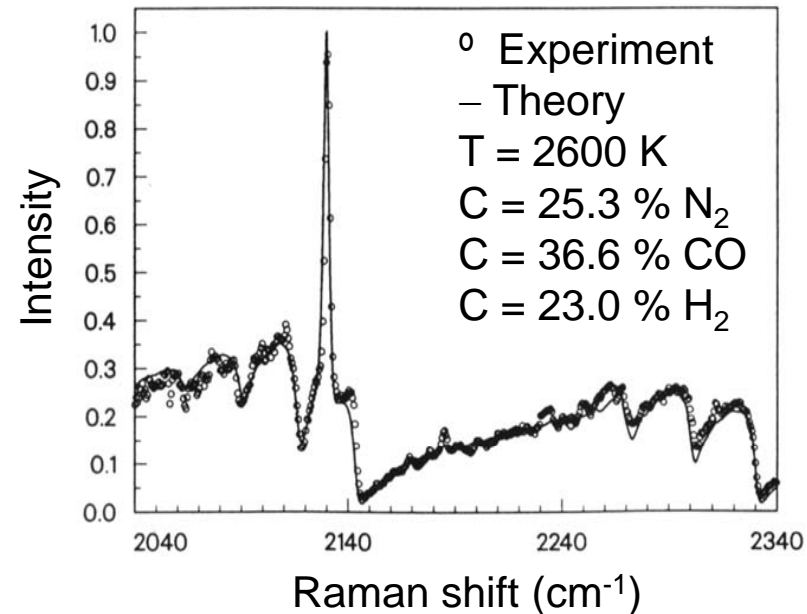
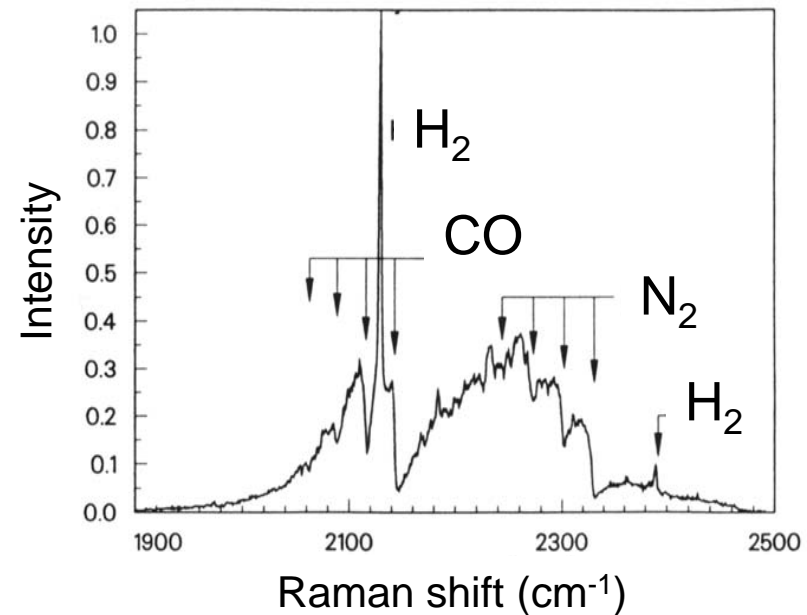
Discharge: $T_{vib} \neq T_{rot}$



Concentration measurements

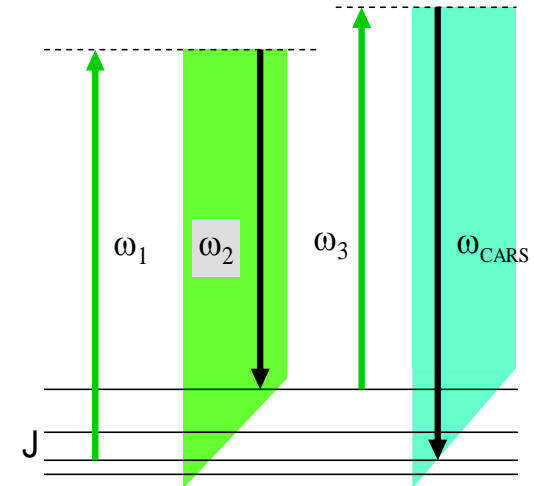
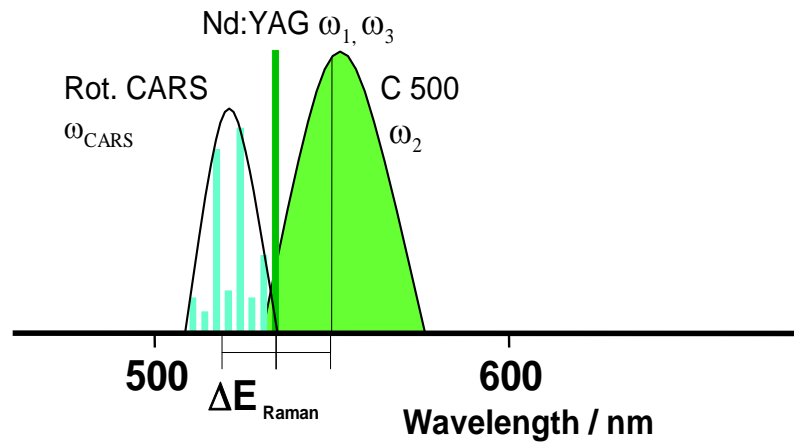
- Concentration measurements are normally made from the spectral shape.
- The figure shows an experimental spectrum with contributions from H₂, CO, and N₂.
- The lower figure shows the same experimental data together with a fitted curve. The evaluation gave all three concentrations together with the temperature.

Figures are from Stufflebeam and Eckbreth, Combust. Sci. and Tech. 66, 163-169 (1989)

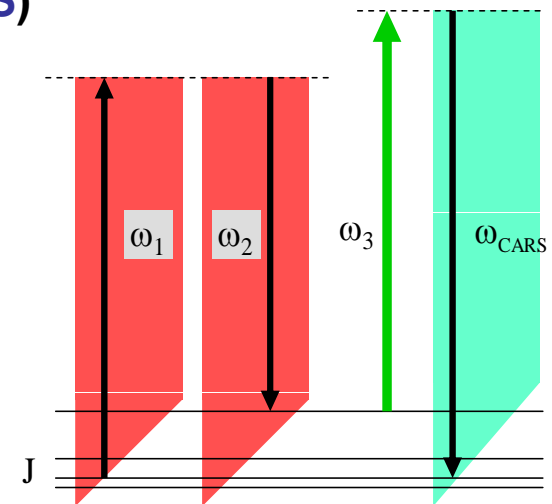
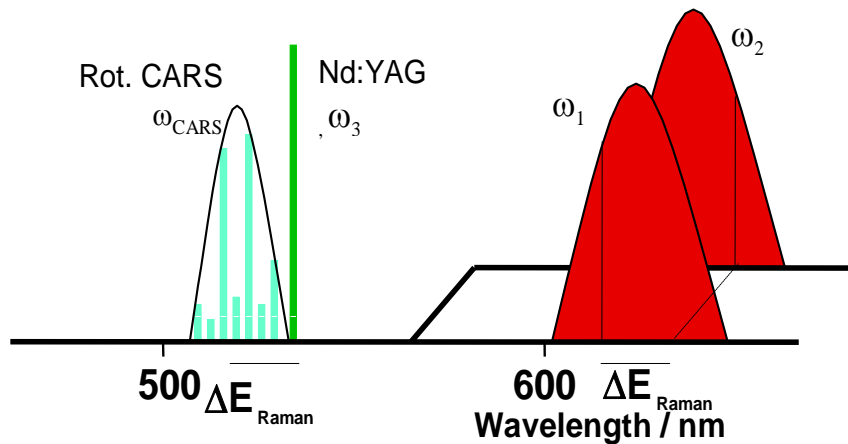


Rotational CARS

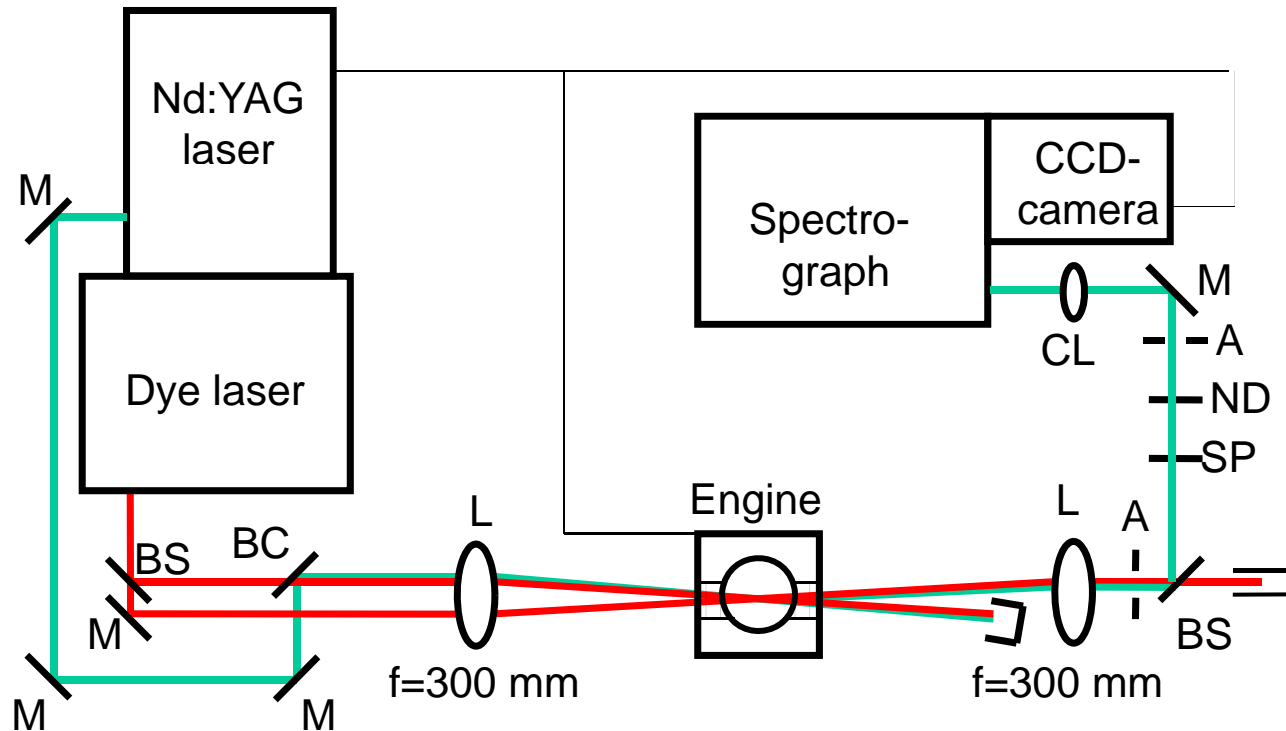
Conventional rotational CARS (C-RCARS)



Dual-broadband rotational CARS (DB-RCARS)



