

# **11. Non-linear optical techniques**

- **Introduction**
- **Degenerate four wave mixing, DFWM**
- **Polarization Spectroscopy, PS**
- **IR measurement (IRPS, IR-DWM)**
- **(Stimulated Emission, SE)**



# Diagnostic dilemma:

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- **LIF**
  - High sensitivity,
  - 2D imaging,
  - Spontaneous technique (sensitive to strong background)
- **CARS**
  - Coherent technique
  - One-point measurements
  - Low sensitivity

**Need: A coherent technique with high sensitivity and 2D imaging possibility**

**Candidates:DFWM, PS and SE (one point)**



# Nonlinear optics

Thus far, the induced polarization of molecules has been assumed to depend linearly on the applied electromagnetic field. This is however only valid for incident radiation of low intensity.

Generally, the induced polarization is a nonlinear function of the applied electromagnetic field:

$$\vec{P} = \vec{P}^{(1)} + \vec{P}^{(2)} + \vec{P}^{(3)} + \dots$$

$$\vec{P} = \epsilon_0 \left( \chi^{(1)} \vec{E} + \chi^{(2)} \vec{E}^2 + \chi^{(3)} \vec{E}^3 + \dots \right)$$

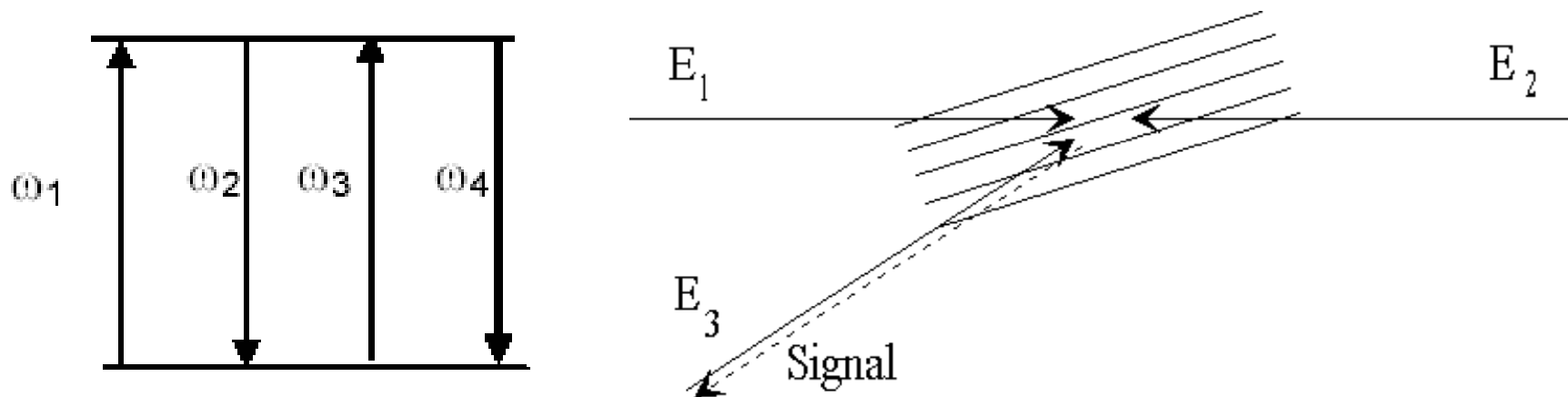
For gases, which are isotropic (inversion symmetry), the even order polarizations vanish

DFWM and PS are four-wave mixing processes based on the nonlinear response via the third-order susceptibility ( $\chi^{(3)}$ ), in the same way as CARS.



# DFWM

Can be described in terms of a grating phenomena

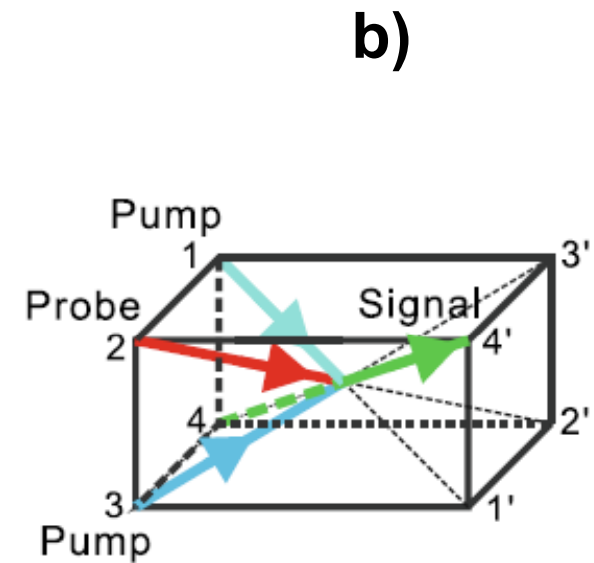
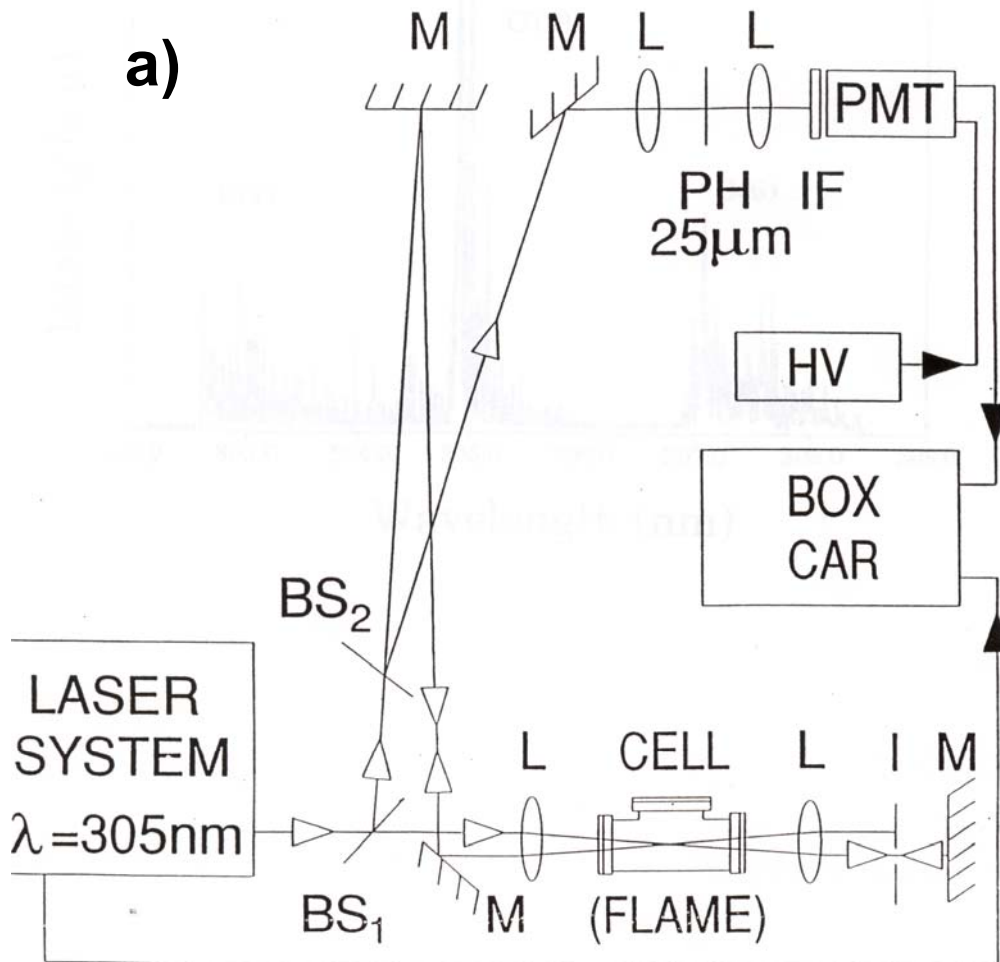


$\omega$ , laser frequency, tuned to an atomic/molecular resonance

$$I_{DFWM} \propto I_P^2 I_{Pr} N^2$$

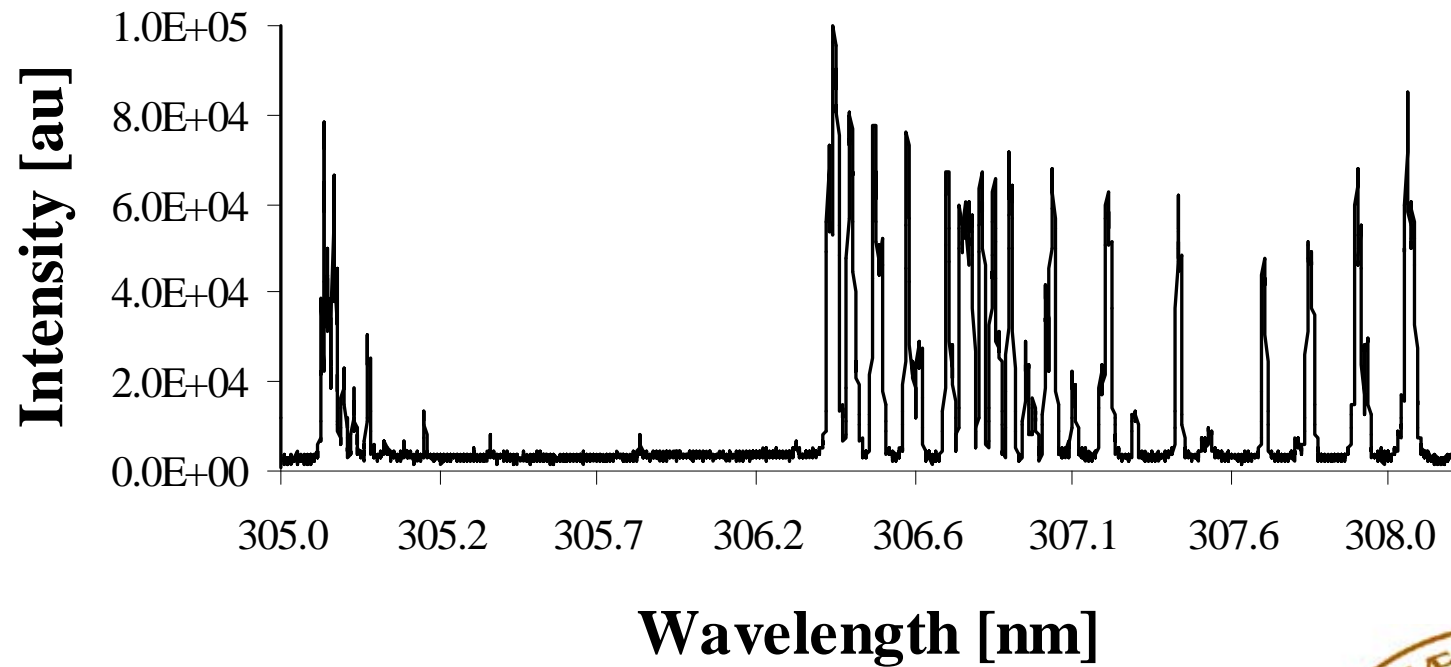


# Experimental set-up

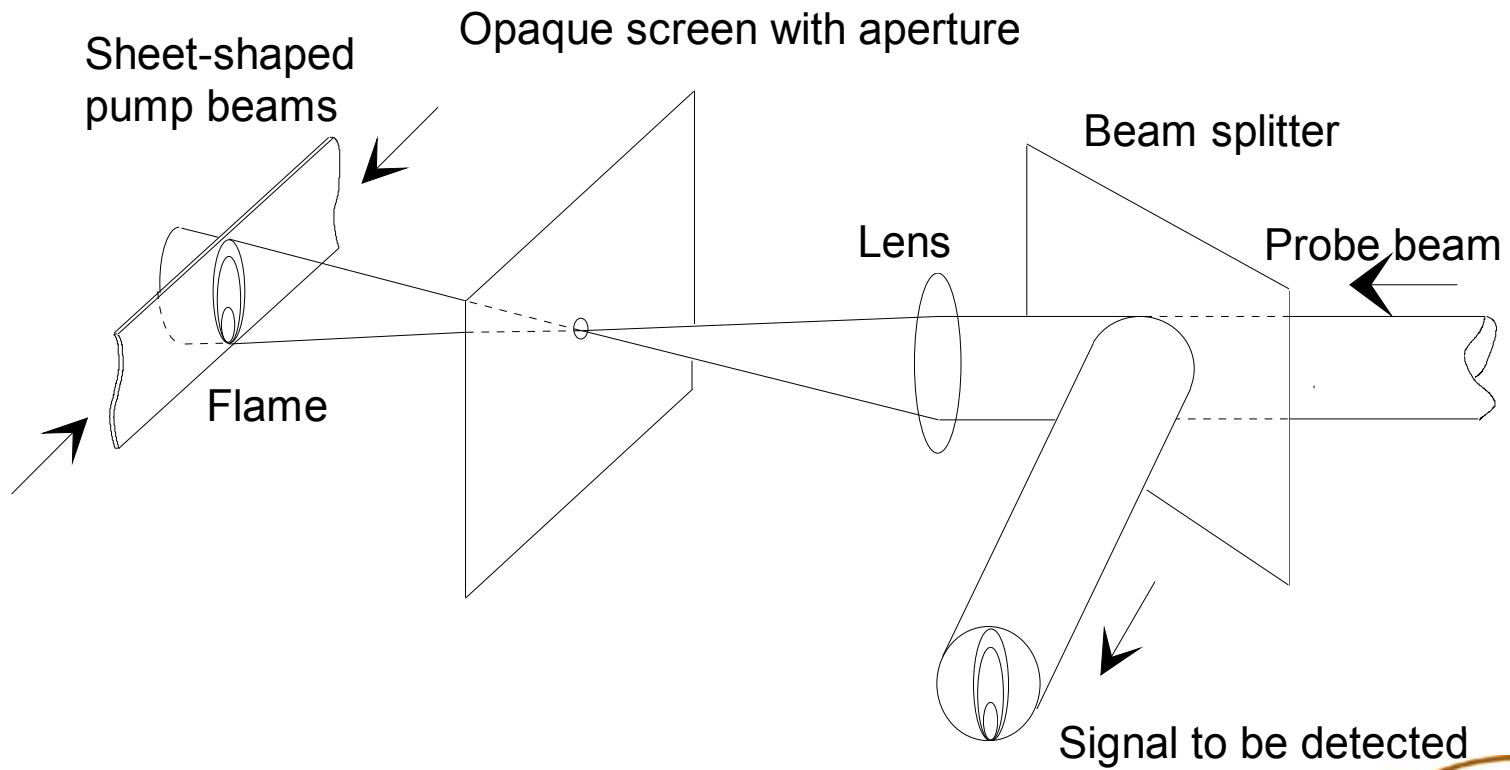


# Detection of ammonia and OH

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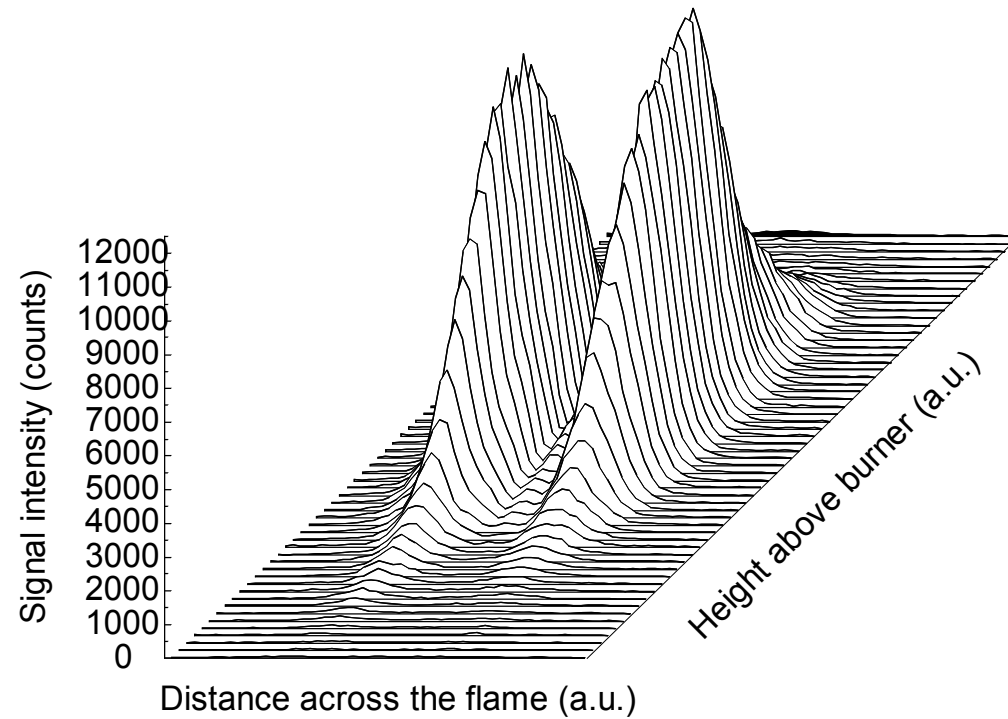