Experimental set-up for rotational CARS



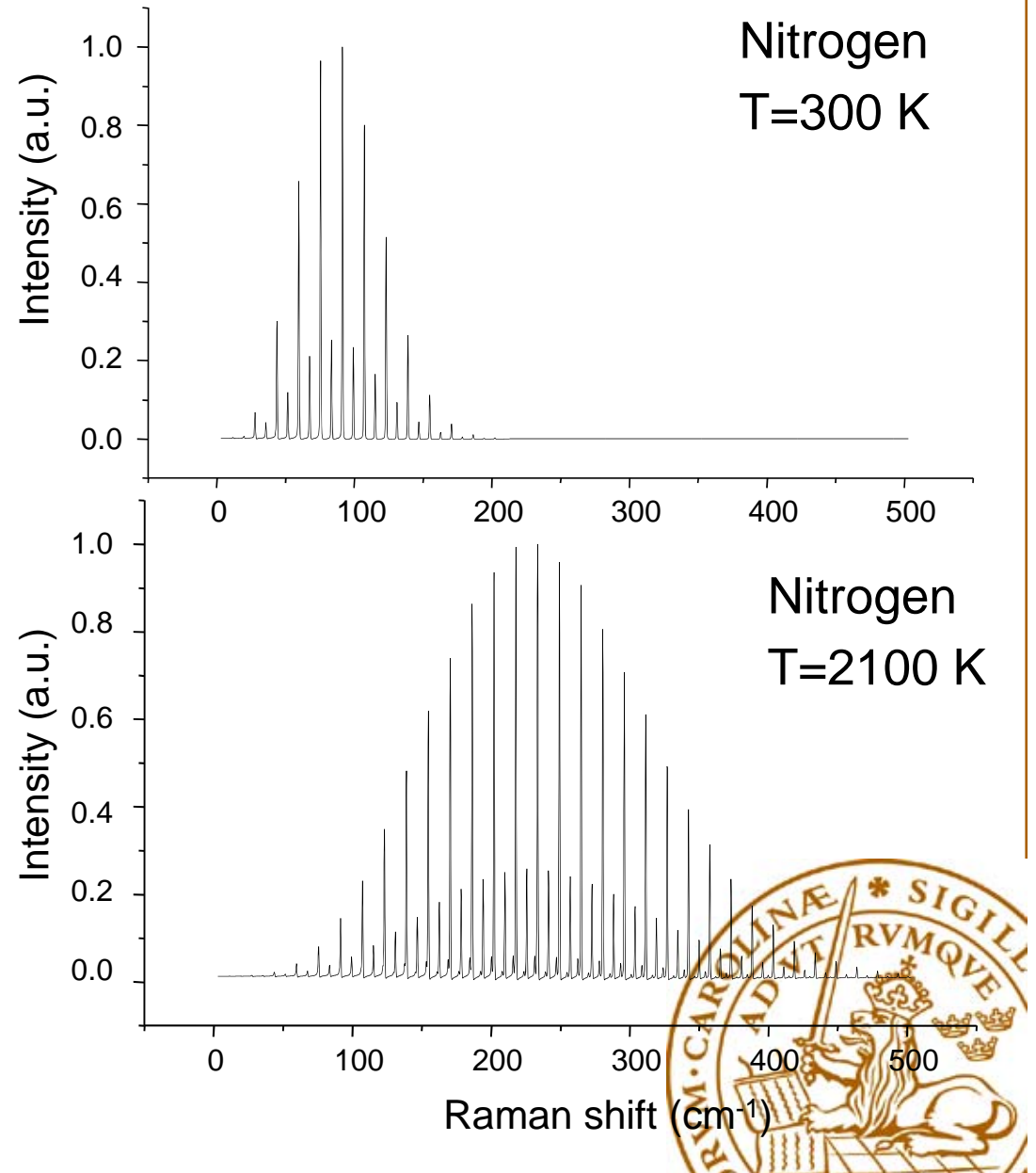
M = Mirror
BS = Beam splitter
BC = Beam combiner
A = Aperture
CL = Cylindrical lens
ND = Neutral density filter
CCD = Charge-coupled device camera
L = Lens
SP = Short-pass filter

- An experimental setup for dual-broadband rotational CARS is using basically the same experimental parts as for vibrational CARS.
- Some optical components such as mirrors and filters are replaced.



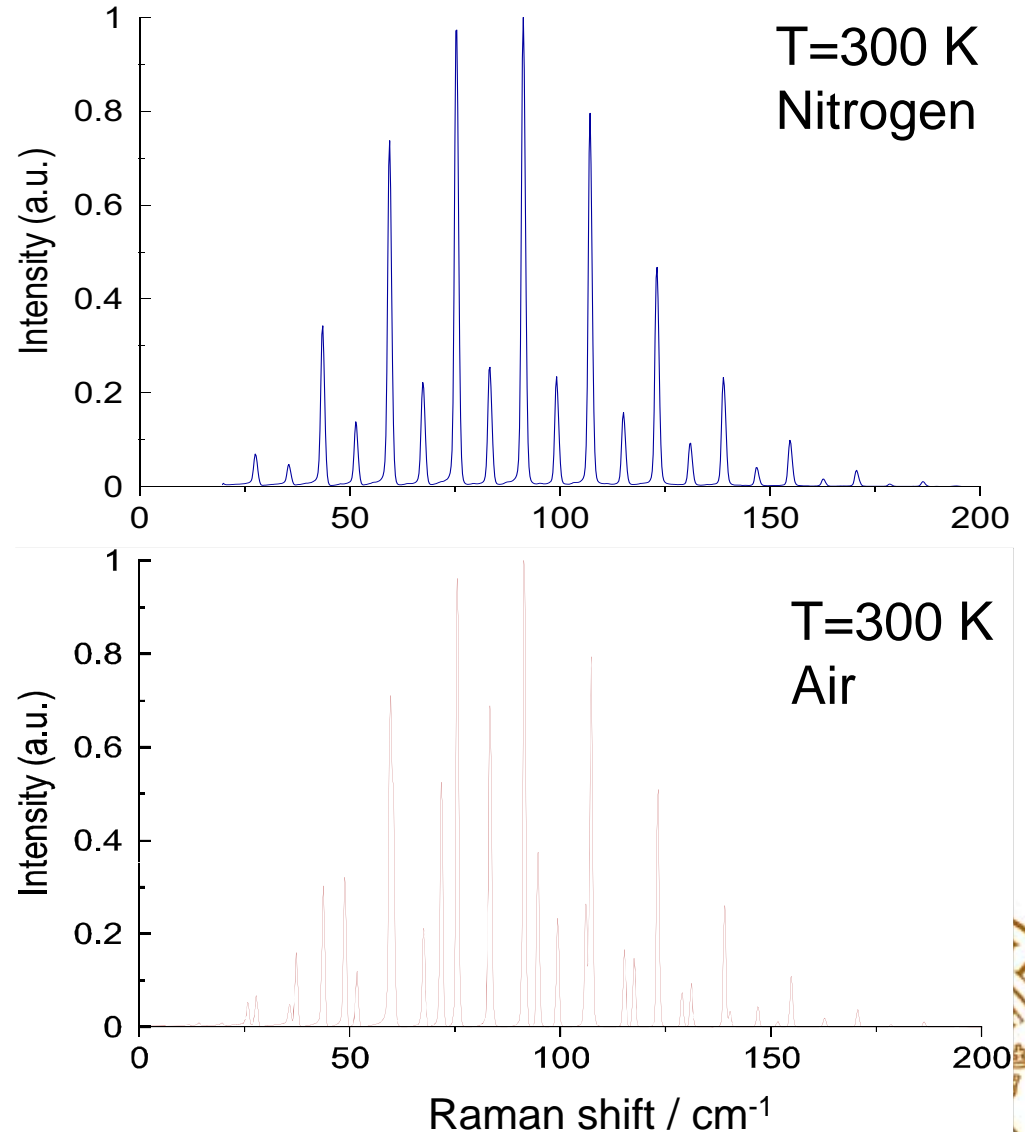
Temperature measurements

- The spectral shape of a rotational CARS spectrum varies strongly as a function of temperature.



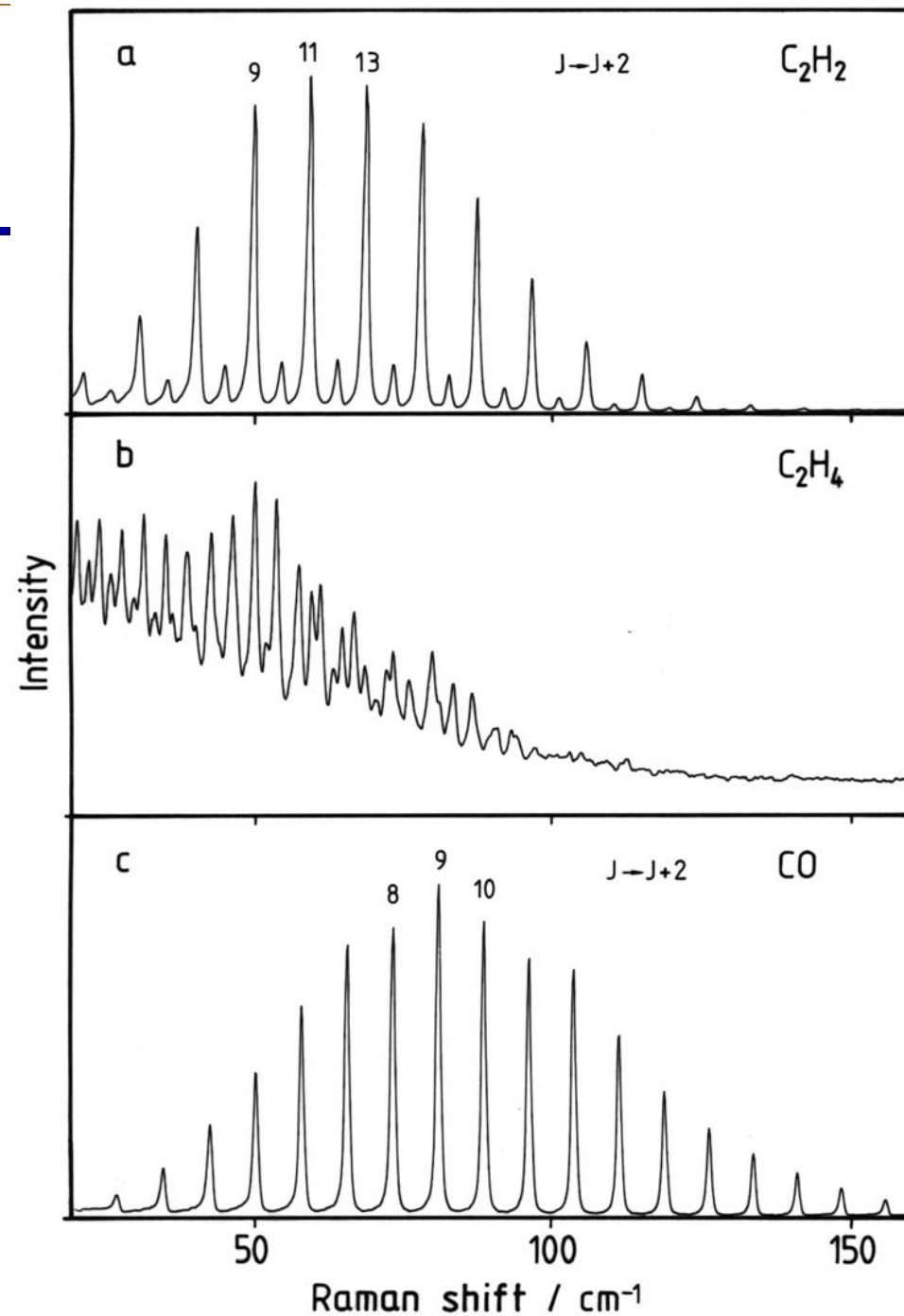
Concentration measurements

- Rotational CARS spectra from many species (N_2 , O_2 , CO , CO_2) appear in the same spectral region.
- Concentration evaluation is made from the relative intensities of the lines.