# 2D imaging



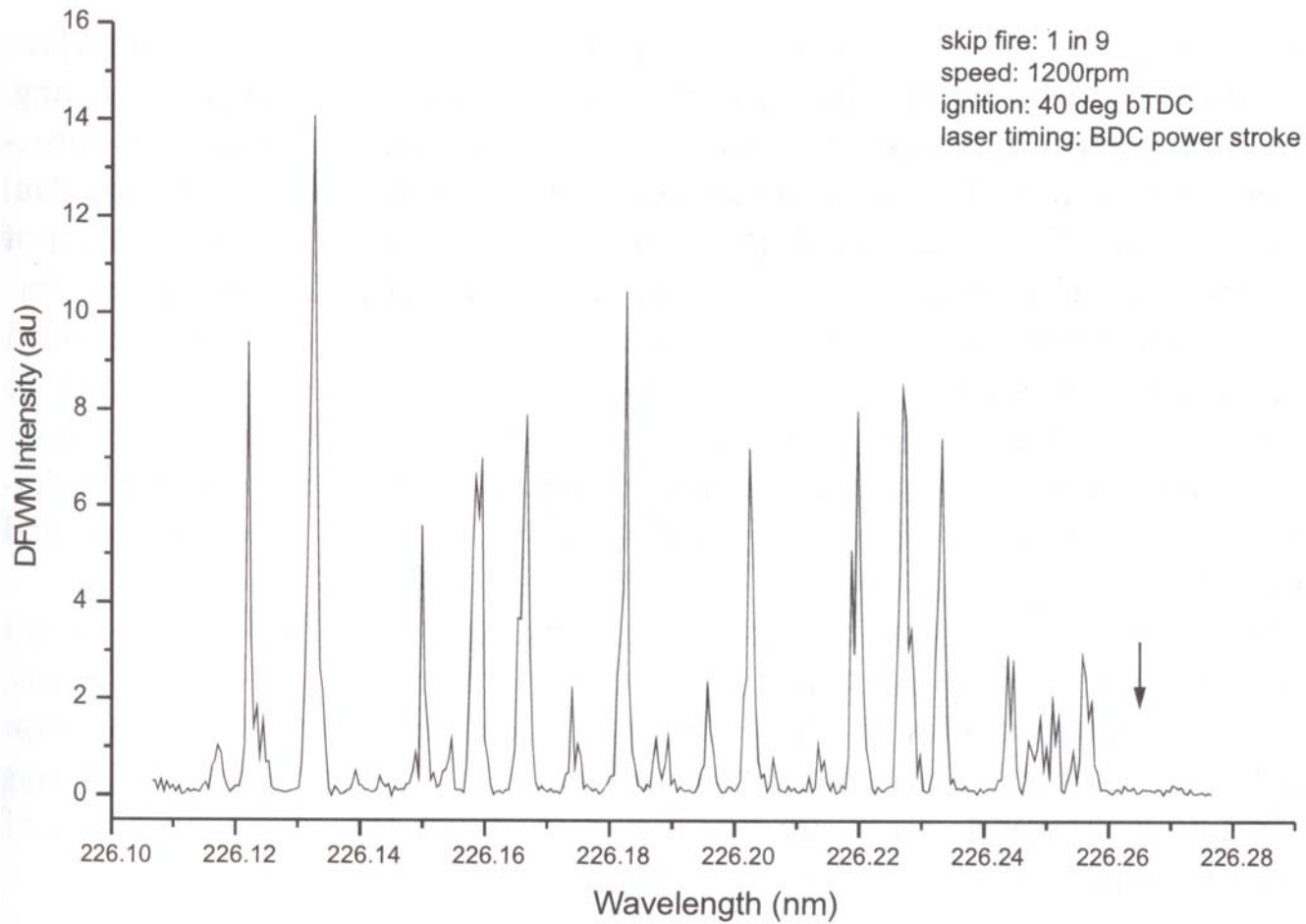
# 2D OH image using DFWM

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# DFWM application



# DFWM summary

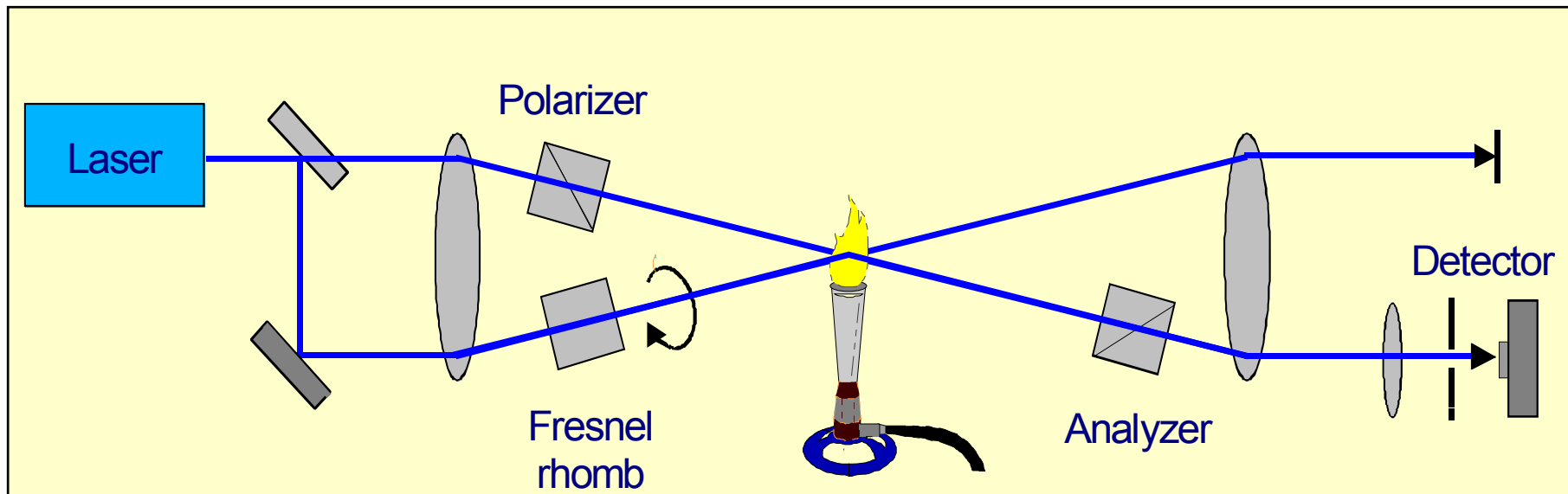
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- Coherent technique with high sensitivity (ppm)
- 2D imaging possible
- Complex theory
- Advanced procedures for laser beam alignment
- Problem with background scattering



# Polarization spectroscopy

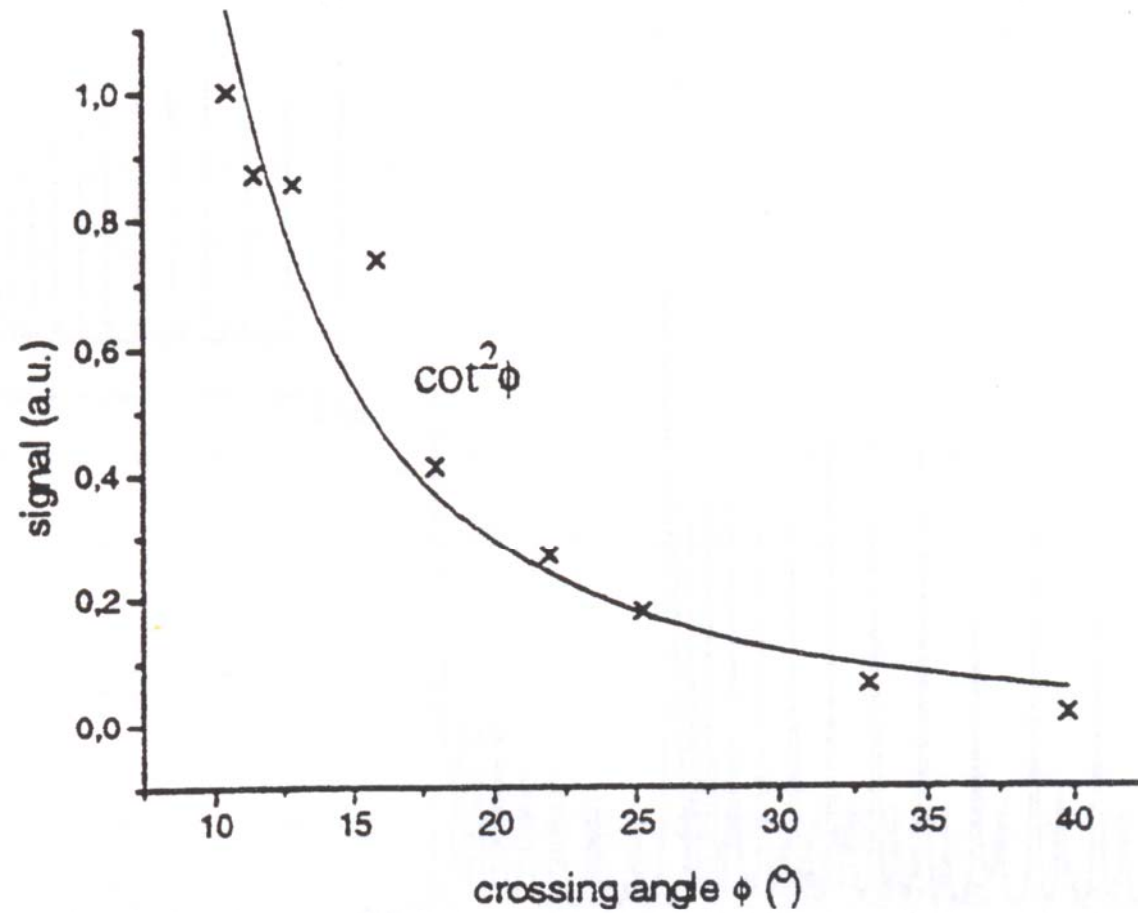
A pump beam induces an optical anisotropy (Birefringence and Dichroism), which is measured as a change in the polarization of the probe beam



$$I_t \propto (N^0 B_{J''J'} \zeta_{J''J'})^2 I_p^2 I_{pr}$$

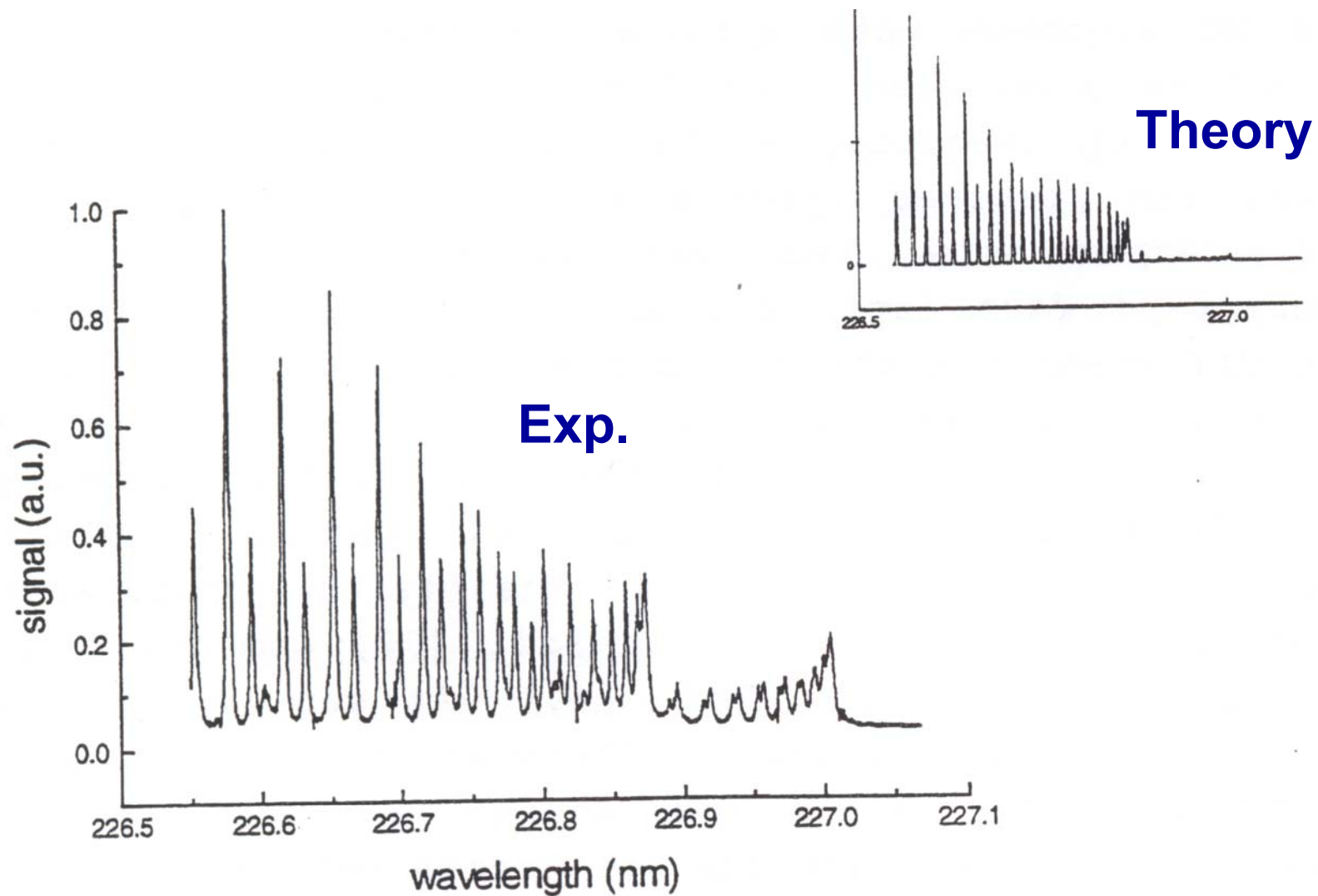


# PS signal dependence on crossing angle



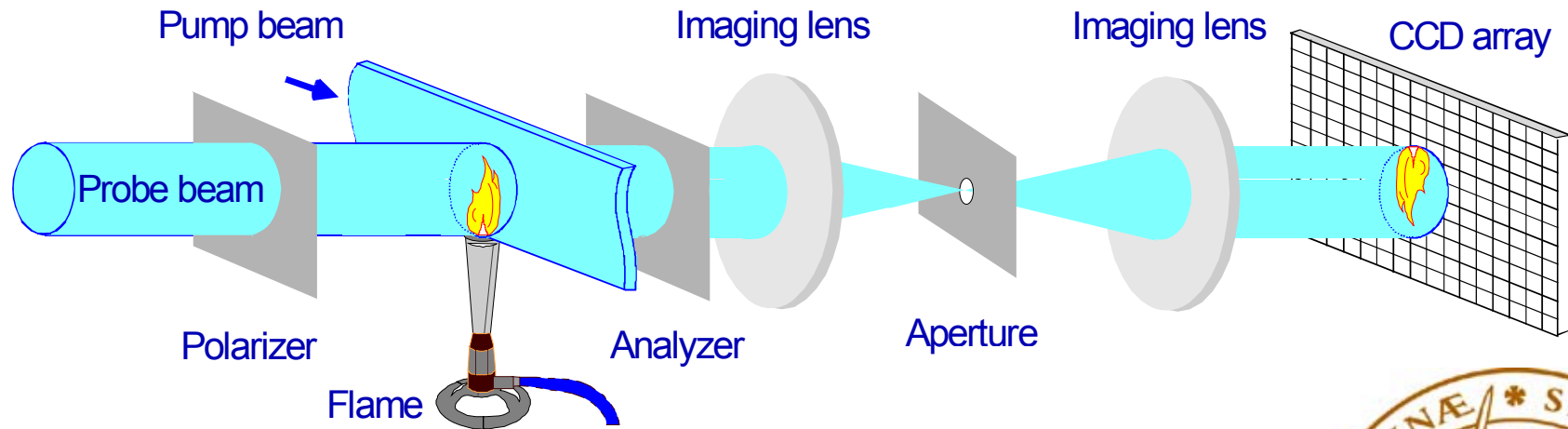
# Scanning experiments (NO)

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# Two dimensional imaging

- A strong linearly polarized pump beam formed to a narrow sheet of light crosses an unfocussed weaker probe beam in the flame
- The intersection volume imaged onto an image intensified CCD camera



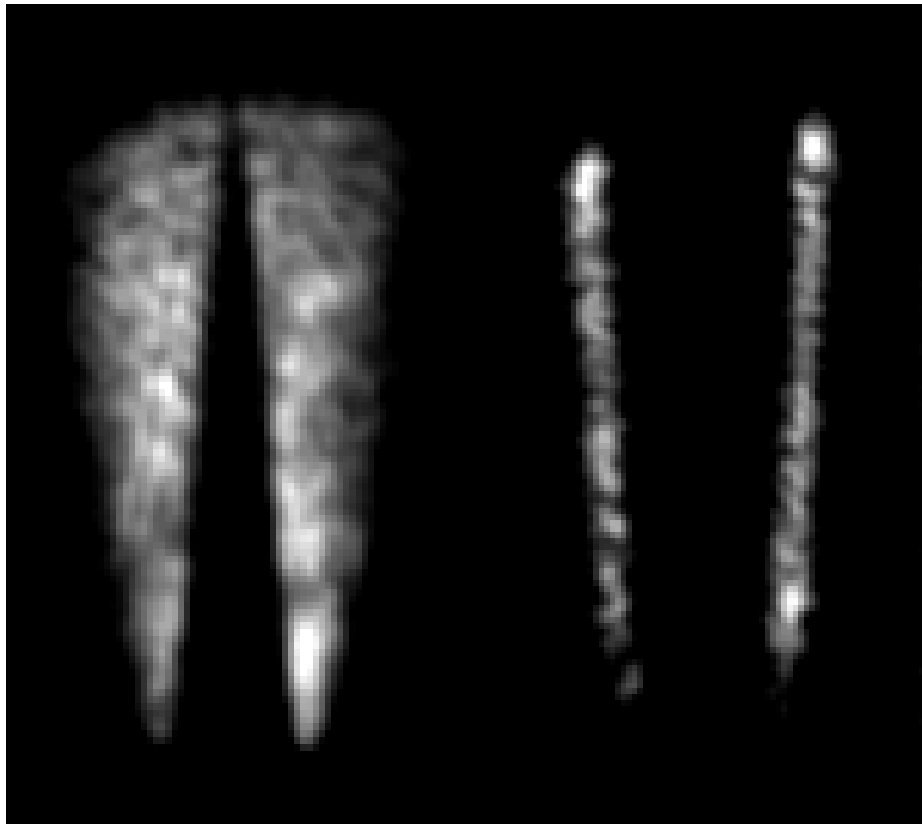
⇒ SIGNAL DISTRIBUTION IN ONE PLANE IN THE FLAME RECORDED



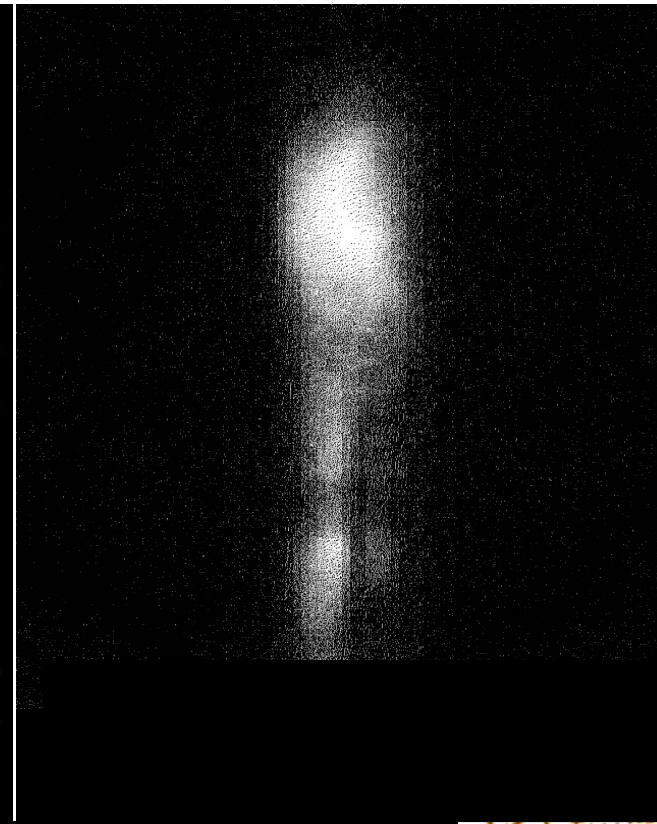
# Imaging of OH and NO in flames

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Images of OH signal  
distributions recorded in a  
 $\text{CH}_4/\text{O}_2$  flame



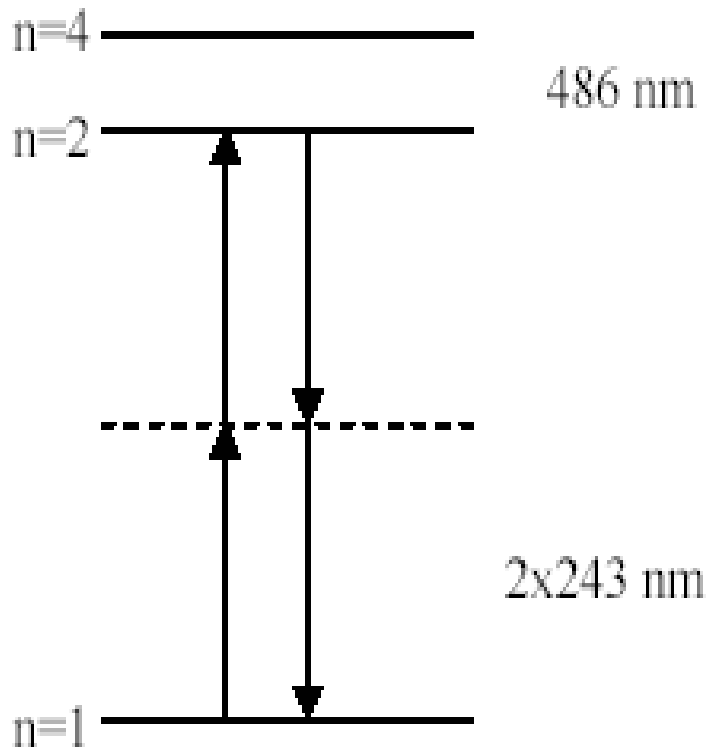
NO PS in a  
 $\text{H}_2/\text{N}_2\text{O}$  flame



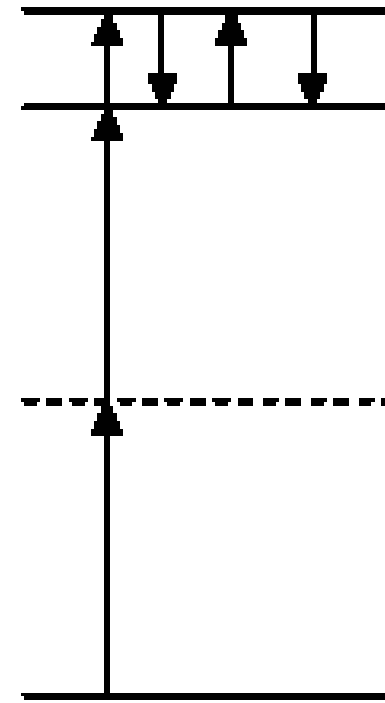
# Two-photon PS

exemplified by H atom detection

## Conventional approach



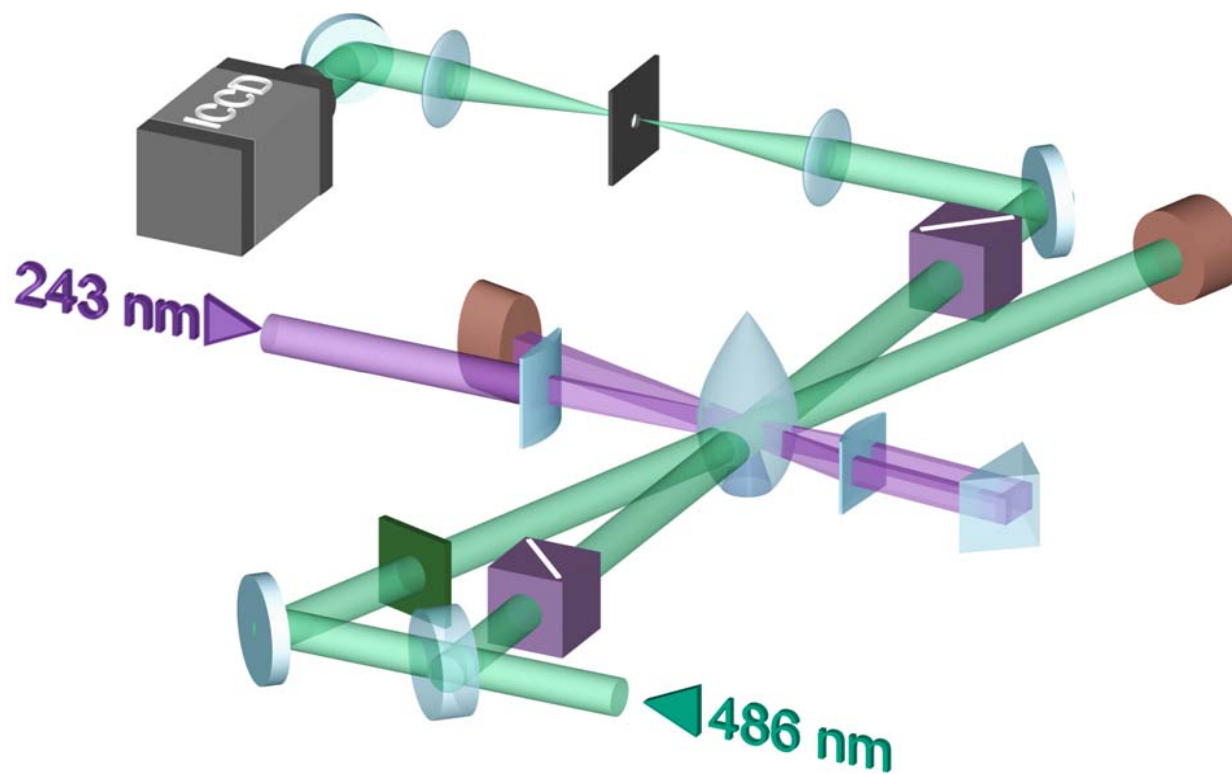
## New approach





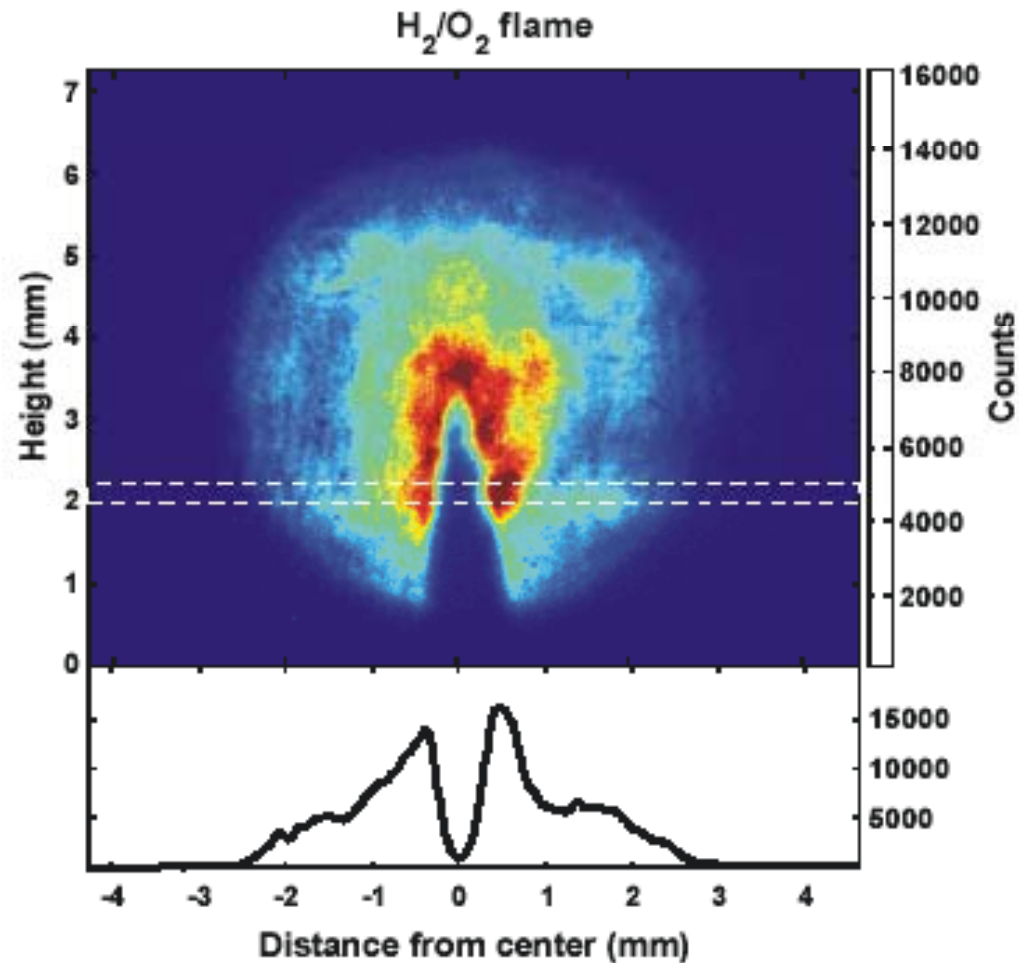
# Experimental set-up

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# Single shot 2D visualization of H