Rotational CARS spectra for different molecules

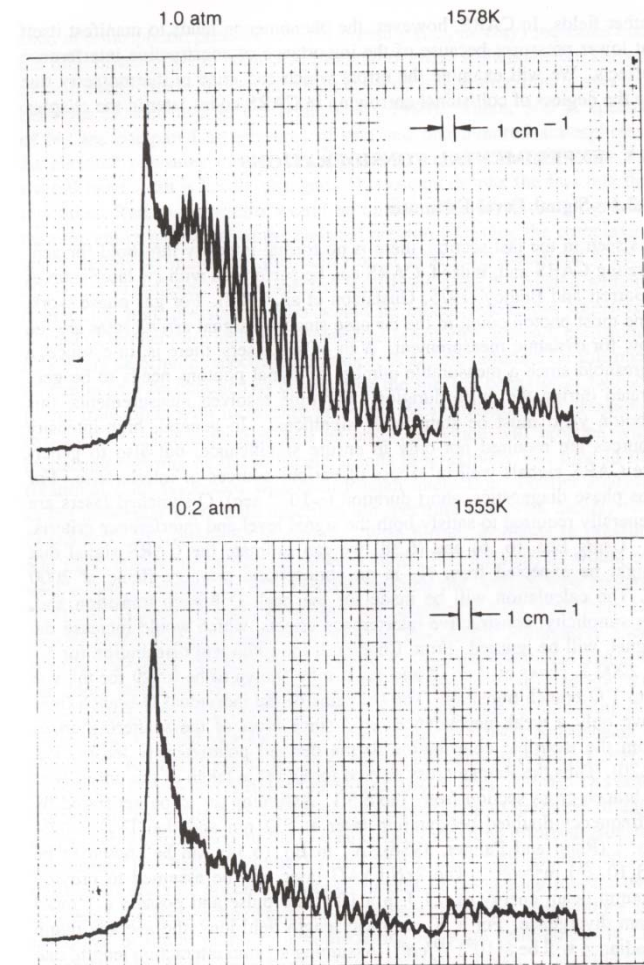
- Linear molecules give clear distinct spectra.
- Smaller molecules have higher B-constants and result in wider spectra.



Advantages with CARS for practical applications

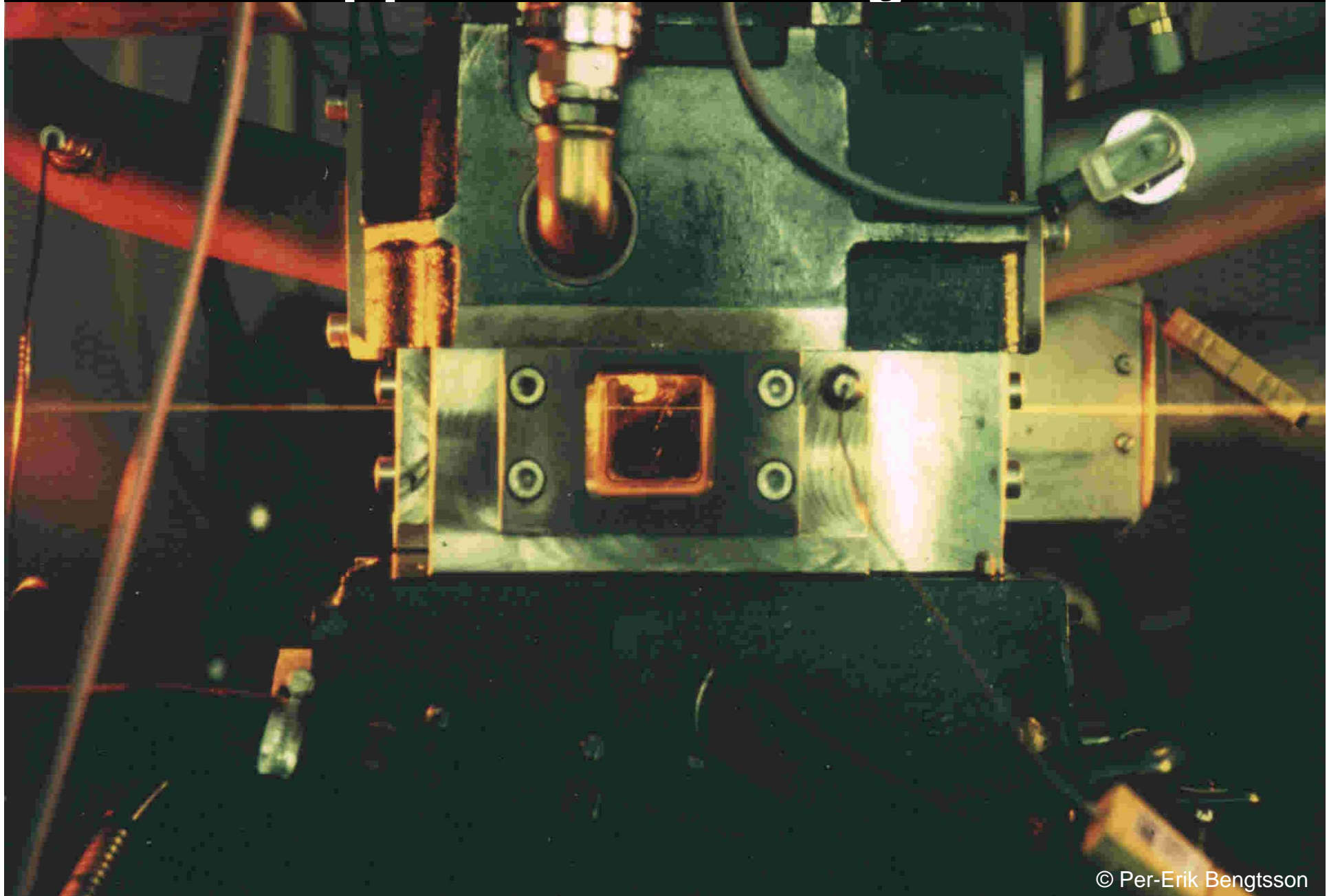
- Signal as a new laser beam
- Strong signal
- CARS signal on the anti-Stokes side
- Measurements on nitrogen molecules (max signal)

Special issue: High pressure effects



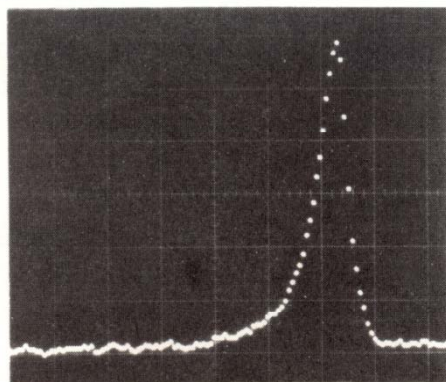
Ref: Eckbreth et al.

Applications to engines

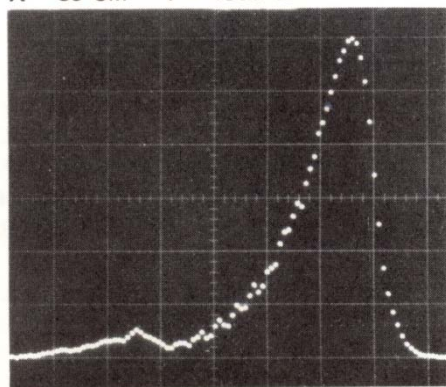


Vibrational CARS applications

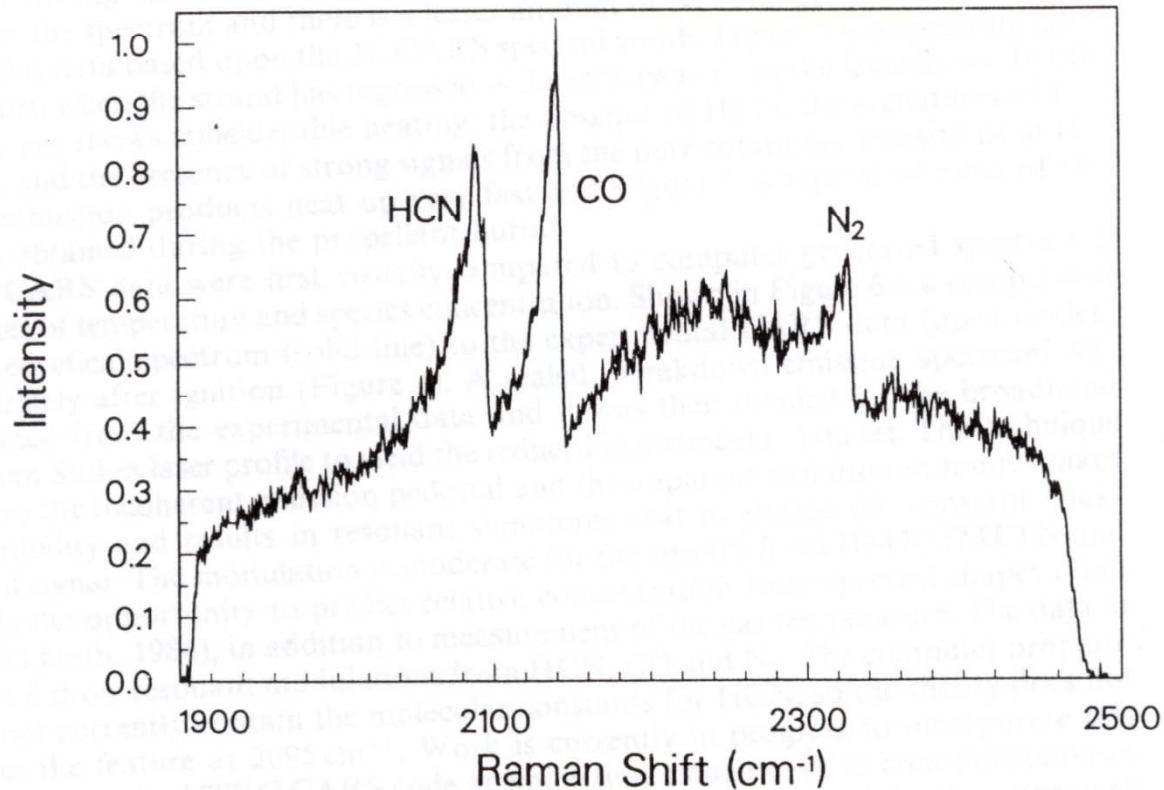
X = 6 CM T = 900 K



X = 39 CM T = 1500 K



FREQUENCY →
0.586 $\text{cm}^{-1}/\text{DOT}$



CARS spectra:

In a burning spray of JET-A

Burning of solid propellants

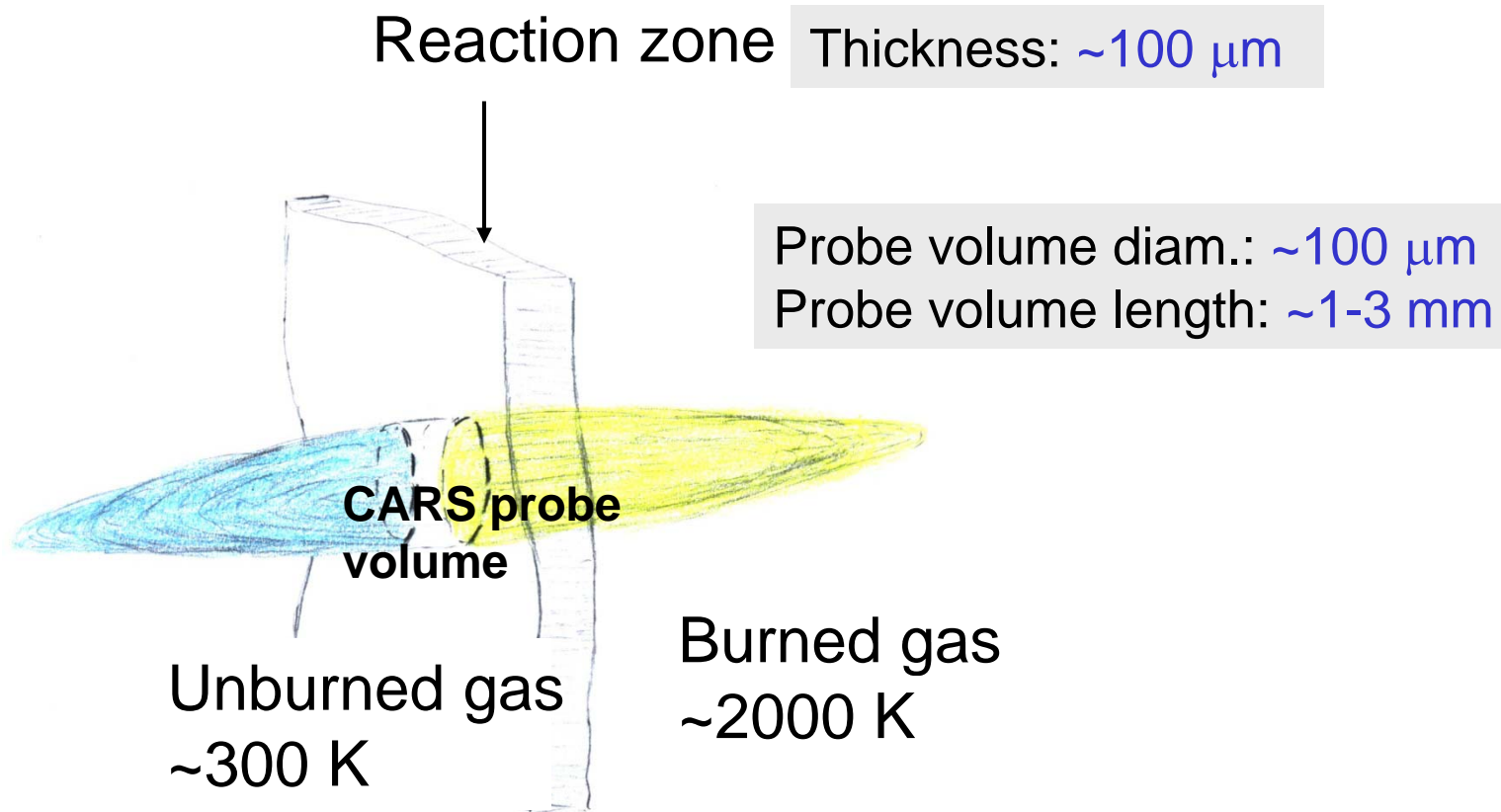


Limitations with conventional CARS

- **Low spatial resolution** - In order to keep a sufficient signal strength it has been important to keep the crossing angle between the laser beams small \Rightarrow rather low spatial resolution ($\sim >1-3$ mm)
- Limited signal strength in an environment with low transmission - The CARS signal scales as (laser intensity)³ \Rightarrow at strong laser attenuation, the signal strength is too low



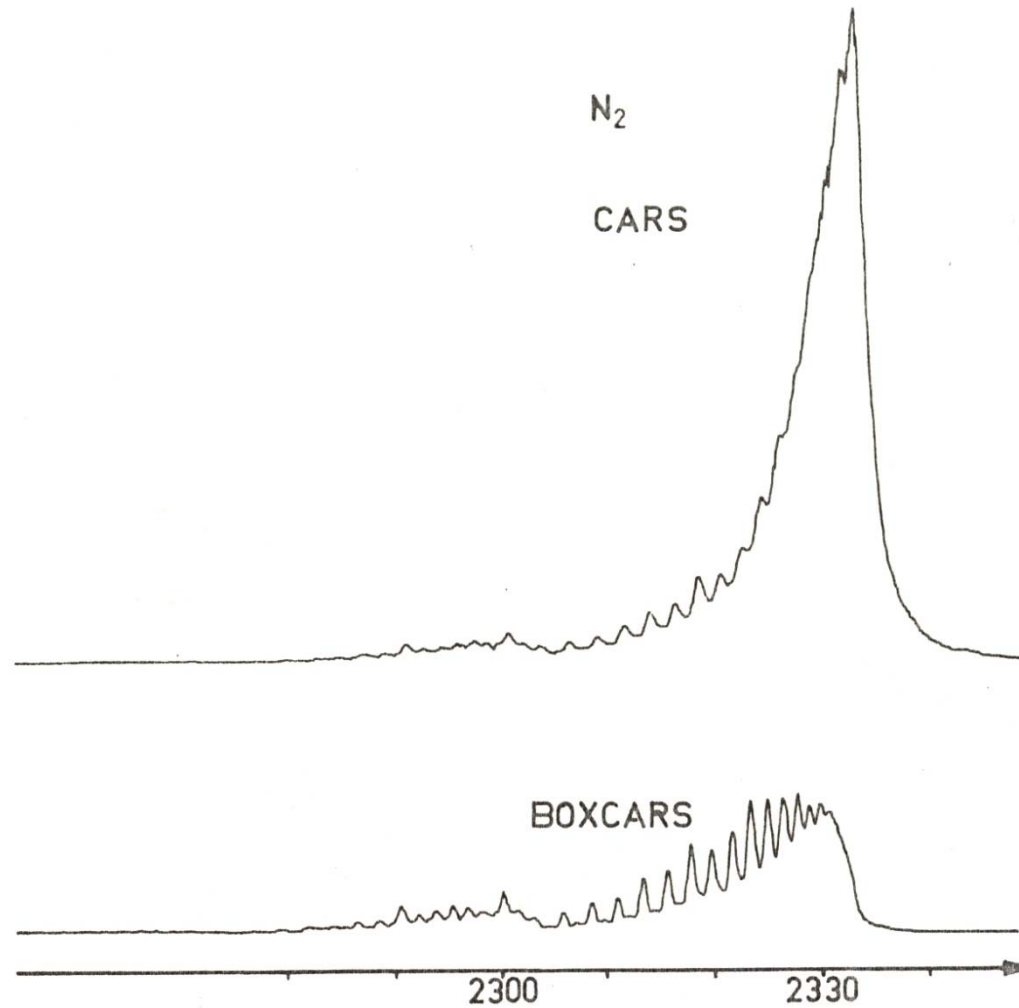
Probe volume considerations



The result will be a “mixed spectrum” with a strong low-temperature part



Comparison between collinear CARS and BOXCARS



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- **Limited signal strength in an environment with low transmission** - The CARS signal scales as $(\text{laser intensity})^3 \Rightarrow$ at strong laser attenuation, the signal strength is too low

Potential solution: $2-\lambda$ CARS



One measuring situation where $2\text{-}\lambda$ CARS has to be used



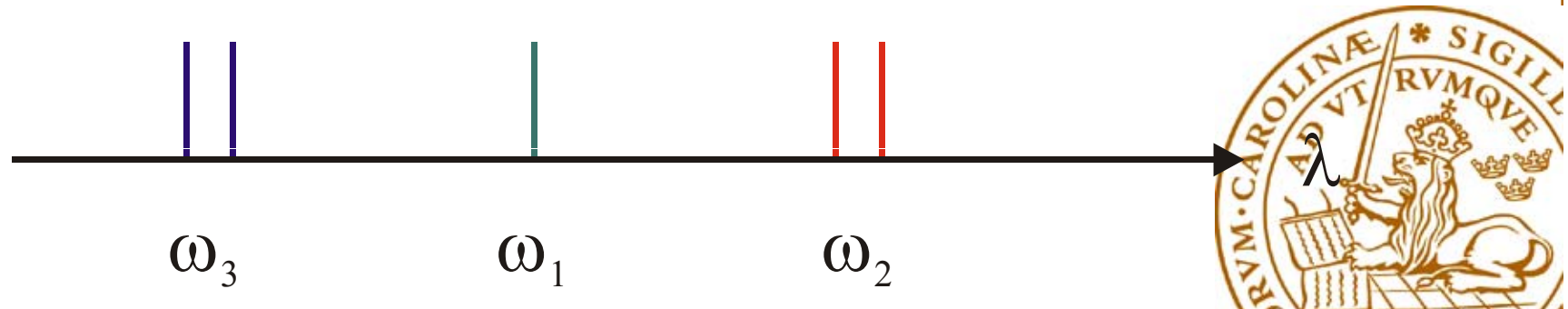
Concept of 2- λ CARS:

Use a two colour dye laser instead of a broad-band dye laser \Rightarrow higher spectral intensity