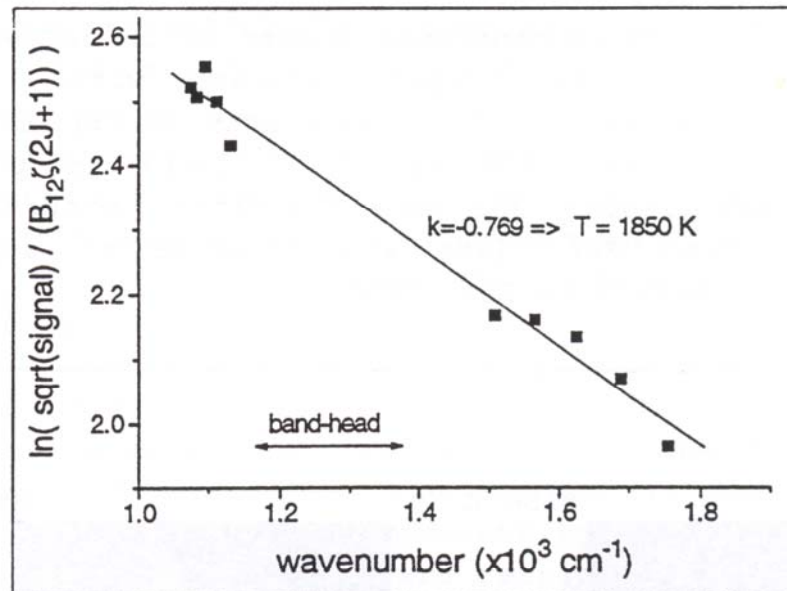
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# Temperature measurements

$$N^0 \propto (2J + 1) \exp(-E_r(J'')hc / kT) ,$$

$$I_t \propto (N^0 B_{J''J'} \zeta_{J''J'})^2 I_p^2 I_{pr} , \quad \ln \left( \frac{\sqrt{I_t}}{(2J + 1) B_{J''J'} \zeta_{J''J'}} \right) \propto \frac{E_r(J'')hc}{kT}$$



# Temperature measurements

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- Need for single shot, if possible, 2D T visualization
- Two-line excitation where T is given by;

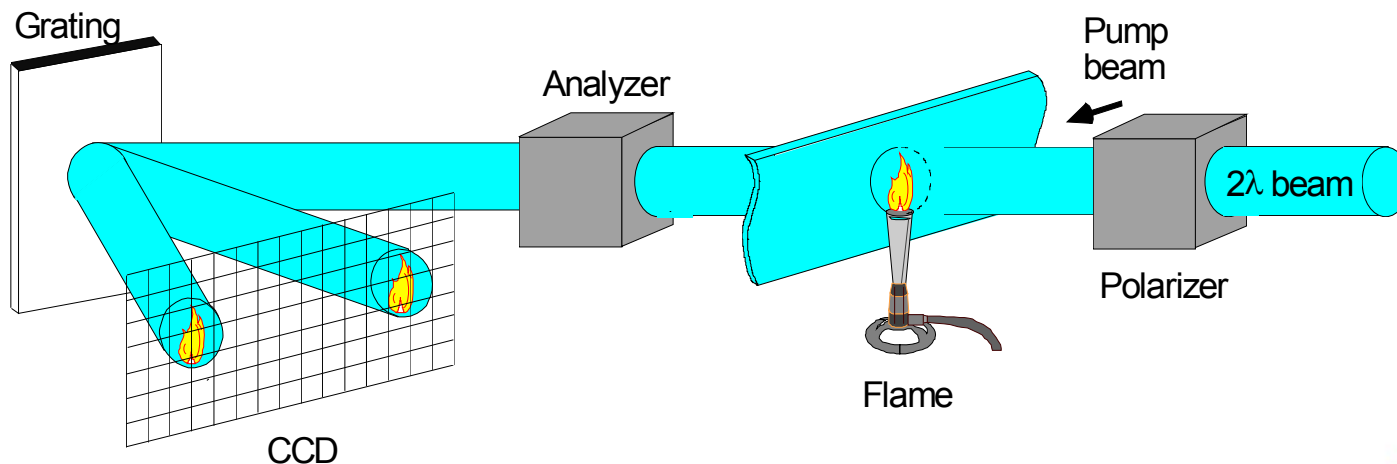
$$T(x, y) = \frac{2(E_1 - E_2) / k}{2 \ln[B_1 \zeta_1 (2J_1 + 1) / B_2 \zeta_2 (2J_2 + 1)] - \ln[I_1(x, y) / I_2(x, y)]}$$



# 2D temperature imaging

2D temperature maps can be extracted from signal distribution images, which are recorded with the laser wavelength tuned to resonance with two different rotational lines

Single pulse two-dimensional temperature visualization

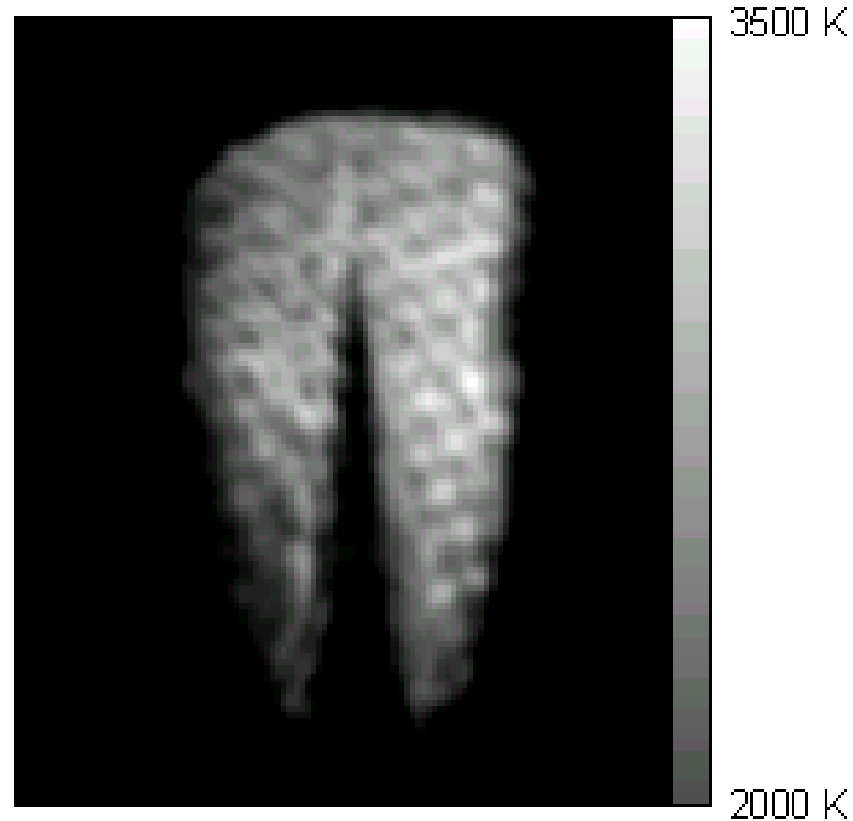


- dual-wavelength dye laser for excitation
- diffraction grating for spatial separation of the two images
- image-intensified CCD camera for image recording



# 2D temperature imaging

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**Challenges to achieve high single shot precision:**

- **Stable dye laser beamprofiles, or proper referencing**



# Polarization spectroscopy: Summary

- High sensitivity, (ppm)
- Good spatial resolution by crossed laser beams
- 2D imaging possibilities
- Two-photon (2D) experiments demonstrated
- Rather complex theory
- Possible problems with pressure induced birefringences (e.g. from windows in an engine)
- Sensitivity limited by extinction ratio of polarizers



# Why measurements in the IR spectral region?

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- In the UV/vis, only a limited number of species (OH, CH, NH, C<sub>2</sub>, NO, CH<sub>2</sub>O, ..) can be probed with resonant LIF, DFWM, PS.
- Many combustion important species CO<sub>2</sub>, CO, H<sub>2</sub>O, N<sub>2</sub>O, C<sub>2</sub>H<sub>2</sub>, CH<sub>4</sub> and other HC molecules or radicals, pose no accessible single-photon electronic transition in the UV/visible, but have strong absorption in the mid-infrared (2-5μm) via ro-vibrational transition.
- Spatially and temporally resolved measurements needed





# IRPS/IRDFWM

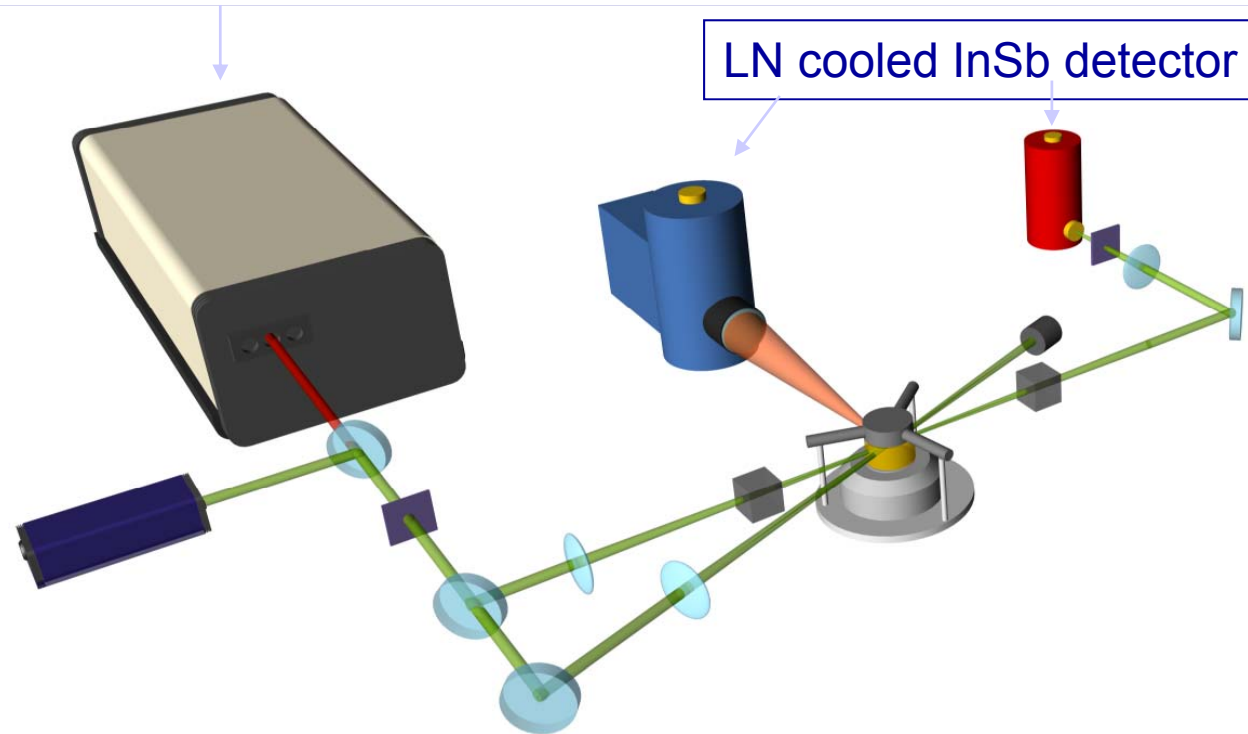
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- Challenges
  - Spectral interferences, especially in combustion environments where many species exist in a narrow spectral range
  - Doing non-linear experiments with invisible laserbeams
- Opportunity
  - Probing sensitively many important combustion intermediate species which otherwise are inaccessible with non-intrusive spatially resolved methods

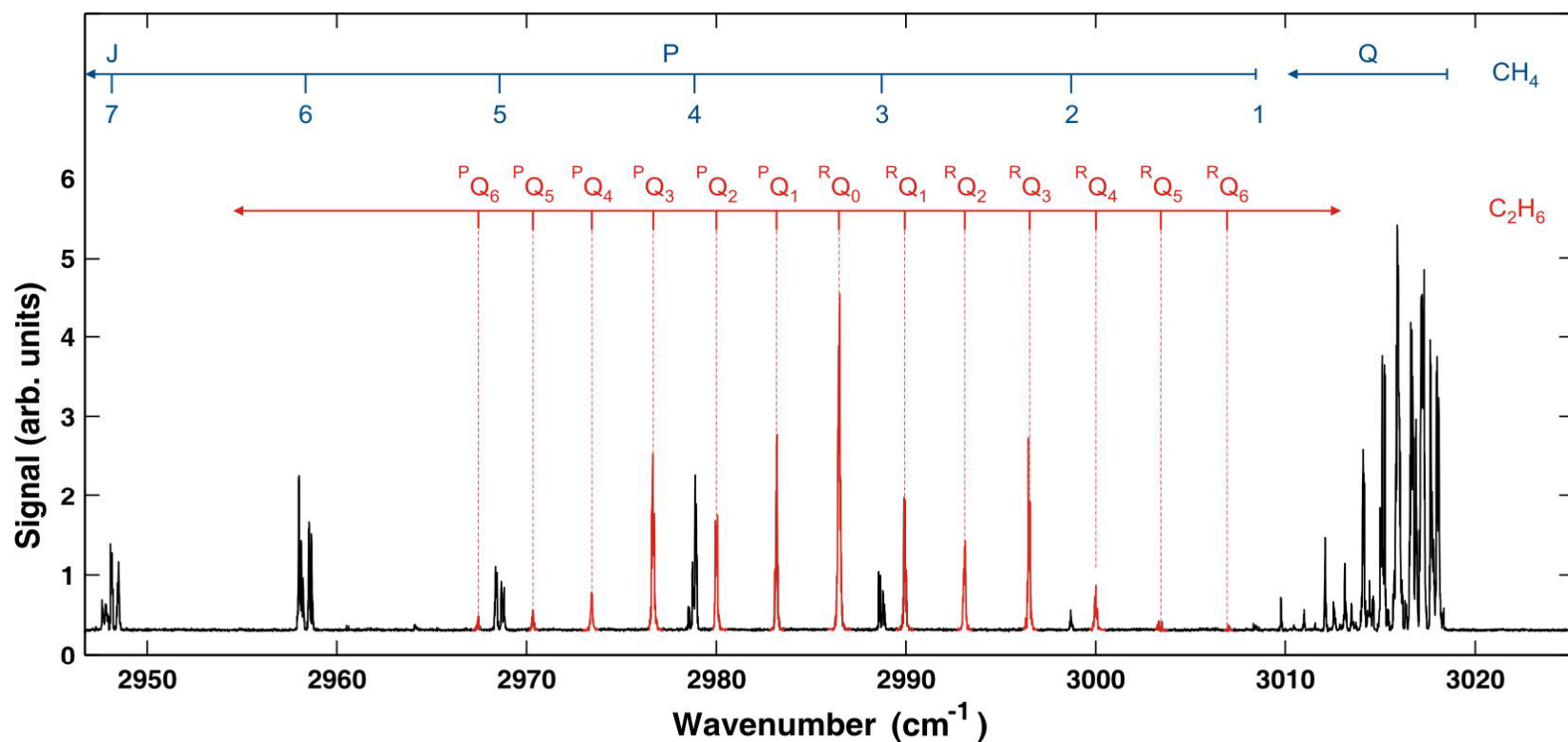


# Typical IRPS experimental setup

Laser system: DFM in  $\text{LiNbO}_3$  crystal, 1~3 mJ at 2-4  $\mu\text{m}$ , 0.03  $\text{cm}^{-1}$



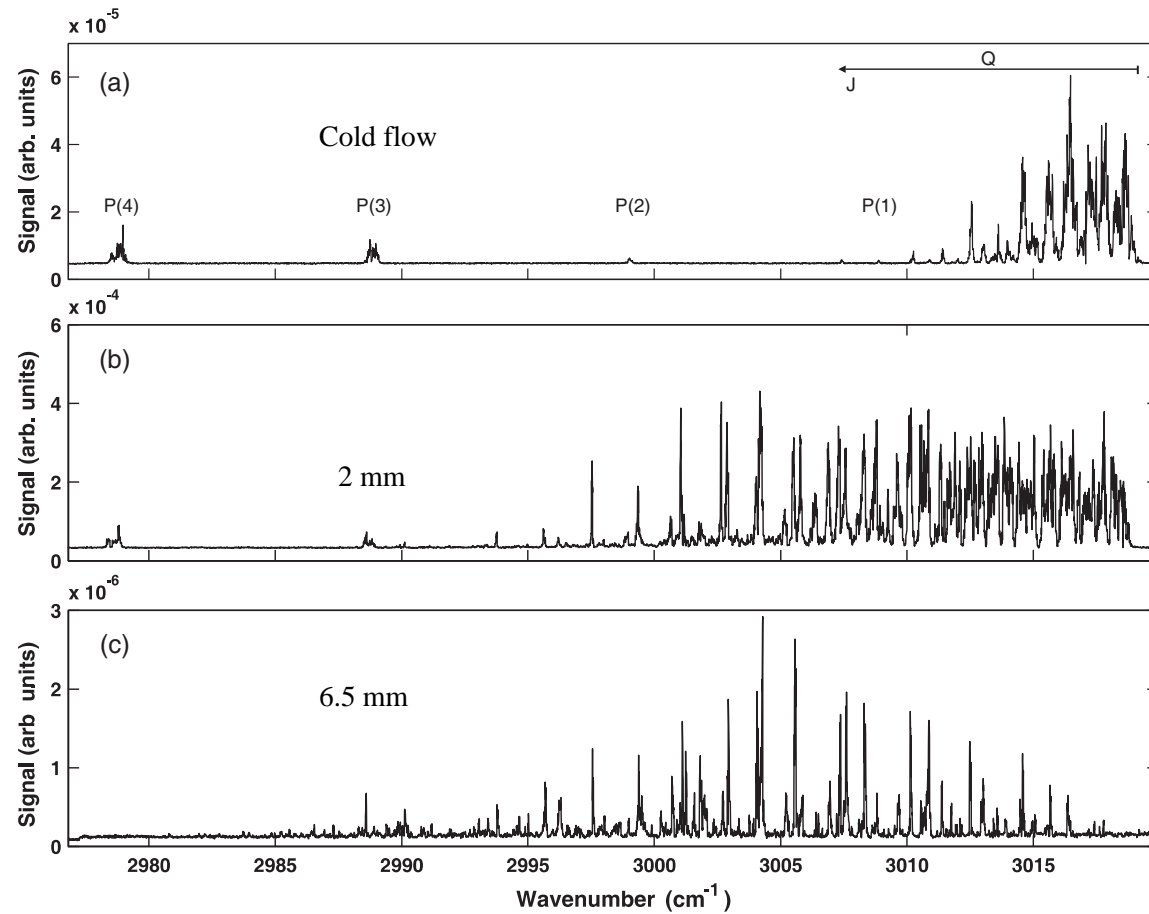
# IRPS spectra of CH<sub>4</sub> and C<sub>2</sub>H<sub>6</sub>



1.93% of CH<sub>4</sub> and 0.57% of C<sub>2</sub>H<sub>6</sub> mixed with Ar at 1 atm pressure.

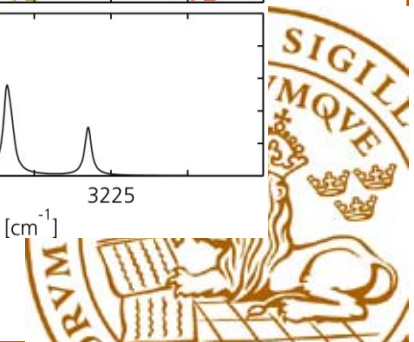
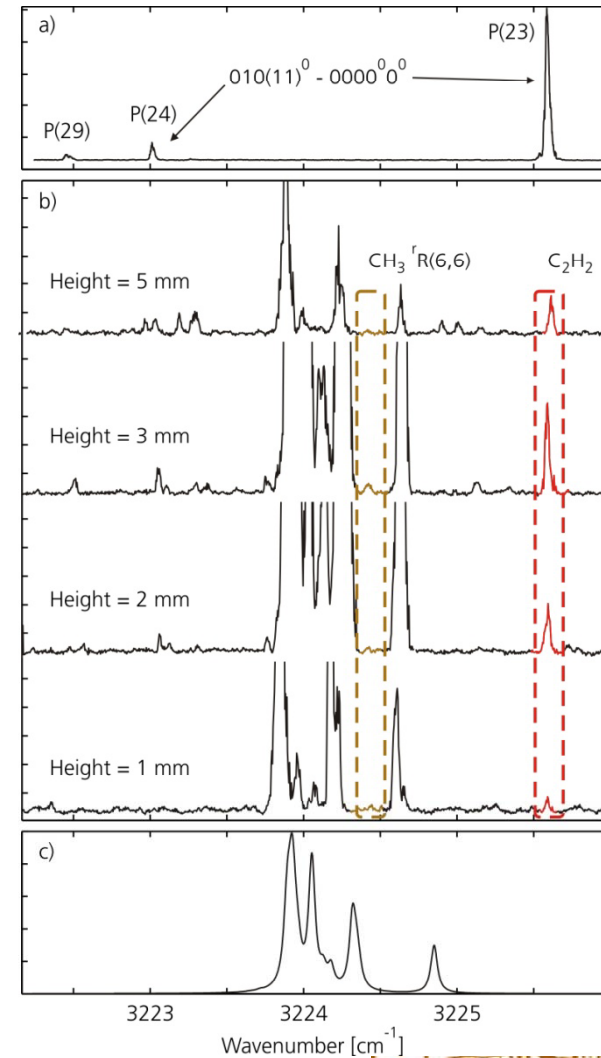


# Methane flame detection

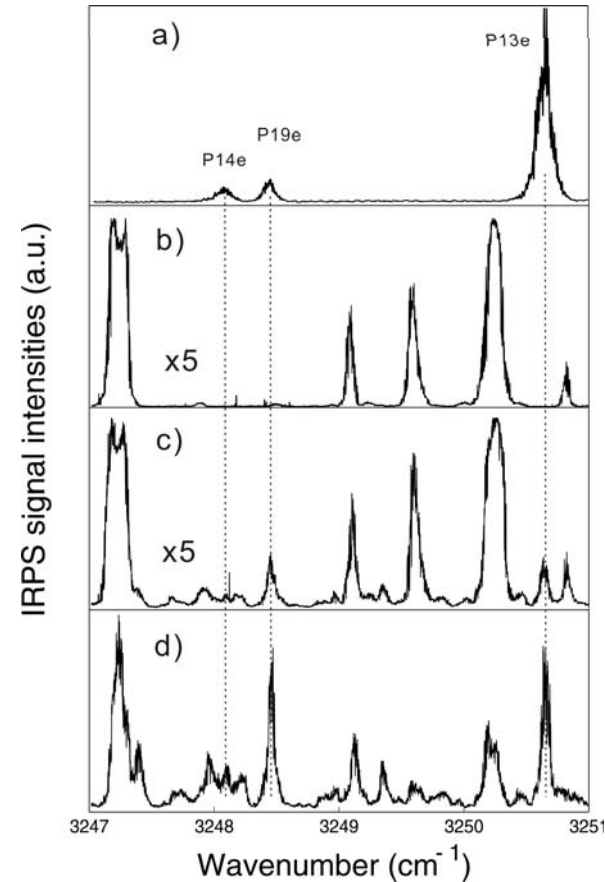
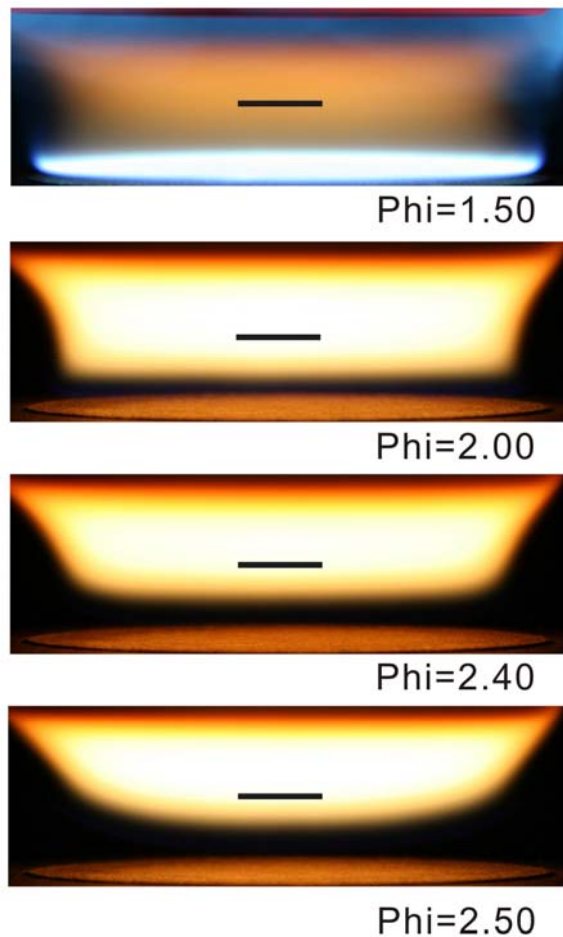


# Detection of acetylene and methyl with IRPS in a $\text{CH}_4/\text{O}_2$ (50 mbar, $\Phi=1.5$ ) flame

- Excitation scan of P(24) and P(23)  $\text{C}_2\text{H}_2$  lines in a gas flow
- Calculated IRPS spectra of hot methane
- Excitation scan above the burner
  - at 1 mm above the burner
  - at 2 mm above the burner
  - at 3 mm above the burner
  - at 5 mm above the burner
- Acetylene,  $\text{C}_2\text{H}_2$ , detected
- Methyl,  $\text{CH}_3$ , detected



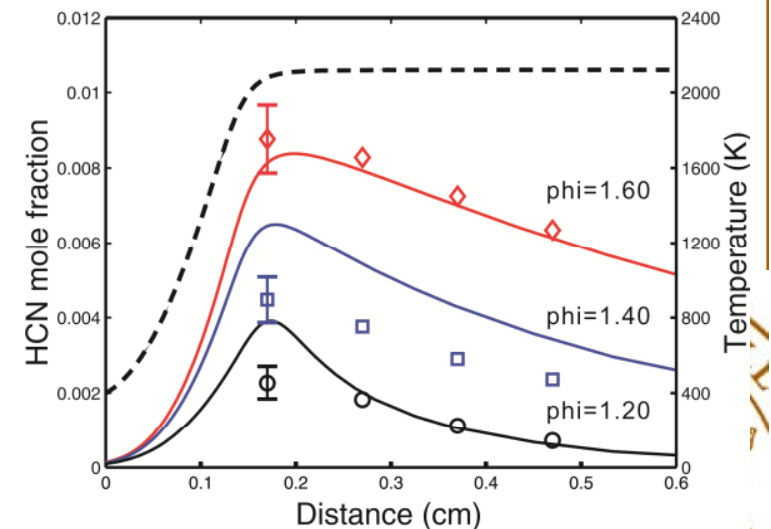
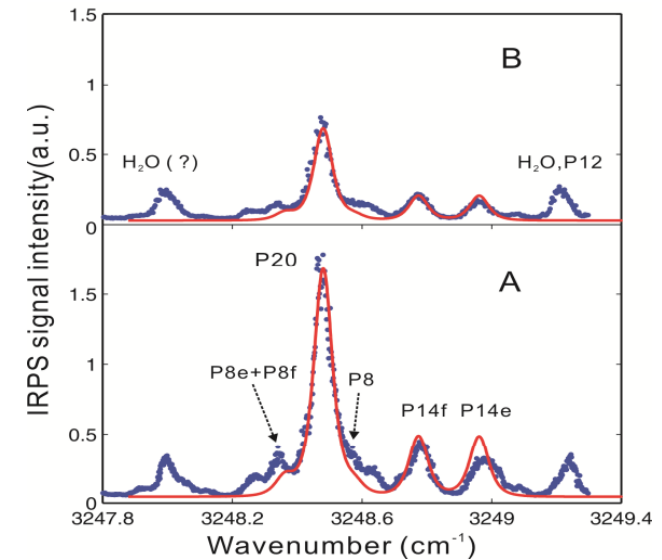
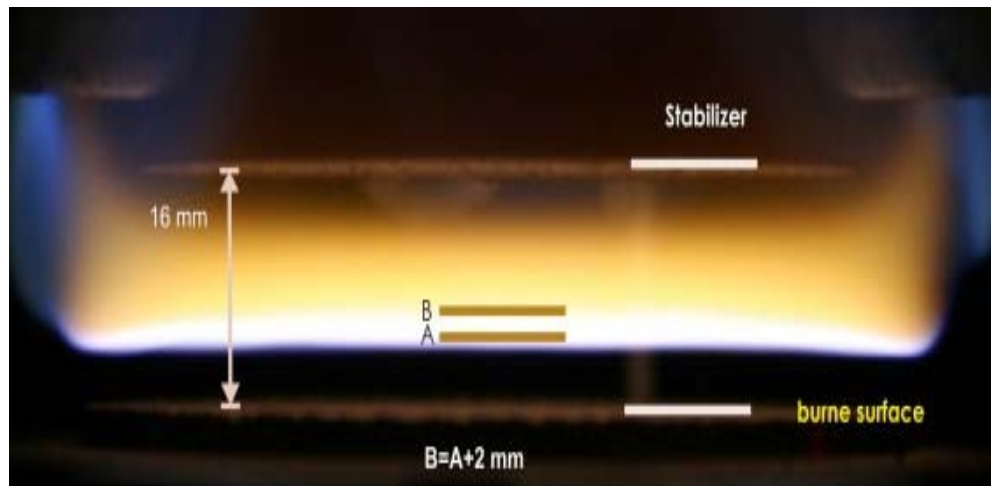
# IRPS detection of $C_2H_2$ in sooty flames



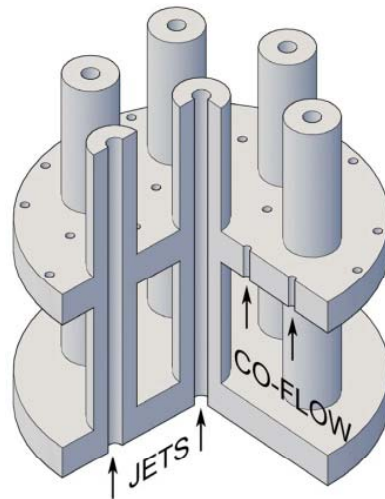
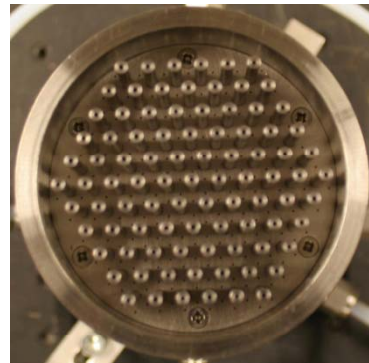
IRPS excitation spectra in calibration gas: a) and flames: b)  $\Phi = 1.00$ , c)  $\Phi = 1.50$  and d)  $\Phi = 2.50$ .