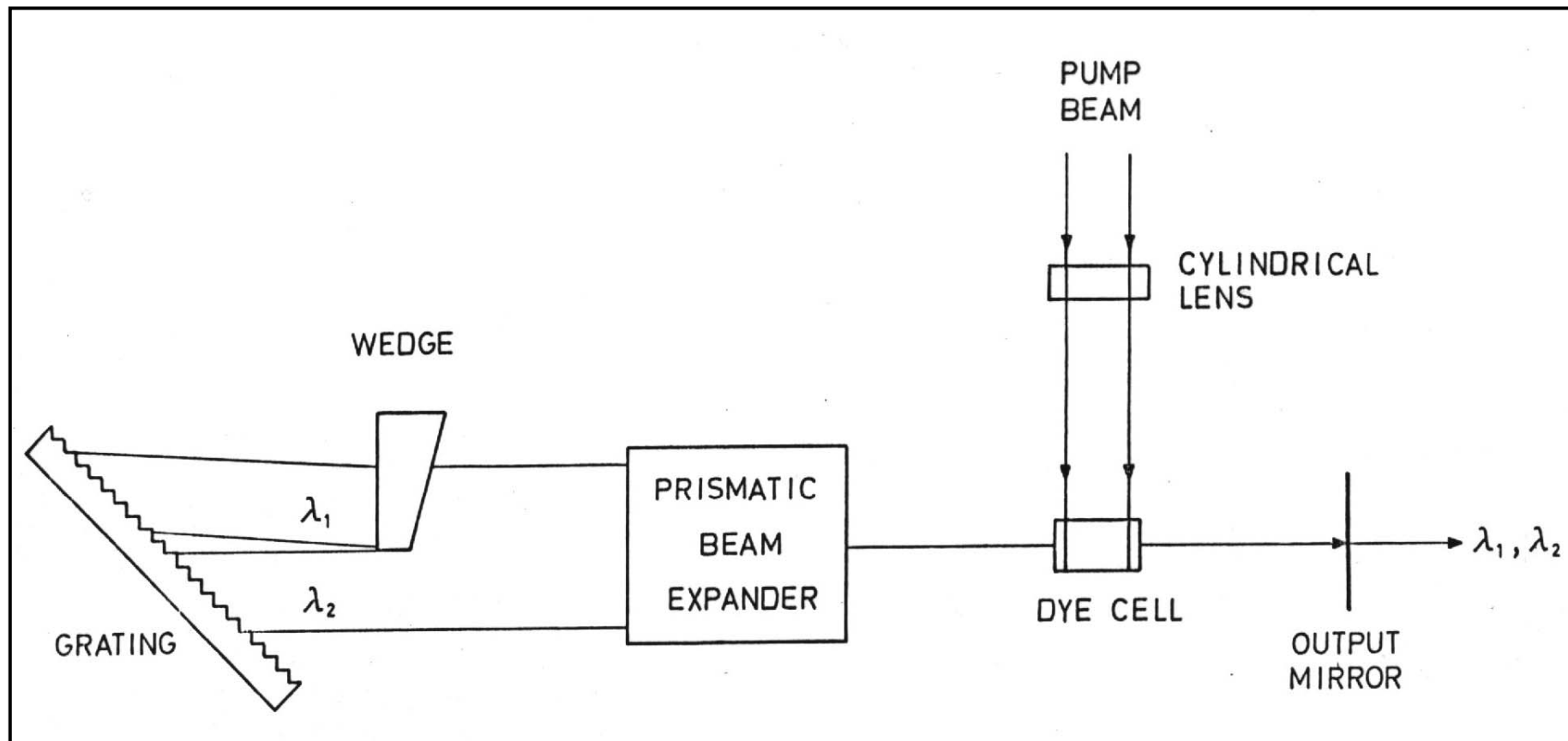
Broadband CARS



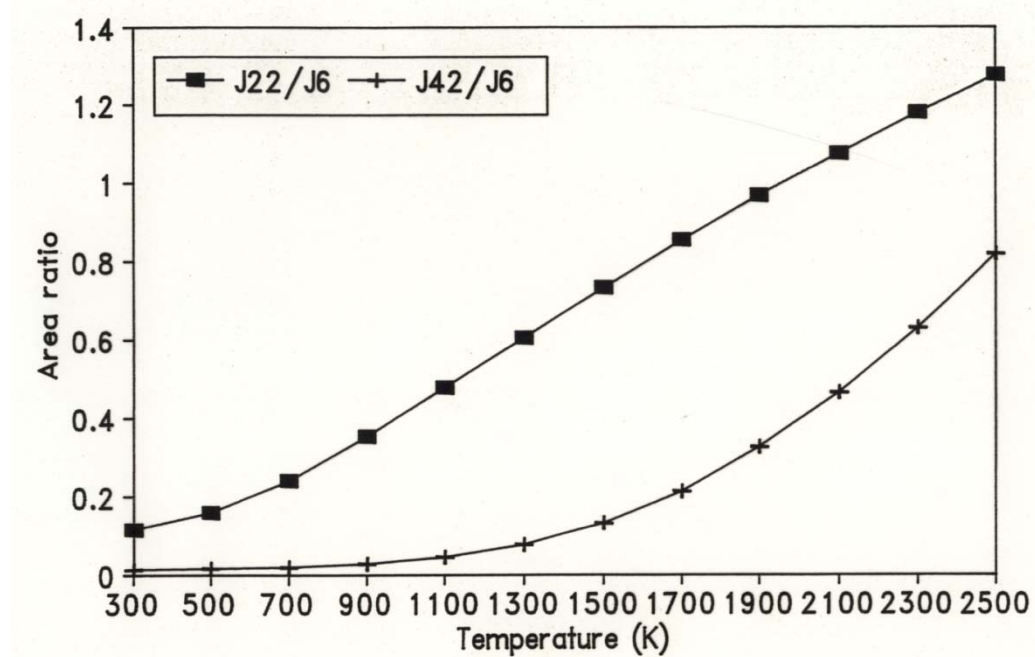
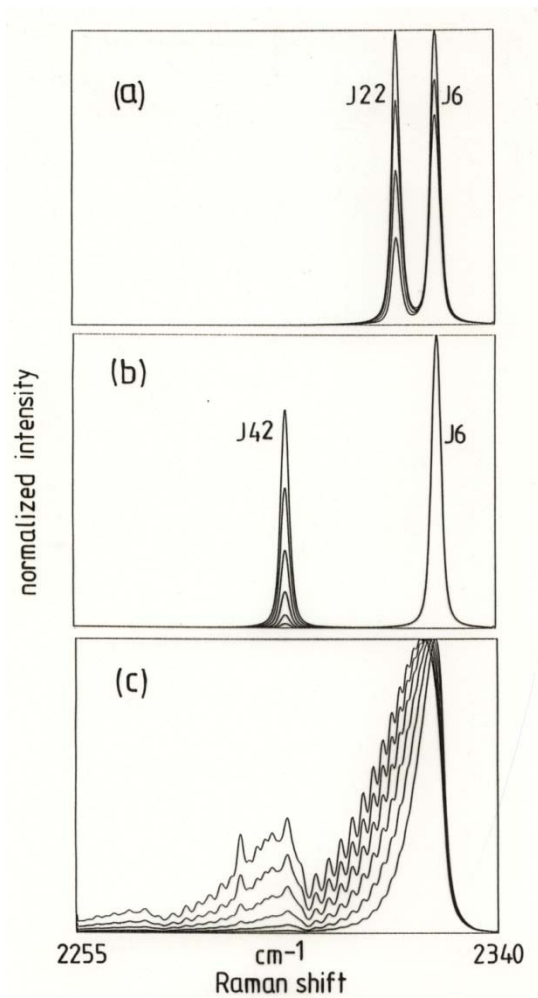
2 λ CARS



Experimental approach to produce a two colour dye laser



Wavelength selection in 2- λ CARS



Features of $2\text{-}\lambda$ CARS

Advantages:

- **~ 30 times higher signal intensity \Rightarrow Higher spatial resolution, experiments in very applied areas possible**

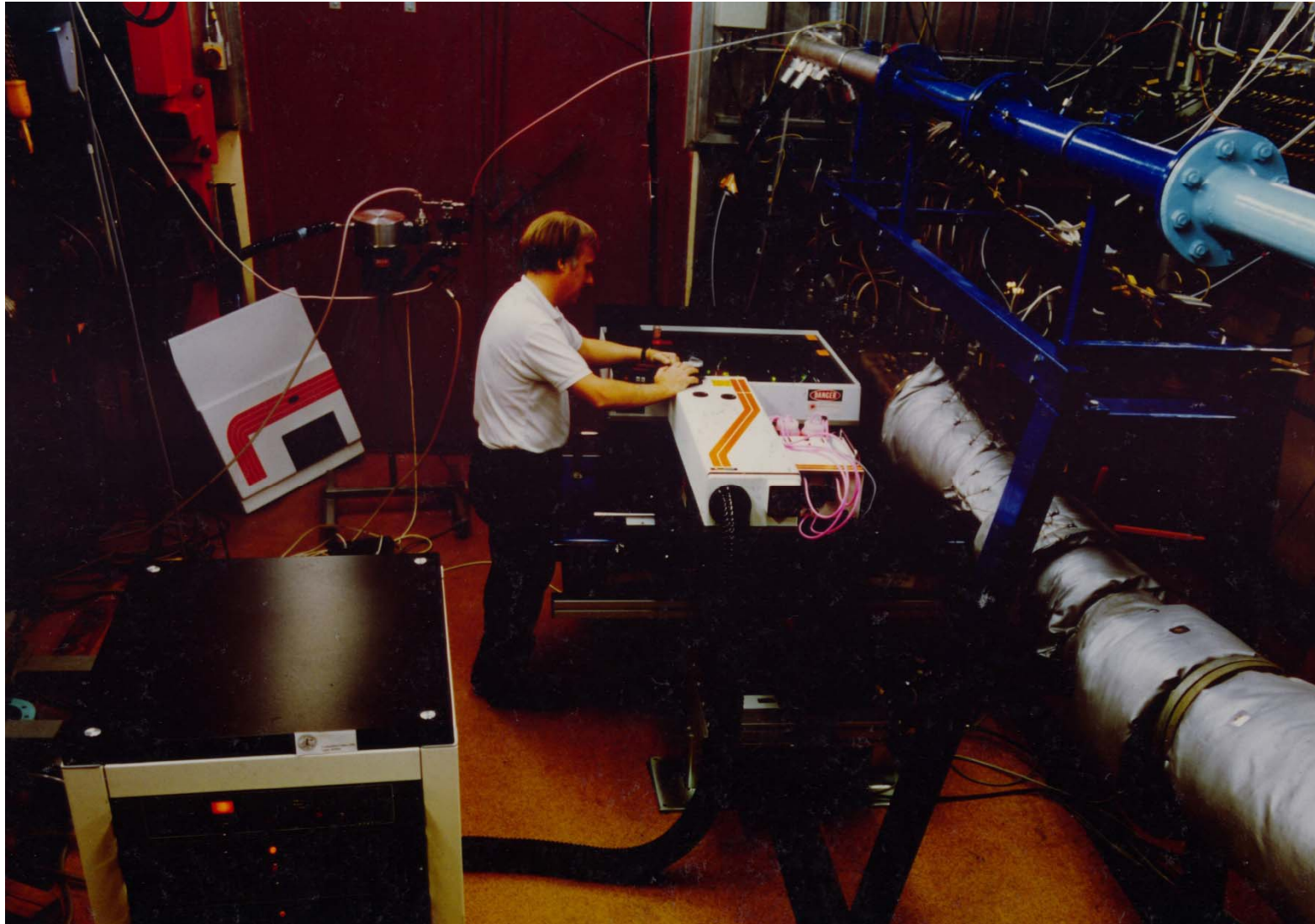
Disadvantages:

- **More complex experimental set-up**
- **Somewhat lower temperature precision ($\sim 5\%$ comp to $\sim 3\%$ with broadband CARS)**



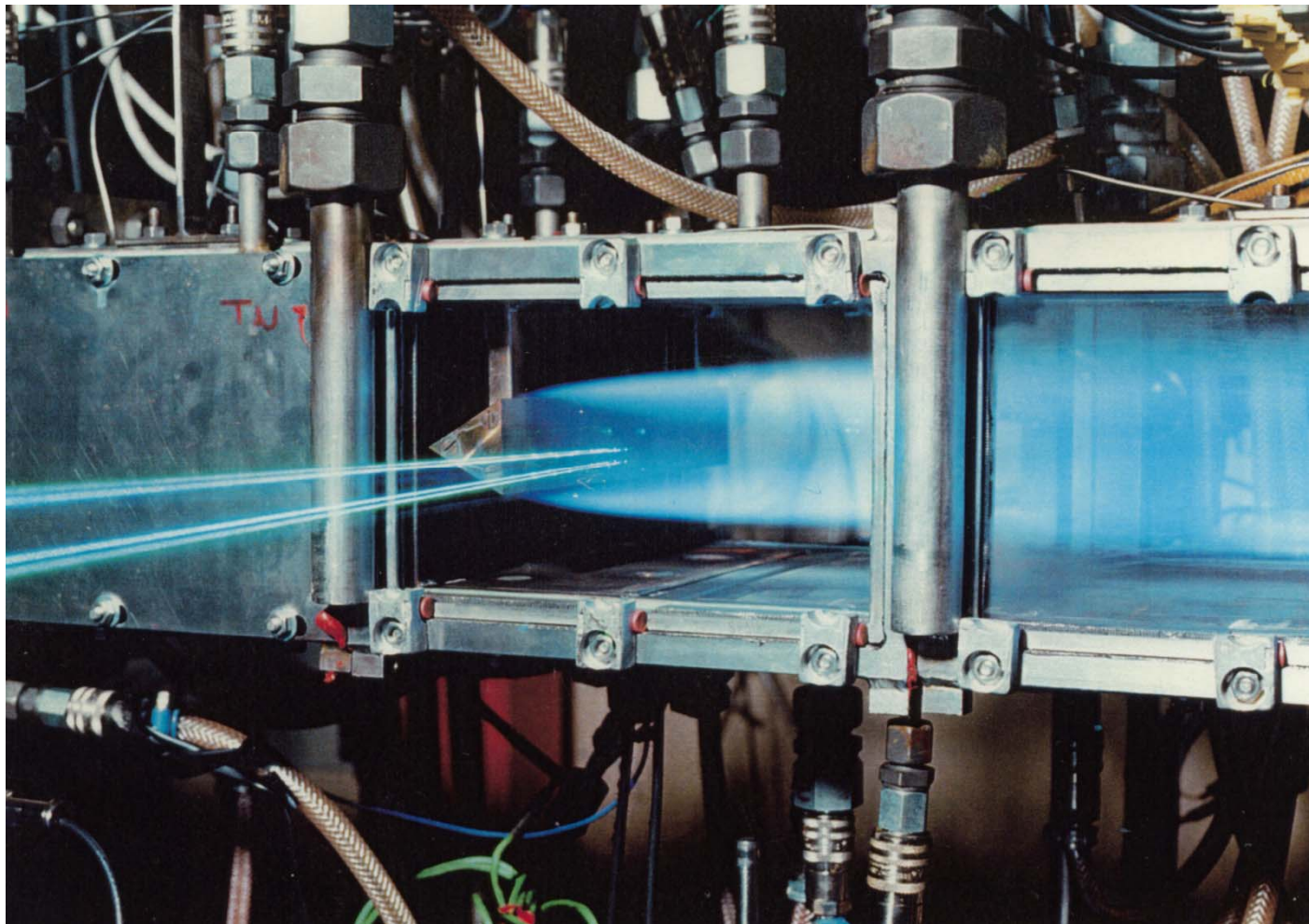
Application: High spatially resolved temperature measurement in model of an after-burner

(VOLVO Aero Corporation, Trollhättan)