# HCN measurements in flames using IRPS



# IRPS measurements: HCN release history of solid fuel combustion/gasification

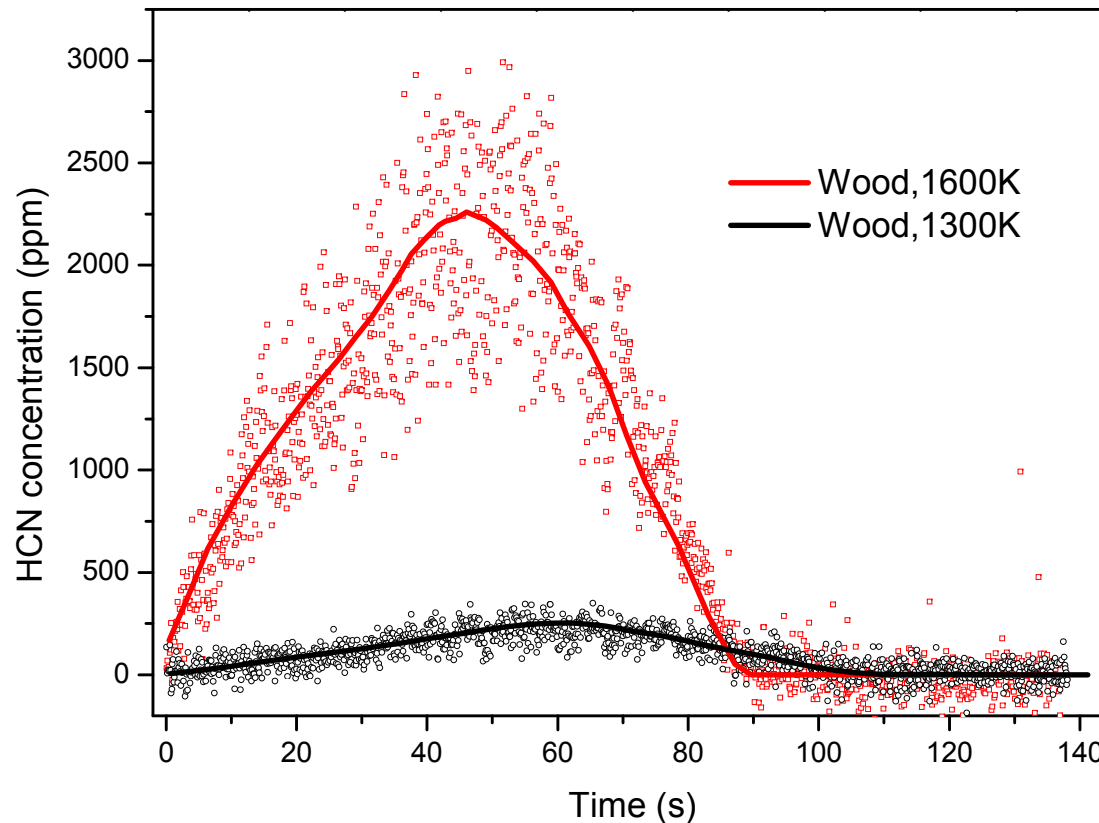


Sun et al. 2011





# IRPS measurements: HCN release history of solid fuel combustion/gasification



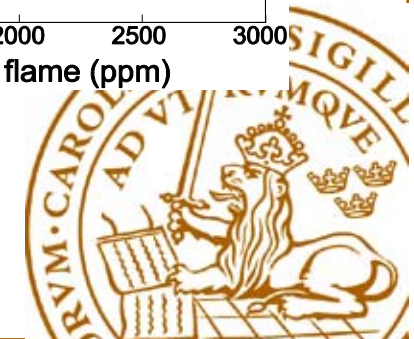
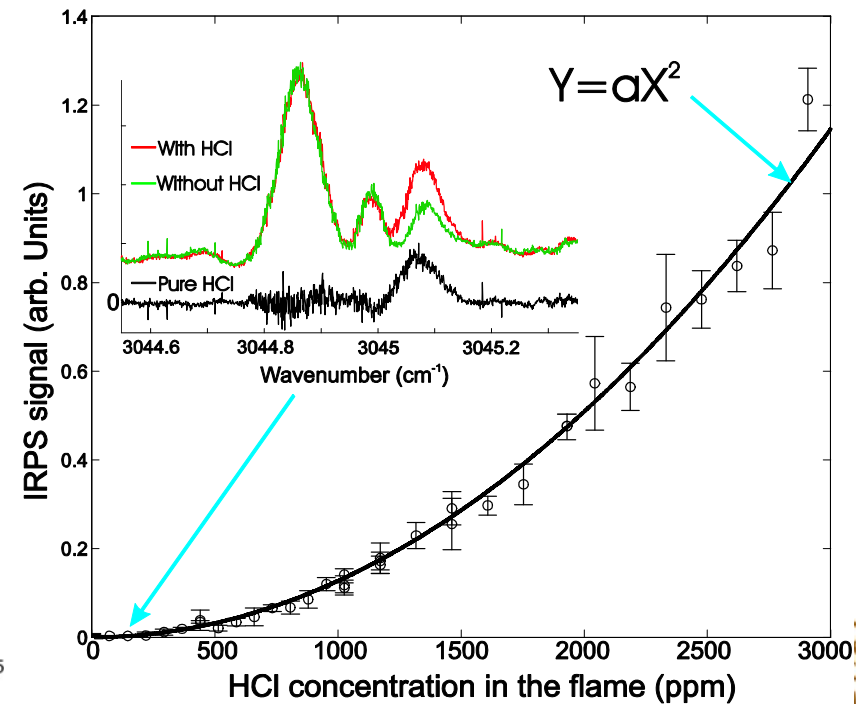
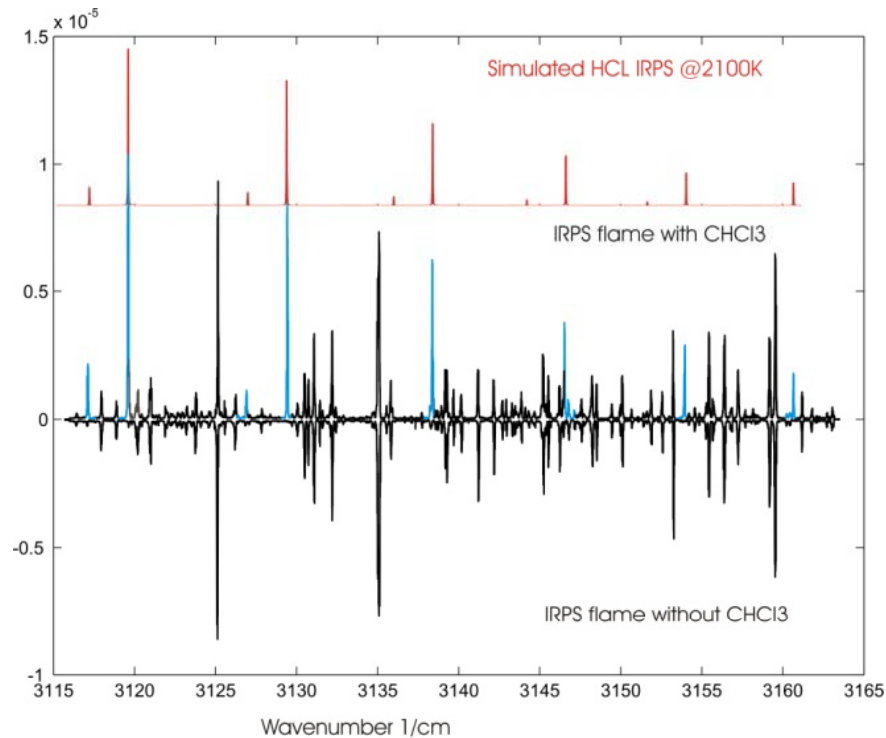
Comparison of HCN release at different temperatures for wood gasification.

Sun et al. 2011

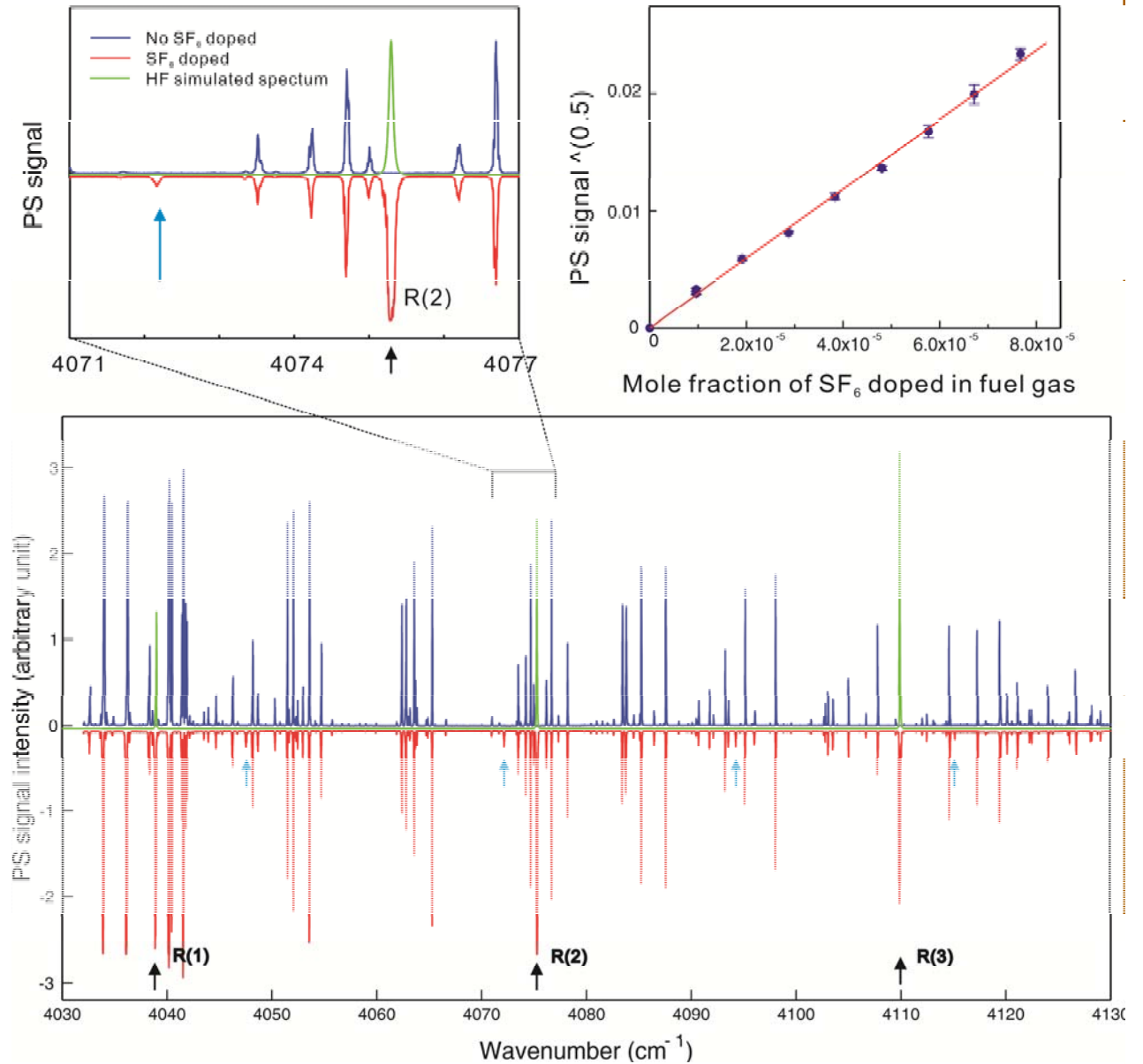


# HCl measurements using IRPS

CH<sub>4</sub> /O<sub>2</sub> flame seeded with chloroform



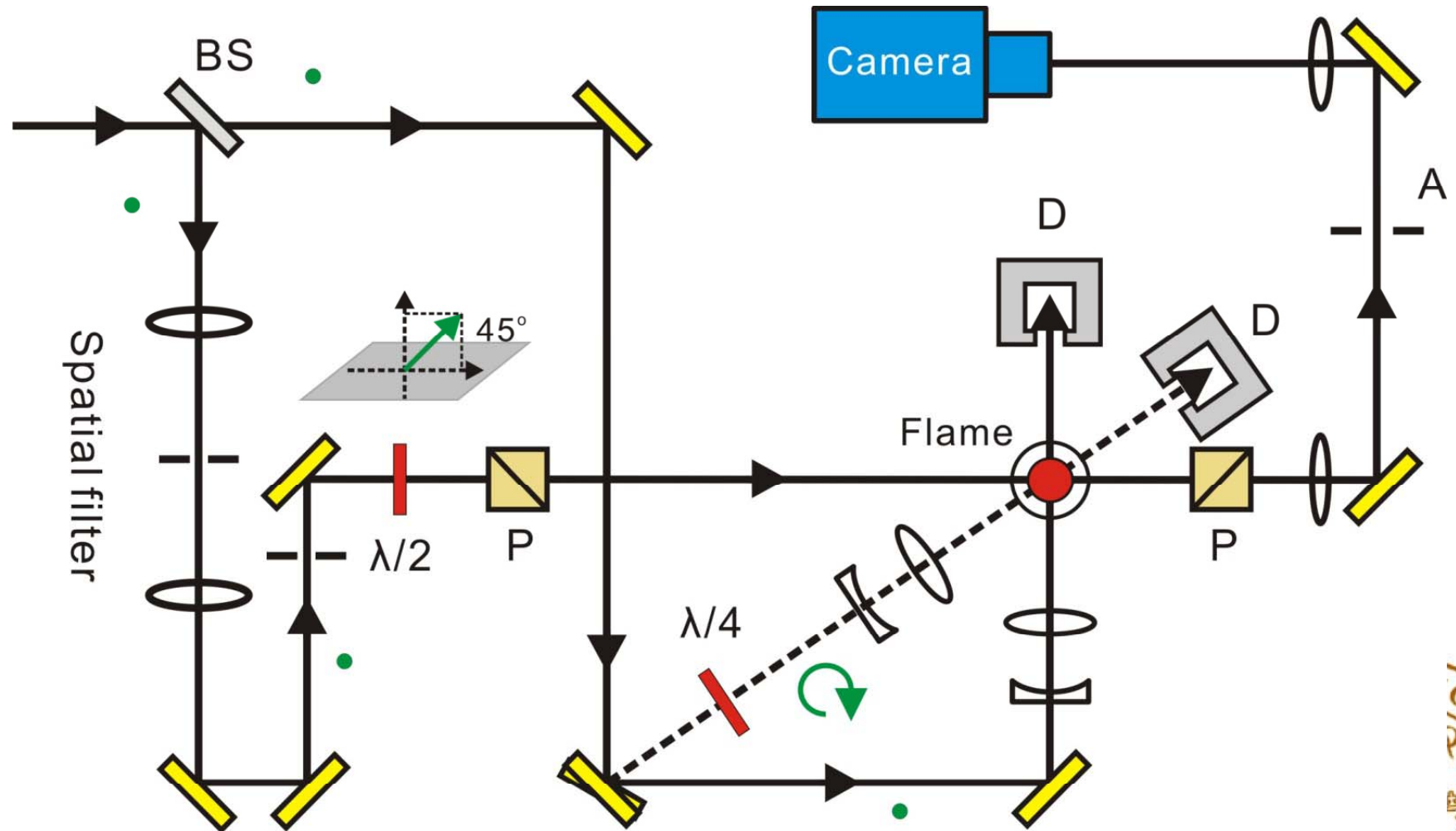
# 2D-IRPS measurements of HF in flames



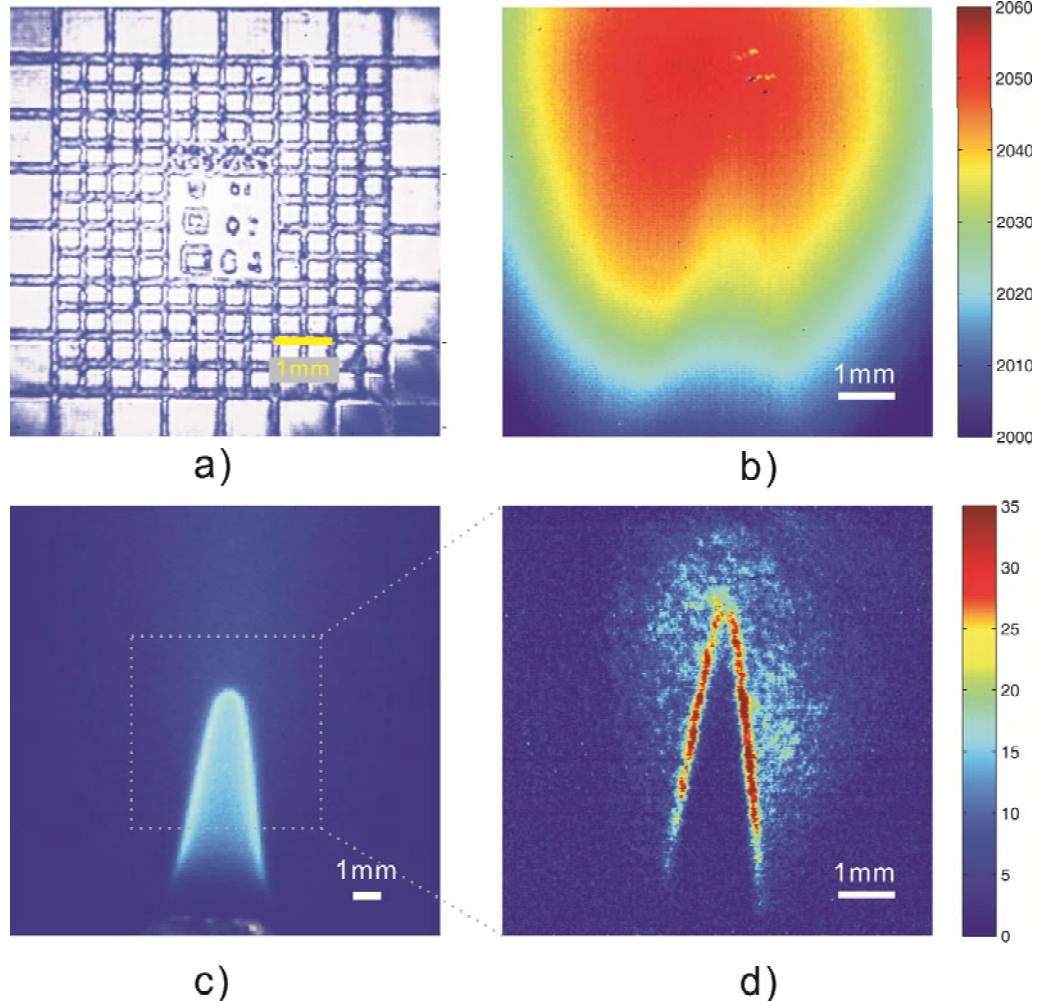
Spectra recorded in SF<sub>6</sub> doped CH<sub>4</sub>/air flames, E.r.= 1.1, McKenna type burner.  
The blue arrows show HF hot lines.

Sun et al. : In preparation

# Experimental set-up: 2D-IRPS



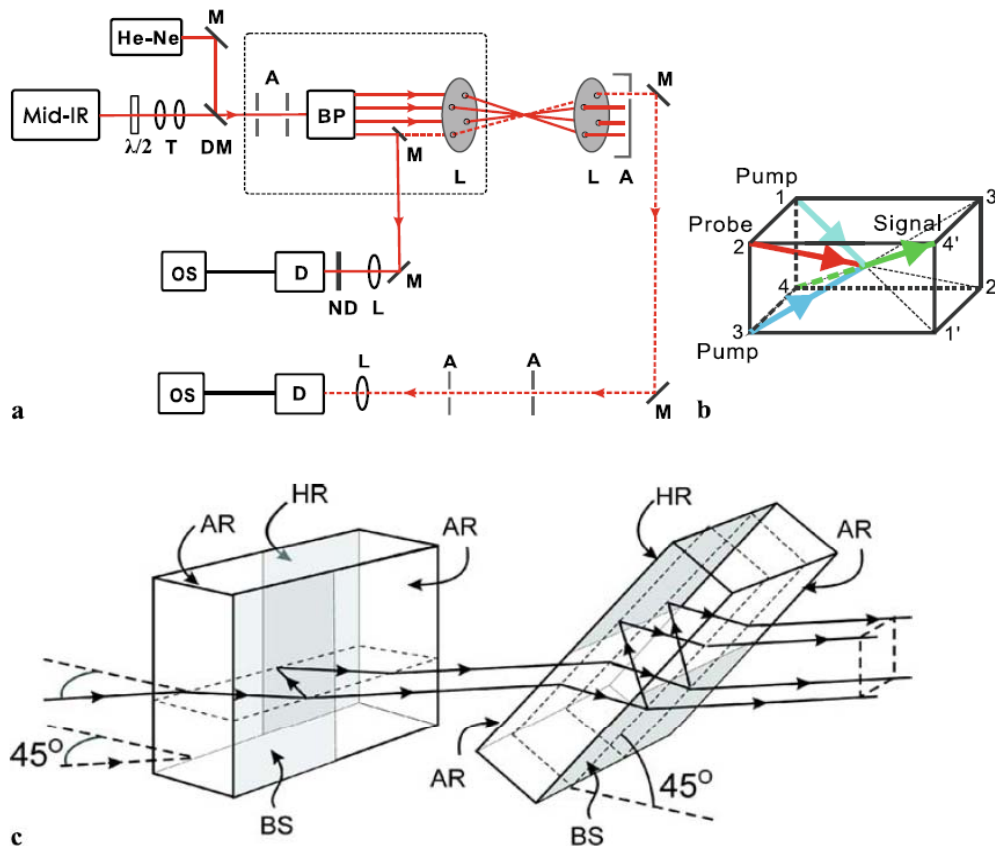
# 2D imaging of HF



- (a) Investigations of the spatial resolution of the imaging system
- (b) Thermal radiation from the flame without spectral filter
- (c) Photograph of the welding torch flame burning with  $\text{CH}_4/\text{O}_2$  ( $\Phi=2$ ) doped with 2%  $\text{SF}_6$
- (d) 2-D IRPS imaging of HF. The wavelength of laser focused on R(3) line of HF



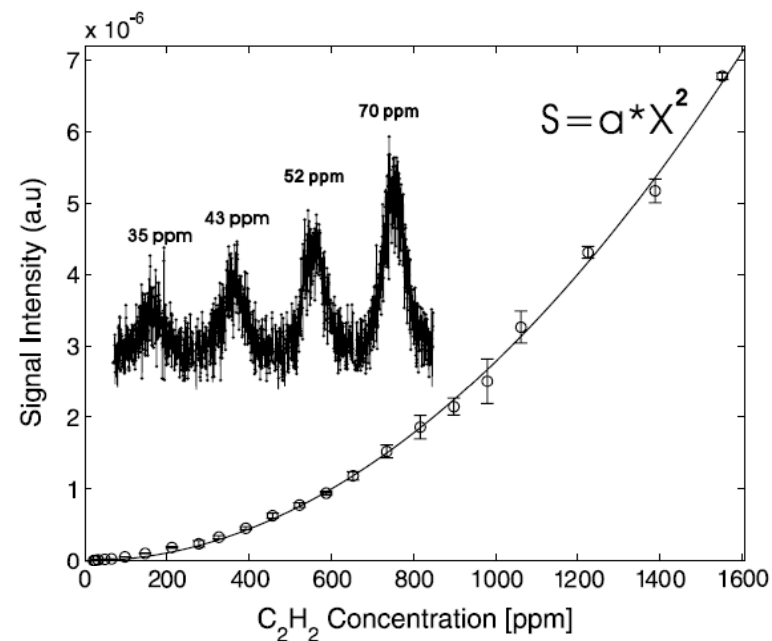
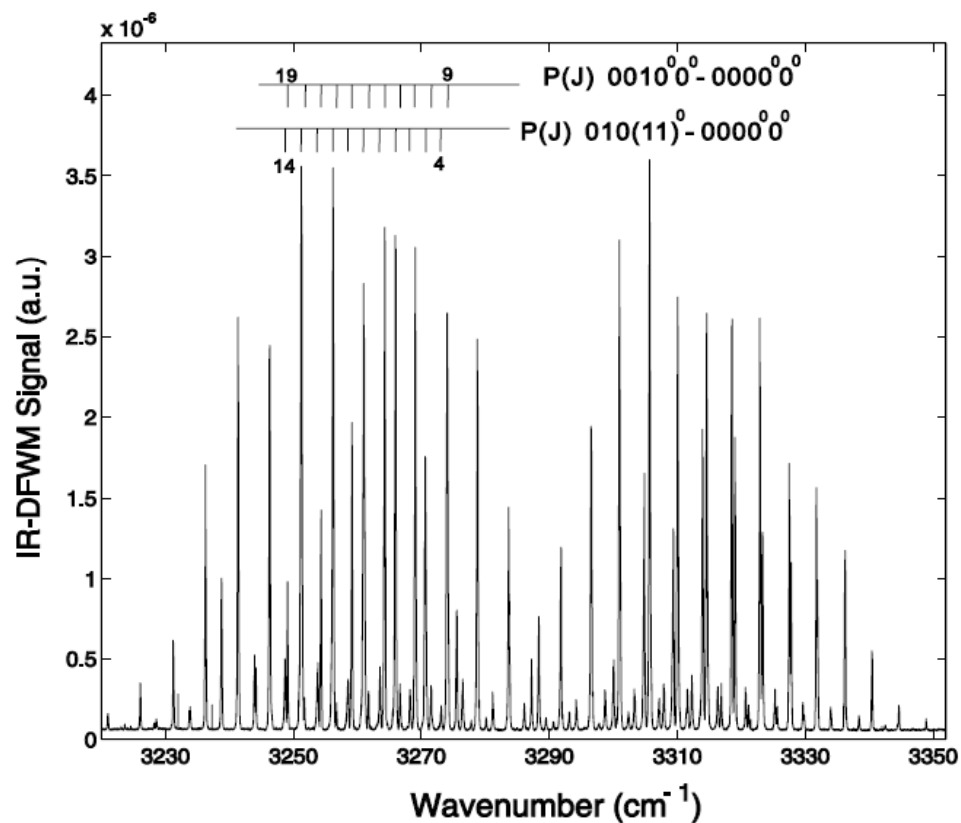
# New experimental scheme for IR DFWM experiments



Z.W. Sun, Z. S. Li, B. Li, M. Aldén and P. Ewart, 'Detection of C<sub>2</sub>H<sub>2</sub> and HCl with mid-infrared degenerate four-wave mixing with stable beam alignment: towards practical in situ sensing of trace molecular species', *Appl. Phys. B* 98, 593-600 (2010)