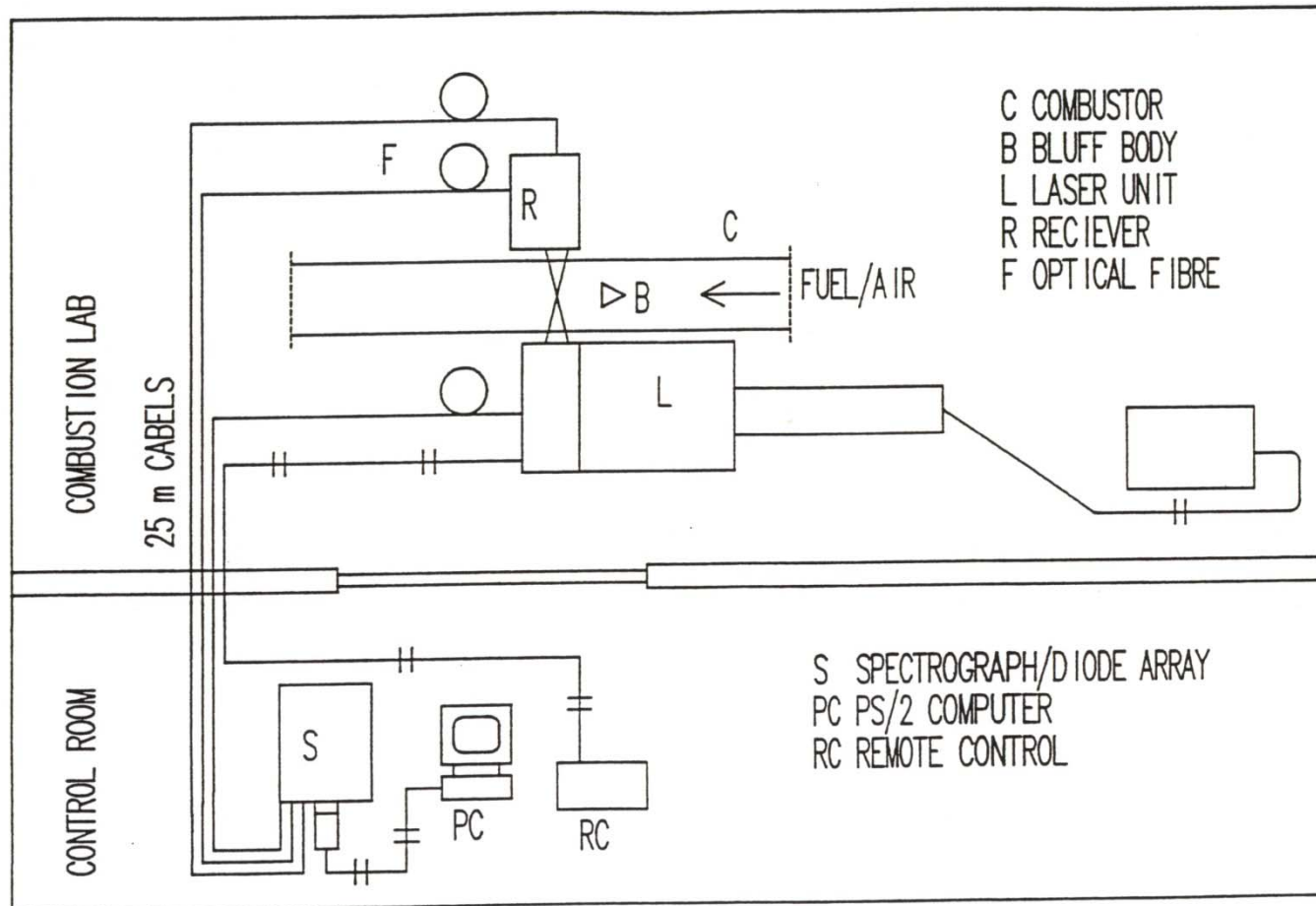
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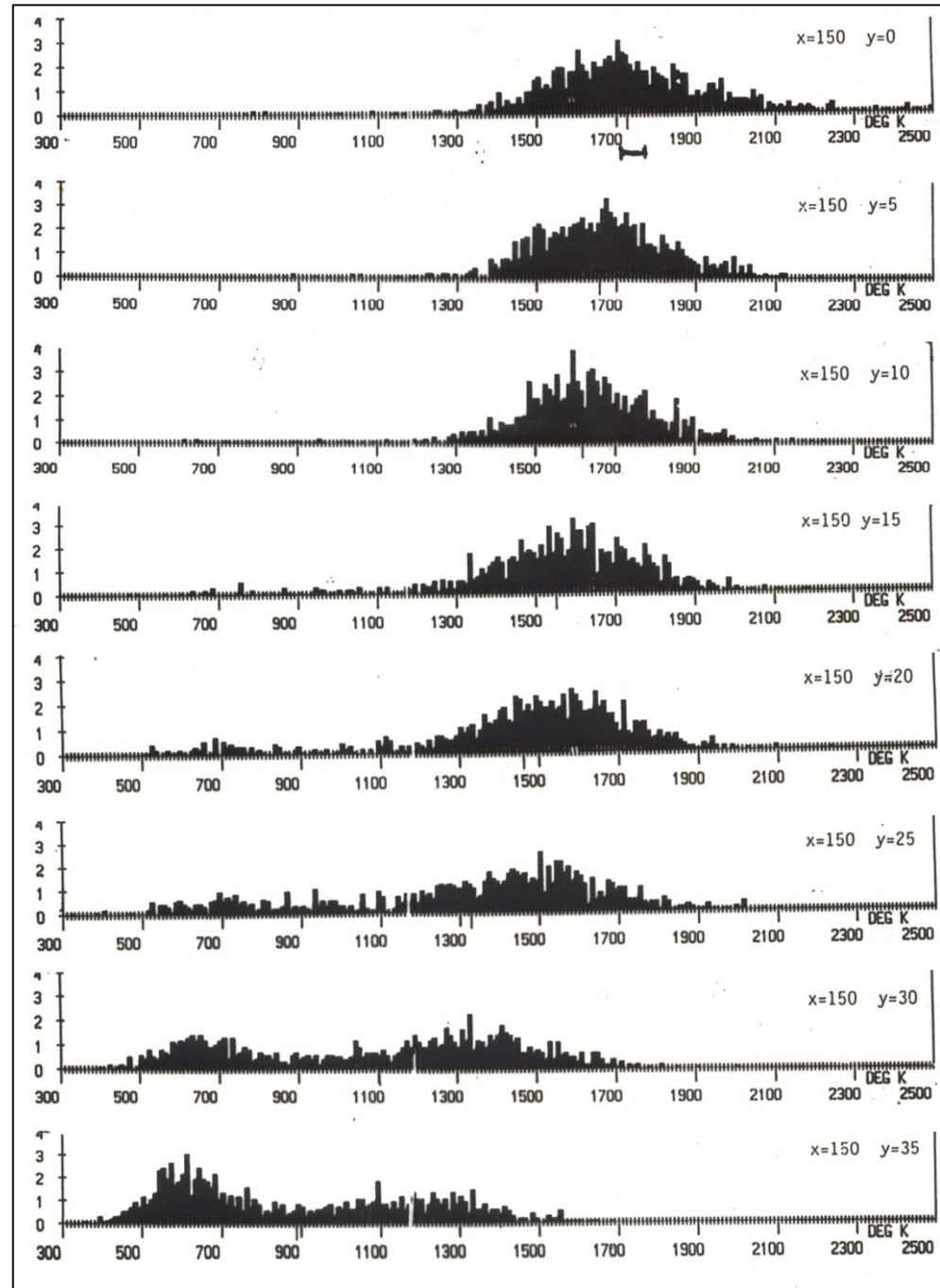


Application: High spatially resolved temperature measurement in model of an after-burner

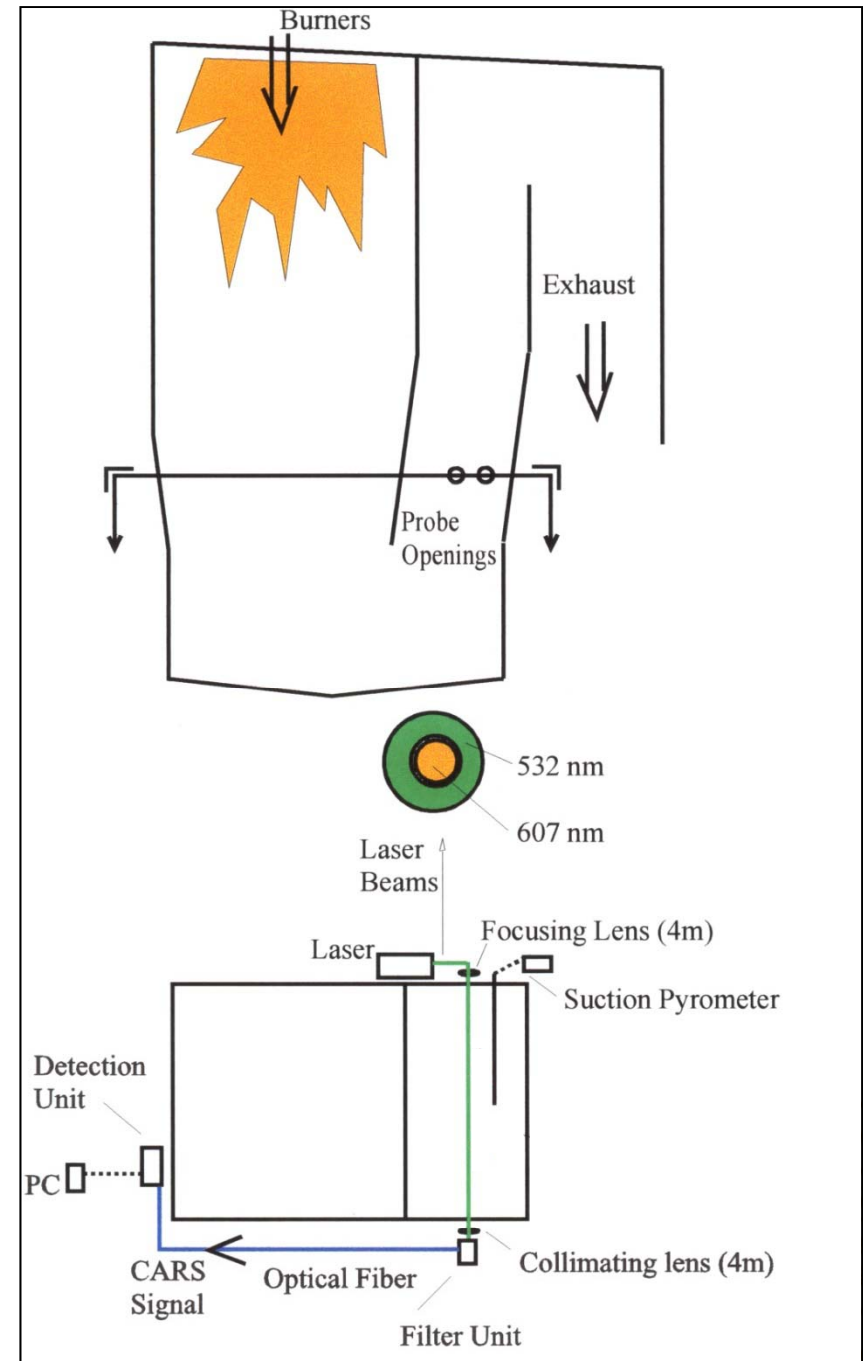
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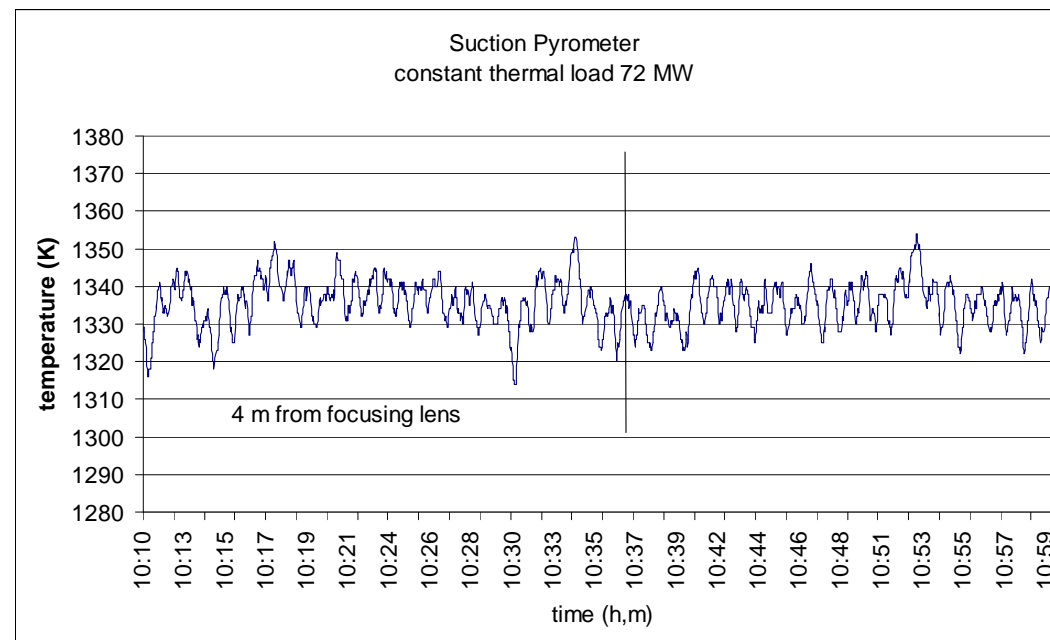
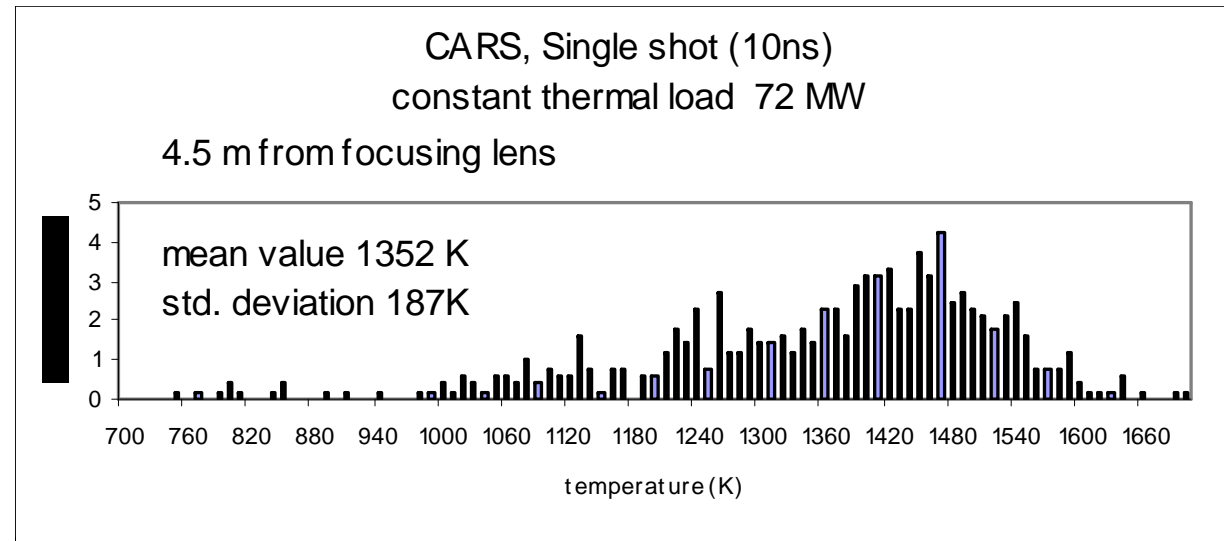
Temperature pdf:s at a distance 150 mm from the bluff body at different radial distances