# IR-DFWM experiments on C<sub>2</sub>H<sub>2</sub>

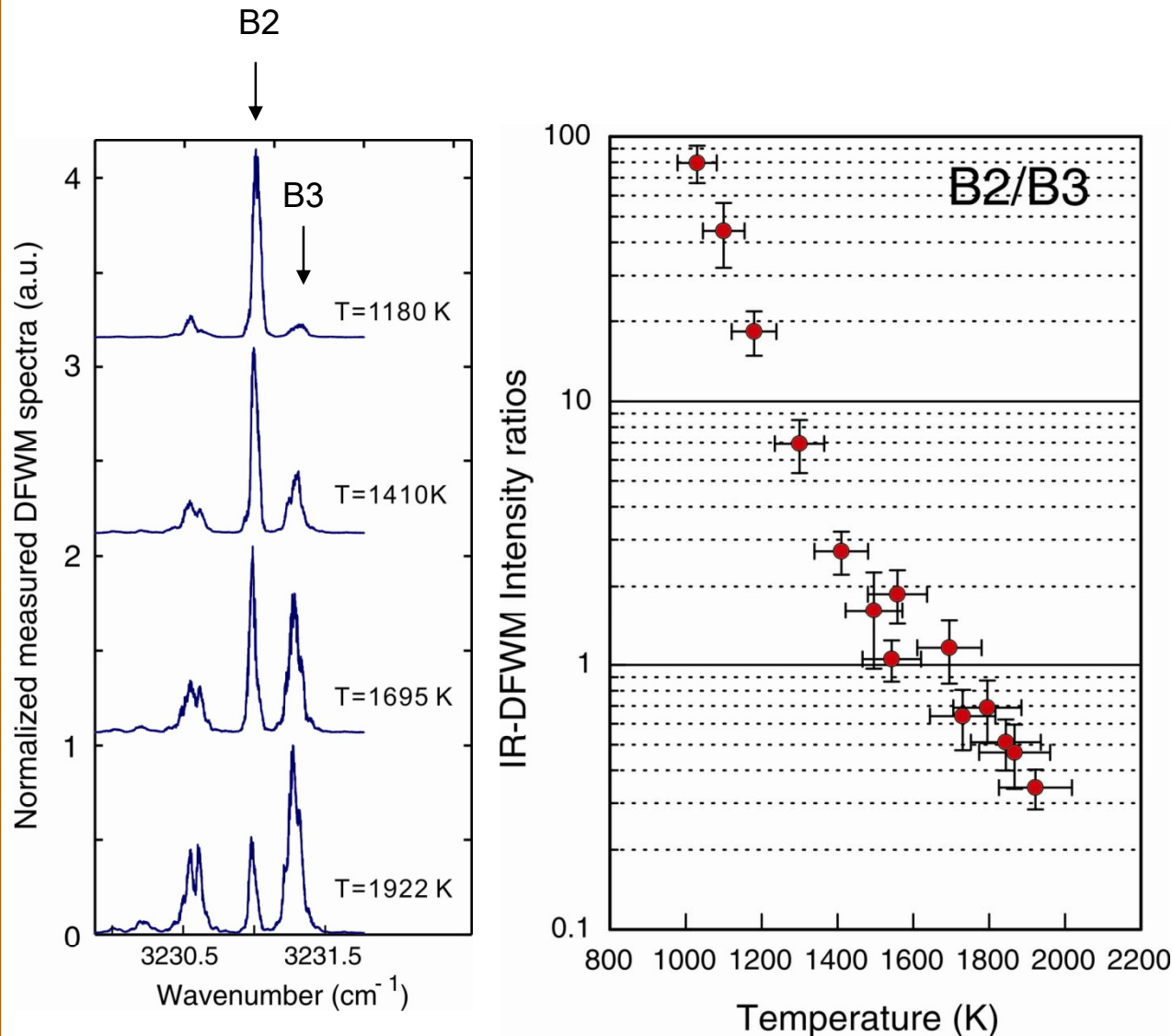


Investigation of detection limits

IR-DFWM spectrum of 510 ppm C<sub>2</sub>H<sub>2</sub> in a nitrogen gas flow. Partial assignments of the spectral lines have been made

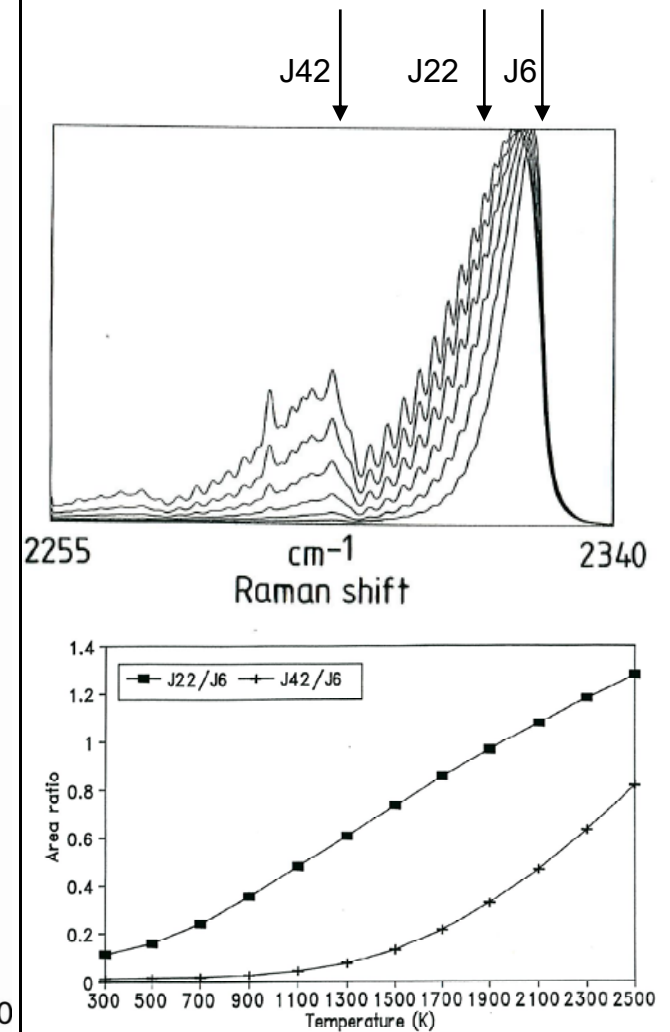


# Thermometry using IR H<sub>2</sub>O lines using IR-DFWM



Courtesy: Sun and Li 2010

## Comparison-CARS



Löfström, Kröll and Aldén, Proc. Comb. Symp. 24, 1637 (1992).

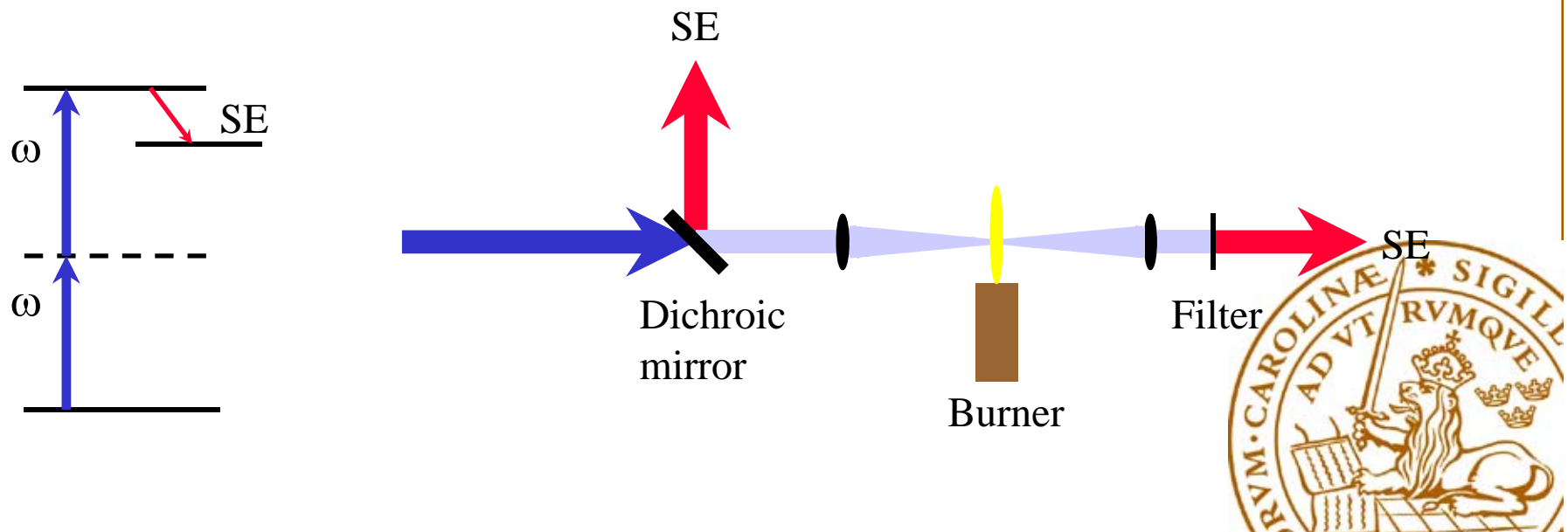




# Stimulated Emission (SE)

## Conceptual behaviour:

Two-photon UV excitation followed by laserlike emission in the visible/IR region



# Stimulated Emission

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## Advantages:

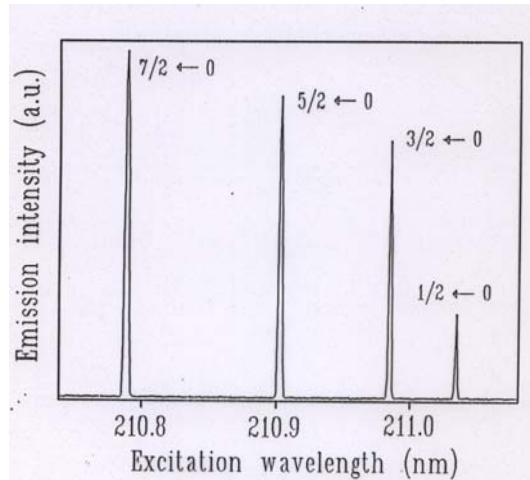
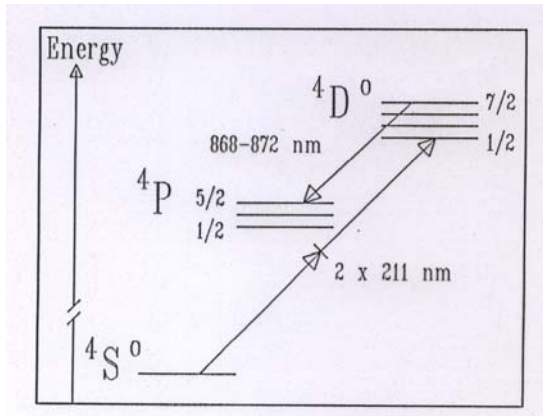
- **Signal generated as a new beam**
- **Bidirectional signal**
- **Very strong signal**
- **Simple set-up**
- **Minor species detection (N,C)**

## Disadvantages

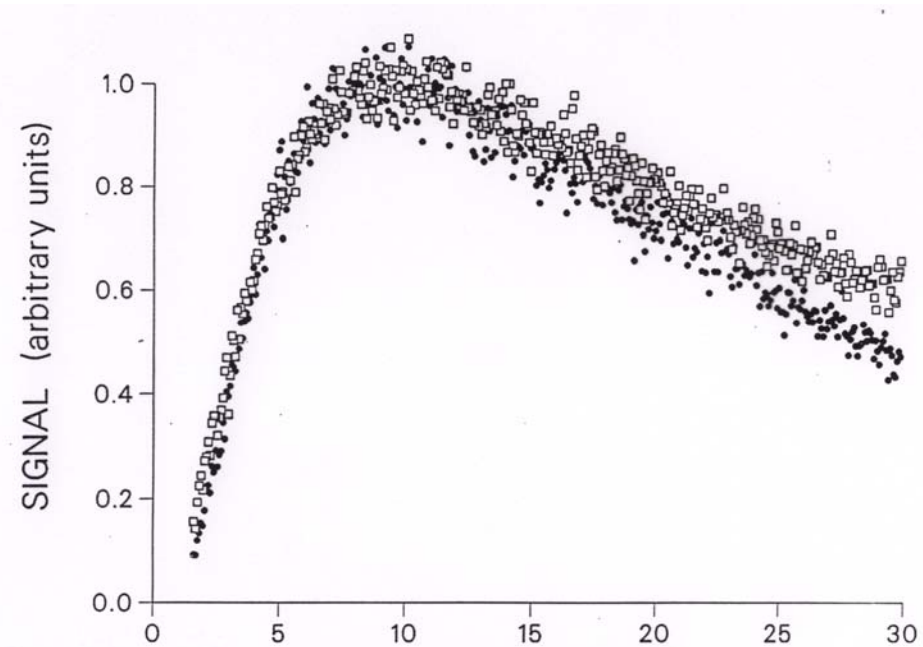
- **Difficult to model**
- **May interfere with LIF**
- **Low spatial resolution (?!)**



# Flame application of SE



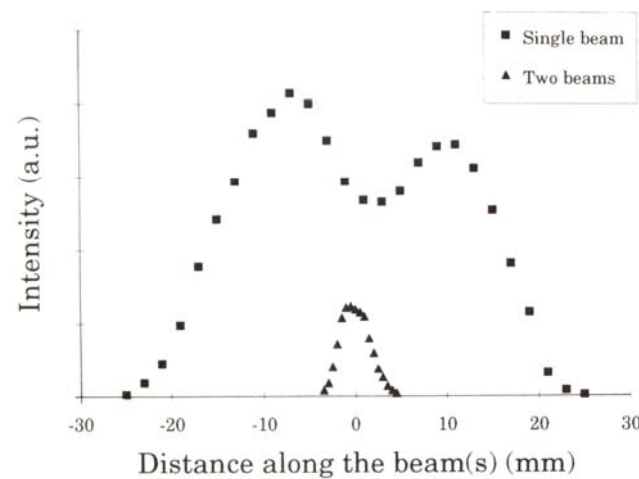
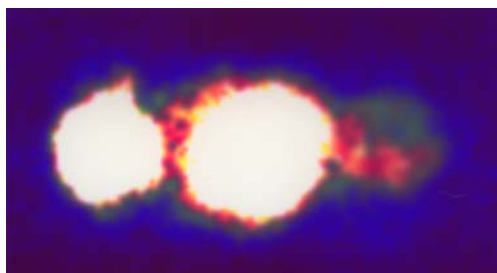
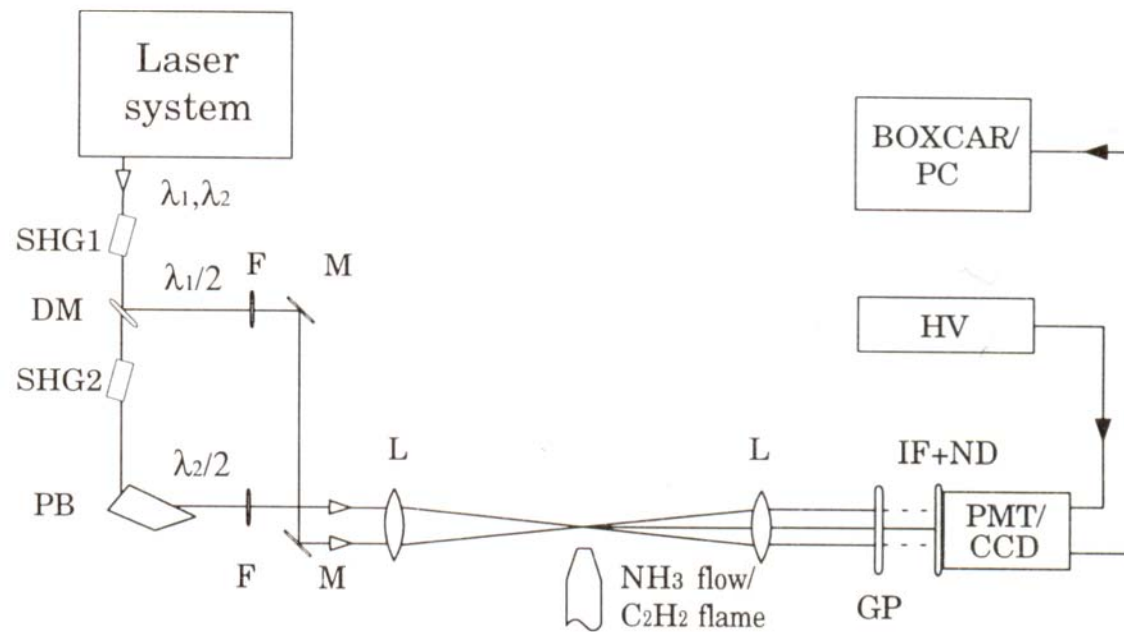
**N atom detection**



**O atom detection**



# Increased spatial resolution using SE



# Photochemical effects?