Practical diagnostics - CARS



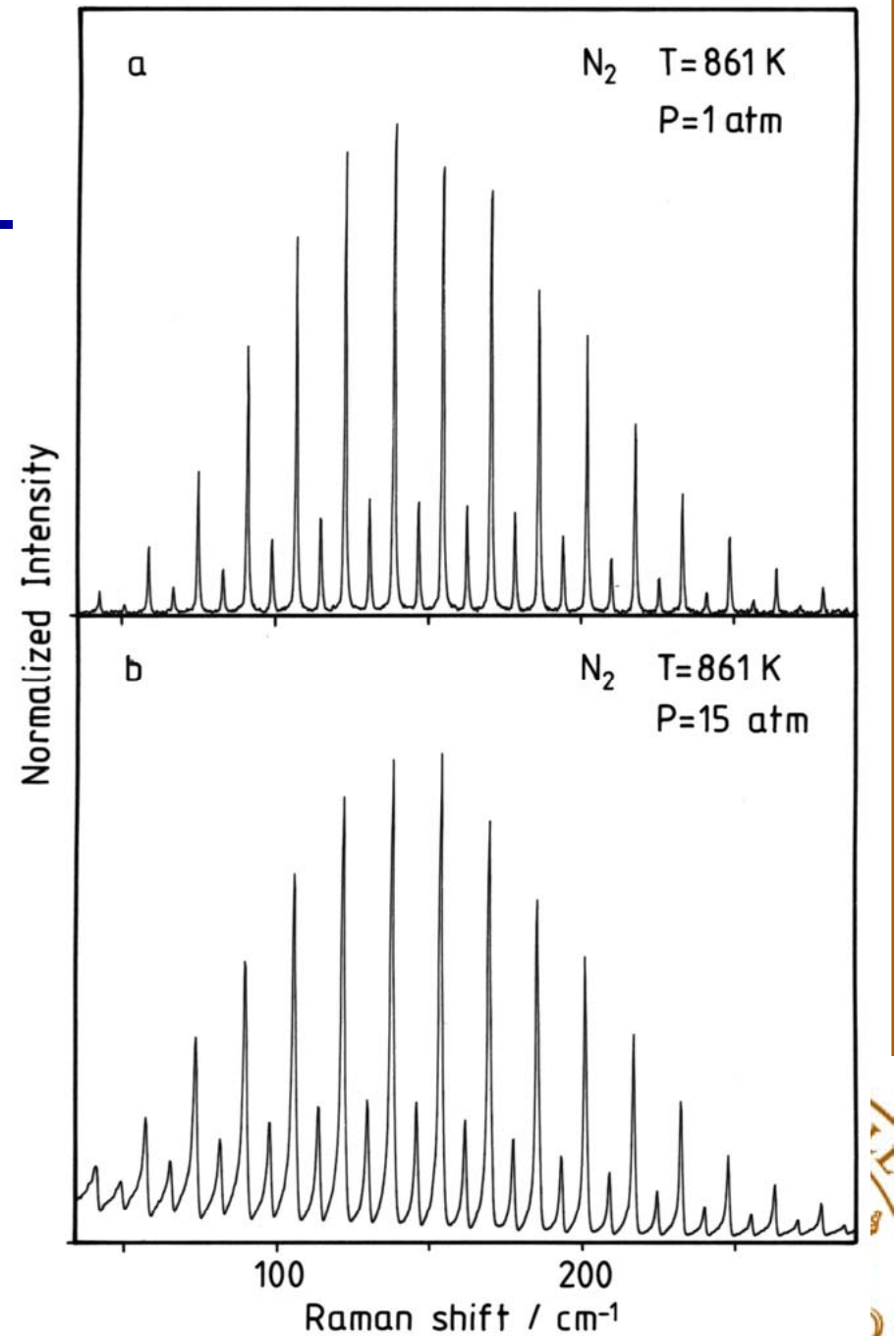
Practical diagnostics - CARS



Pressure dependence for rotational CARS

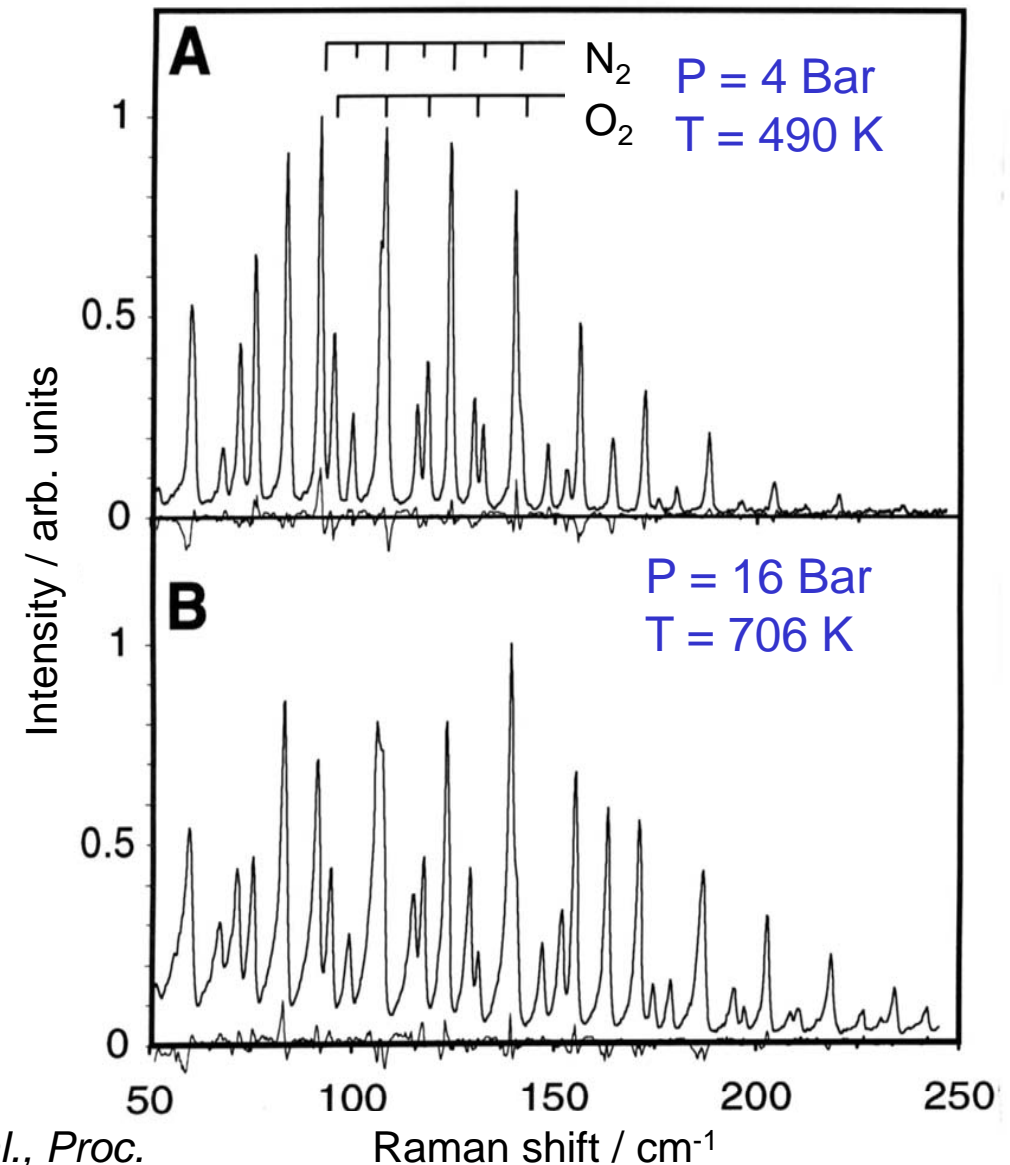
If pressure increases

- the signal increases
- the linewidths will be broader



Rotational CARS spectra from an engine

- Single-shot spectra in the compression phase of engine burning natural gas.
- Both nitrogen and oxygen lines are observed.
- Below each experimental spectrum the difference between the experimental spectrum and the best-fit theoretical spectrum is shown, illustrating good spectral fits.

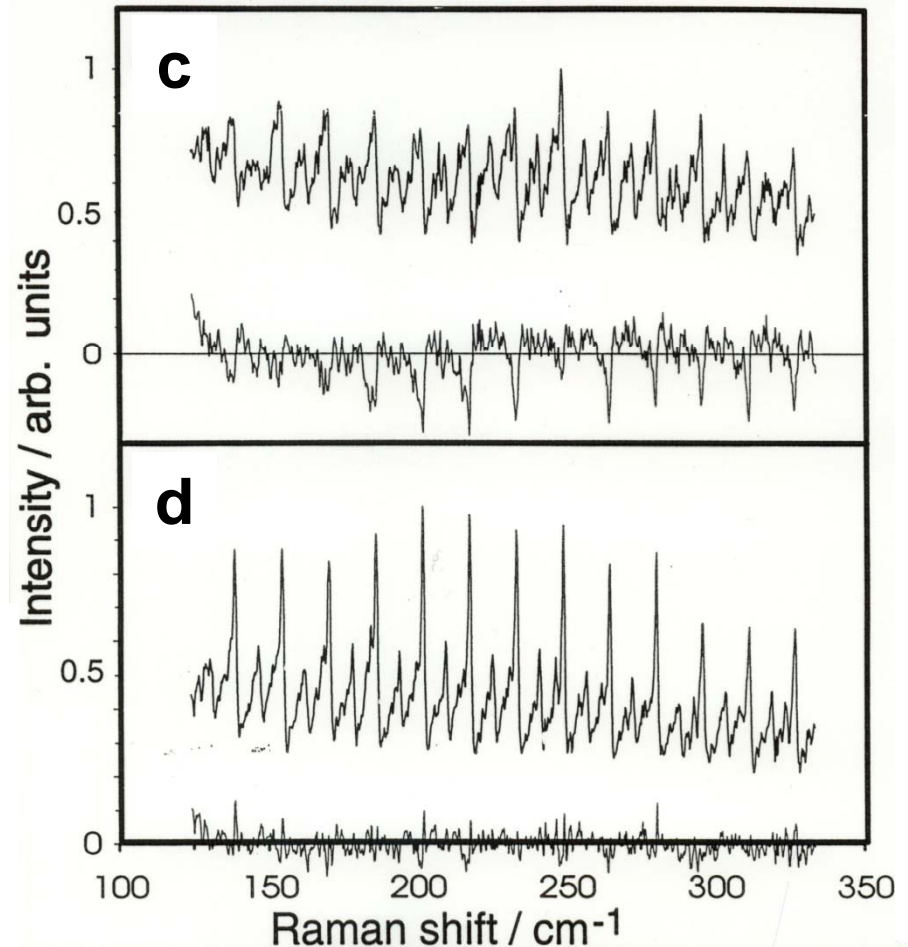


Figures are from Bengtsson et al., *Proc. Combust. Inst.* **25**,: 1735-1742 (1994)



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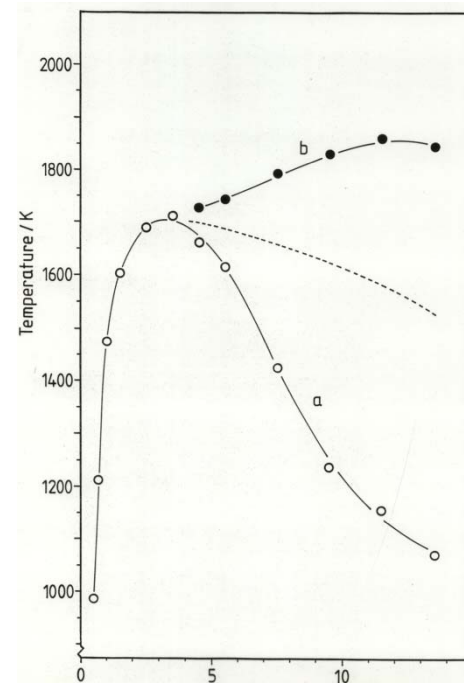
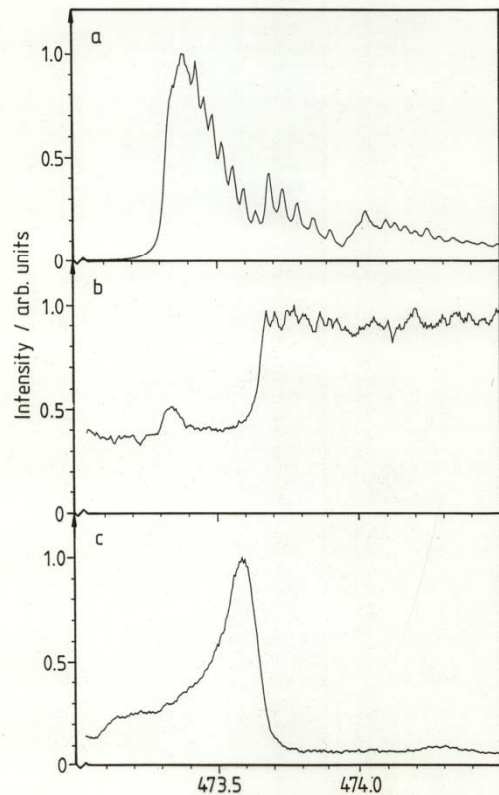
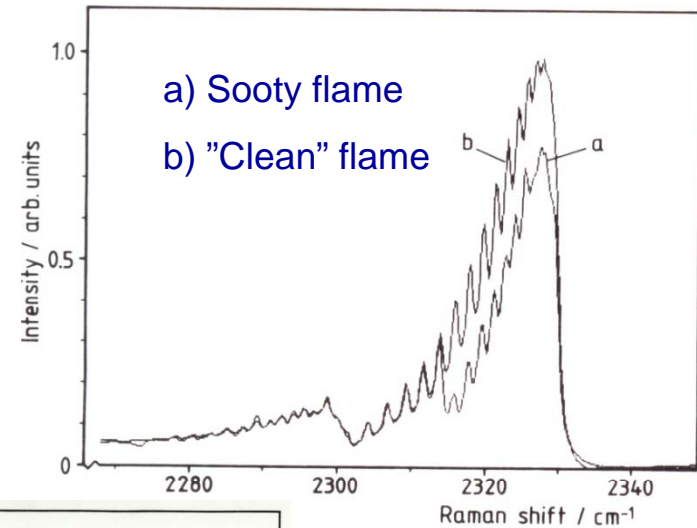
c: 32 CAD ATDC, 32 bar, 2550 K

d: 56 CAD ATDC, 15.5 bar, 2435 K

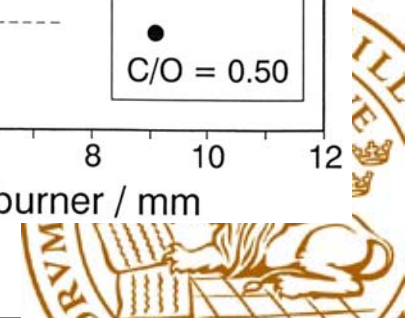
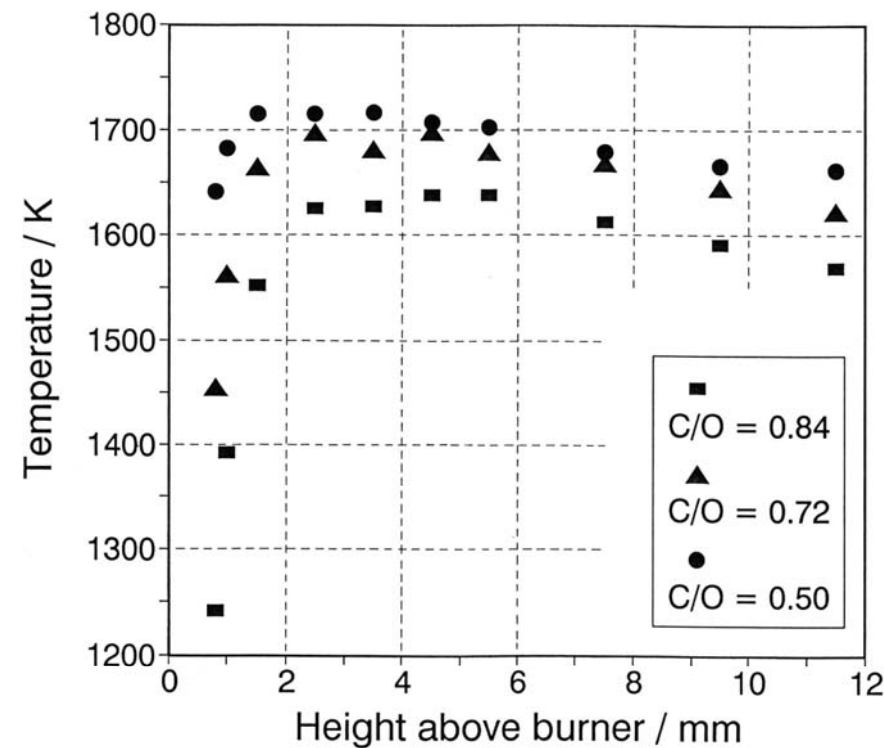
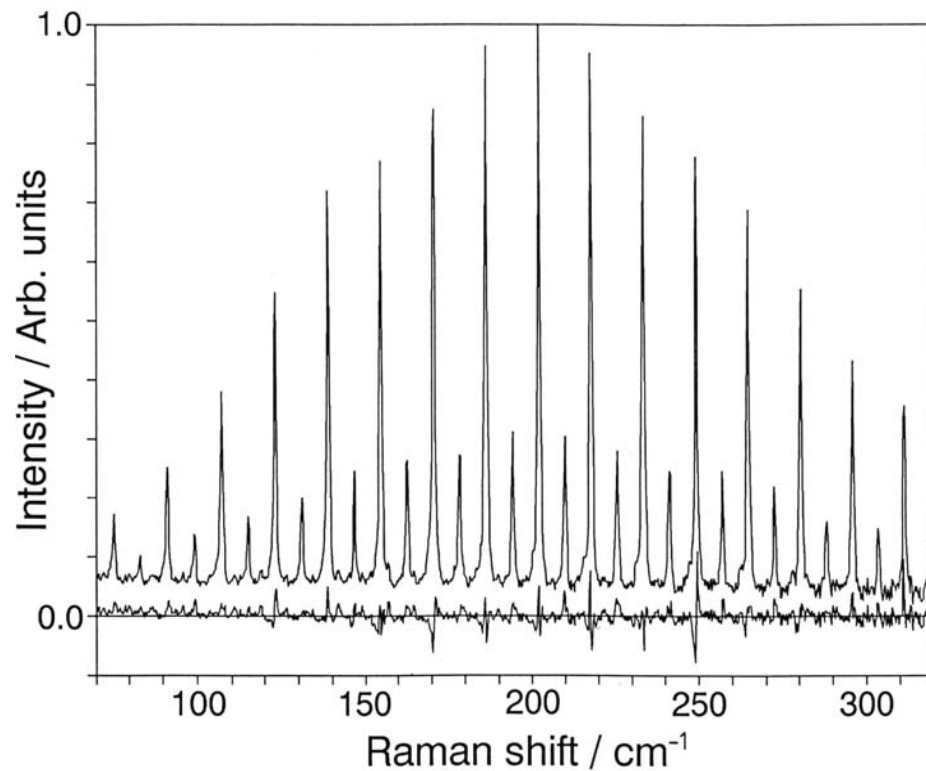
Figures are from Bengtsson et al., *Proc. Combust. Inst.* **25**,: 1735-1742 (1994)



Sooty flame diagnostics: Vib CARS



Sooty flame diagnostics: Rot. CARS



Summary: CARS

- strong coherent signal
- point measurement technique
- complex theory
- double-ended technique
- can be applied to harsh environments, since background radiation is easily discriminated, anti-Stokes signal (low fluorescence interferences)
- mainly for temperature measurements, where temperature is measured from spectral profile
- high accuracy for temperature measurements
- can sometimes be used for concentration measurements, but for major species only.



Summary:

Rotational CARS vs Vibrational CARS

- Same experimental complexity
- Both methods are very accurate thermometers:
Uncertainty 1-2% of T
- Rotational CARS: better at high P, low T
Vibrational CARS: better at low P, high T
- Rotational CARS: many species simultaneously
Vibrational CARS: often one species only with a setup
- Rotational CARS: Problems may arise with rejection of light at 532 nm because of spectral closeness.